

An easy entry to novel early–late oligonuclear transition metal complexes containing π -conjugated systems

Stephan Back^a, Wolfgang Frosch^a, Ignacio del Río^b, Gerard van Koten^b, Heinrich Lang^{a,*}

^a Technische Universität Chemnitz, Institut für Chemie, Lehrstuhl Anorganische Chemie, Straße der Nationen 62, 09111 Chemnitz, Germany

^b Debye Institute, Department of Metal-Mediated Synthesis, Utrecht University, Padualaan 8, 3584 CH Utrecht, The Netherlands

Received 24 September 1999

Abstract

Heterotrinnuclear Ti–Cu–Ru (**5**) and heterotetranuclear Ti–Cu–Pt–Fe (**7**) containing complexes are accessible by using $\{[\text{Ti}](\text{C}\equiv\text{C}^t\text{Bu})_2\}\text{CuMe}$ (**1**) ($[\text{Ti}] = (\eta^5\text{-C}_5\text{H}_4\text{SiMe}_3)_2\text{Ti}$) as key molecule; in **5** and **7**, the corresponding early and late transition metal atoms are linked by π -conjugated organic moieties. ©1999 Elsevier Science S.A. All rights reserved.

Keywords: Heterometallic complexes; Conjugated ligand complexes; π -Tweezer complexes; Early–late complexes; Transition metal complexes; Oligonuclear complexes

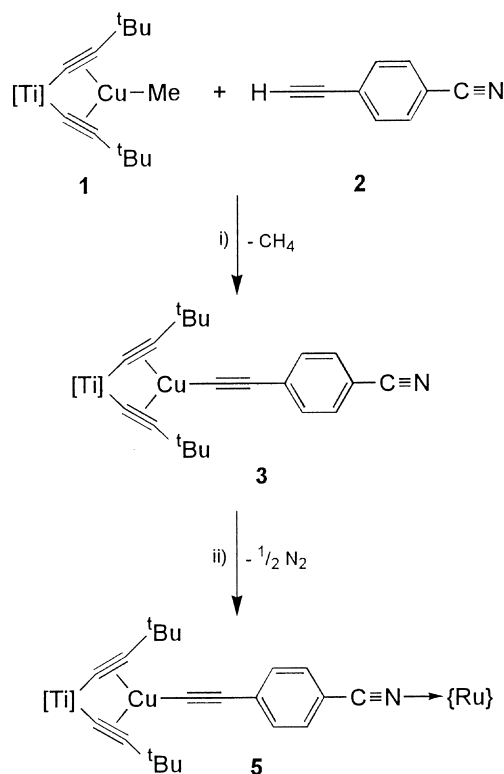
1. Introduction

Since the early work of Creutz and Taube (see Ref. [1]), there has been a growing interest in the synthesis, chemical and physical properties of transition metal (TM) complexes in which a π -conjugated organic ligand spans the corresponding TM atoms [2–6]. Besides homometallic complexes [7–10], a number of heterometallic species, which mainly feature late TM ions, are also known [11–16]. In contrast, little is known about early–late TM complexes [17–19]. We report in this communication, the preparation of TM complexes of the latter type containing up to four different TM centres.

2. Results and discussion

A key step in the synthesis of heteronuclear TM complexes in which the corresponding early and/or late TM centres are spanned by π -conjugated organic ligands, is the reaction of monomeric alkyne-stabilised CuMe, $\{[\text{Ti}](\text{C}\equiv\text{C}^t\text{Bu})_2\}\text{CuMe}$ (**1**) [20] ($[\text{Ti}] = (\eta^5\text{-C}_5\text{H}_4\text{SiMe}_3)_2\text{Ti}$), with CH-acidic organic molecules. Thus, complex **1** reacted with an equimolar amount of $\text{HC}\equiv\text{CC}_6\text{H}_4\text{CN}$ -4 (**2**) [21] at -70°C to yield the yellow Ti–Cu acetylide complex $\{[\text{Ti}](\text{C}\equiv\text{C}^t\text{Bu})_2\}\text{CuC}\equiv\text{CC}_6\text{H}_4\text{CN}$ -4 (**3**) (Scheme 1).

The heterodinuclear complex **3** contains a CN group which can further coordinate. When **3** is treated with

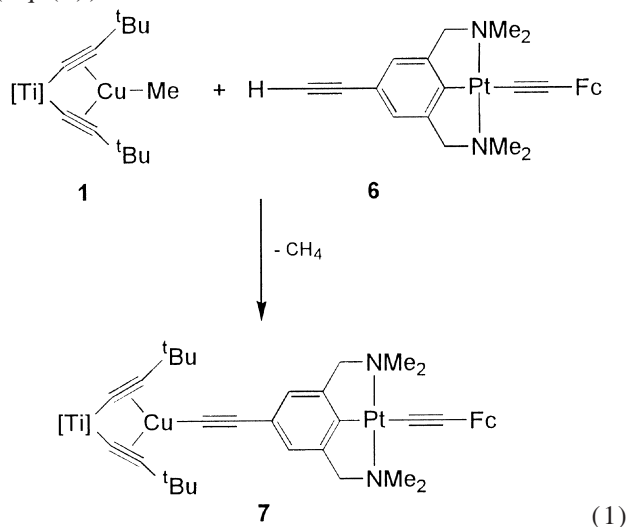


Scheme 1. Preparation of complexes **3** and **5**. (i) 0°C , n-pentane; (ii) $1/2 \text{ {Ru}}\text{N}\equiv\text{N}\{\text{Ru}\}$, 0°C , THF.

* Corresponding author. Tel.: +49 371 531 1200; fax: +49 371 531 1833; e-mail: heinrich.lang@chemie.tu-chemnitz.de

$\{\text{Ru}\}\text{N}\equiv\text{N}\{\text{Ru}\}$ (**4**) ($\{\text{Ru}\} = [\eta^3\text{-mer-(2,6-(Me}_2\text{NCH}_2)_2\text{-C}_5\text{H}_3\text{N})\text{RuCl}_2]$) [22] in molar ratio of 2:1 in THF at 0°C, N_2 is eliminated and heterotrinuclear **5** is formed (Scheme 1)¹.

An access to a heterotetranuclear complex is given by joining dinuclear **1** with the preformed late-late Pt–Fe organometallic acetylide $\text{HC}\equiv\text{C}\{\text{Pt}\}\text{C}\equiv\text{CFc}$ (**6**) ($\text{HC}\equiv\text{C}\{\text{Pt}\} = [\text{Pt}(\text{C}_6\text{H}_2\{\text{CH}_2\text{NMe}_2\}_2\text{-2,6-(C}\equiv\text{CH)-4}], \text{Fc} = (\eta^5\text{-C}_5\text{H}_4)\text{-Fe}(\eta^5\text{-C}_5\text{H}_5)$) [23]. In THF at 0°C, the Ti–Cu–Pt–Fe complex $\{[\text{Ti}](\text{C}\equiv\text{C}^t\text{Bu})_2\}\text{CuC}\equiv\text{C}\{\text{Pt}\}\text{C}\equiv\text{CFc}$ (**7**) is obtained in almost quantitative yield accompanied by the loss of CH_4 (Eq. (1))².



The heterometallic complexes **5** and **7** are less stable as compared to **1** [20], **3** or **6** [23].

The formation of **5** and **7** has been inferred through the IR spectroscopic data. The $\text{C}\equiv\text{N}$ stretching vibration for **5** has been shifted to lower wavenumbers (2208 cm^{-1}) as compared to that of **3** (2220 cm^{-1}), indicating the N-coordination

of the CN group to ruthenium. A similar observation has been made for $\{\text{Ru}\}\text{N}\equiv\text{CPh}$ [24] showing a slight decrease in bond order in $\text{C}\equiv\text{N}$. Besides the $\nu_{(\text{C}\equiv\text{N})}$ frequency, two typical $\text{C}\equiv\text{C}$ stretching vibrations have been found at 1922 cm^{-1} ($\text{TiC}\equiv\text{C}^t\text{Bu}$) and 2080 cm^{-1} ($\text{CuC}\equiv\text{CC}_6\text{H}_4$). These two absorptions are in characteristic regions observed for tweezer-stabilised Cu(I)-acetylides [25,26]. Complex **7** exhibits $\nu_{(\text{C}\equiv\text{C})}$ frequencies in this region for the corresponding organometallic acetylide units $\text{TiC}\equiv\text{C}^t\text{Bu}$ (1942 cm^{-1}), $\text{PtC}\equiv\text{CFc}$ (2071 cm^{-1}) and $\text{CuC}\equiv\text{CC}_6\text{H}_2$ (2080 cm^{-1}).

With the exception of the acetylide linkage, in the $^{13}\text{C}\{^1\text{H}\}$ NMR of **7**, the signals of the respective organic fragments are found at chemical shifts typical for the independent molecules **1** and **6**. However, adequate $^{13}\text{C}\{^1\text{H}\}$ NMR studies for **5** could not be carried out due to its instability in solution. This decomposition leads to a mixture of products of which $\{[\text{Ti}](\text{C}\equiv\text{C}^t\text{Bu})_2\}\text{CuCl}$ [25] could be separated and characterised unequivocally by spectroscopic means.

In contrast thereto, ^1H NMR studies of **5** reveal that the *ortho*- CH_2NMe_2 substituents of the $\text{C}_5\text{H}_3\text{N}(\text{CH}_2\text{NMe}_2)_2\text{-2,6}$ ligand remain datively bonded to the Ru(II) centre. The respective resonance signals are shifted to lower field by 0.5 ppm (CH_2) and 0.3 ppm (NMe_2) upon $\eta^1\text{-N}$ -coordination of the $\text{C}\equiv\text{N}$ moiety [24].

The result of this study shows that organic and organometallic π -systems, such as $\text{TiC}\equiv\text{C}^t\text{Bu}$, $\text{C}_6\text{H}_2(\text{CH}_2\text{NMe}_2)_2\text{-2,6-(C}\equiv\text{C)-4}$, $\text{C}\equiv\text{CC}_6\text{H}_4\text{CN-4}$ and $\text{C}\equiv\text{CC}_5\text{H}_4$, can act as linkage for different early and/or late TM complexes. This gives an entry into the synthesis of novel heterometallic complexes with the TM centres in different coordination modes.

Acknowledgements

This work was supported by the Volkswagen-Stiftung, Fonds der Chemischen Industrie and Konrad-Adenauer-Stiftung (W.F.).

References

- [1] C. Creutz, Prog. Inorg. Chem. 30 (1983) 1.
- [2] D.R. Kanis, P.G. Lacroix, M.A. Ratner, T.J. Marks, J. Am. Chem. Soc. 116 (1994) 10089.
- [3] M.B. Sponsler, Organometallics 14 (1995) 1920.
- [4] D. Astruc, Acc. Chem. Res. 30 (1997) 383.
- [5] P. Belanzoni, N. Re, A. Sgamellotti, C. Floriani, J. Chem. Soc., Dalton Trans. (1998) 1825.
- [6] N. Re, A. Sgamellotti, C. Floriani, J. Chem. Soc., Dalton Trans. (1998) 2521.
- [7] F. Coat, C. Lapinte, Organometallics 15 (1996) 477.
- [8] K. Heinze, G. Huttner, O. Walter, Eur. J. Inorg. Chem. (1999) 513.
- [9] W.Y. Wong, W.K. Wong, P.R. Raithby, J. Chem. Soc., Dalton Trans. (1998) 2761.
- [10] A. Buttinelli, E. Viola, E. Antonelli, C. Lo Sterzo, Organometallics 17 (1998) 2574.
- [11] S. Sakanishi, D.A. Bardwell, S. Couchman, J.C. Jeffery, J.A. McCleverty, M.D. Ward, J. Organomet. Chem. 528 (1997) 35.
- [12] J. Mata, S. Uriel, E. Peris, R. Llugar, S. Houbrechts, A. Persoons, J. Organomet. Chem. 562 (1998) 197.

¹ Synthesis of **5**. 33 mg (0.05 mmol) of **3** [26] and 17 mg (0.02 mmol) of **4** [21] were dissolved in 10 ml THF (0°C). After stirring for 4 h at 0°C the mixture was evaporated to dryness and washed with n-pentane (5×5 ml) and then dried in vacuo to yield **5** (40 mg, 0.04 mmol, 80%); m.p. 112°C (decomp.). IR (KBr) (cm^{-1}): ν 2208 (CN), 2080, 1922 ($\text{C}\equiv\text{C}$). ^1H NMR (250.13 MHz, CDCl_3): δ 0.22 (s, 18 H, SiMe_3), 1.38 (s, 18 H, ^tBu), 2.52 (s, 12 H, NMe_2), 4.01 (s, 4 H, CH_2), 5.95 (t, $^3J_{\text{HH}} = 1.3$ Hz, C_5H_4), 6.01 (t, $^3J_{\text{HH}} = 1.3$ Hz, C_5H_4), 7.1–7.8 (m, 3 H, $\text{C}_5\text{H}_3\text{N}$).

² Synthesis of **7**. To **1** (100 mg, 0.18 mmol) [20] in THF (50 ml, 25°C), **6** [23] (110 mg, 0.18 mmol) was added in one portion. After stirring for 1 h, the mixture was filtered through a pad of Celite. Then, all volatiles were removed in vacuo to yield **7** (200 mg, 0.17 mmol, 95%); m.p. 176°C (decomp.). IR (KBr) (cm^{-1}): 2080, 2071, 1942 ($\text{C}\equiv\text{C}$). ^1H NMR (250.13 MHz, CDCl_3): δ 0.27 (s, 18 H, SiMe_3), 1.38 (s, 18 H, ^tBu), 3.20 (s, $^3J_{\text{PH}} = 46.2$ Hz, 12 H, NMe_2), 4.05 (t, $^3J_{\text{HH}} = 1.2$ Hz, 2 H, C_5H_4), 4.08 (s, 4 H, CH_2), 4.16 (s, 5 H, C_5H_5), 4.37 (t, $^3J_{\text{HH}} = 1.2$ Hz, 2 H, C_5H_4), 5.49 (t, $^3J_{\text{HH}} = 1.5$ Hz, 2 H), 5.69 (t, $^3J_{\text{HH}} = 1.5$ Hz, 2 H), 7.03 (s, 2 H, C_6H_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (62.895 MHz, CDCl_3): δ 0.3 (SiMe_3), 31.1 (^tBu), 32.0 (C/ ^tBu), 55.3 (NMe_2), 67.1 (C_5H_4), 68.9 (C_5H_5), 70.5 (C_5H_4), 75.2 (*ipso*-C, C_5H_4), 79.6 (CH_2), 110.2 (C_5H_4), 113.5 (C_5H_4), 113.9 ($\text{CuC}\equiv\text{C}$), 115.2 (*ipso*-C, C_5H_4), 116.1 ($\text{CuC}\equiv\text{C}$), 123.4 ($\text{C}\equiv\text{CSi}$), 138.2 (C_6H_2), 144.2 (*ipso*-C, C_6H_2), 146.1 (TiC), 168.1 (*ipso*-C, C_6H_2); due to the instability of **7** in solution the carbon signals of $\text{PtC}\equiv\text{C}$ could not be detected.

- [13] S.B. Falloon, S. Szafert, A.M. Arif, J.A. Gladysz, *Chem. Eur. J.* 4 (1998) 1033.
- [14] U. Siemeling, U. Vorfeld, B. Neumann, H.G. Stammer, P. Zanello, F. Fabrizi de Biani, *Eur. J. Inorg. Chem.* (1999) 1.
- [15] S. Le Stang, D. Lenz, F. Paul, C. Lapinte, *J. Organomet. Chem.* 572 (1999) 189.
- [16] T.Y. Liu, Y.J. Chen, C.C. Tai, K.S. Kwan, *Inorg. Chem.* 38 (1999) 674.
- [17] F.R. Lembke, D.J. Szallida, R.M. Bullock, *J. Am. Chem. Soc.* 113 (1991) 8466.
- [18] Y. Hayashi, M. Osawa, K. Kobayashi, Y. Wakatsuki, *Chem. Commun.* (1996) 1617.
- [19] S. Back, R.A. Gossage, I. del Río, G. Rheinwald, H. Lang, G. van Koten, *J. Organomet. Chem.* 582 (1999) 126 and Refs. therein.
- [20] H. Lang, K. Köhler, L. Zsolnai, M. Büchner, A. Driess, G. Huttner, J. Strähle, *Organometallics* 18 (1999) 598.
- [21] S. Takahashi, Y. Kuroyama, K. Sonogashira, N. Hagihara, *Synthesis* (1980) 627.
- [22] R.A.T.M. Abbenhuis, I. del Río, M.M. Berghoef, J. Boersma, N. Veldman, A.L. Spek, G. van Koten, *Inorg. Chem.* 37 (1998) 1749.
- [23] S. Back, R.A. Gossage, H. Lang, G. van Koten, *Eur. J. Inorg. Chem.* (1999) submitted for publication.
- [24] I. del Río, S. Back, M.S. Hannu, G. Rheinwald, H. Lang, G. van Koten, *Inorg. Chim. Acta* (1999) submitted for publication.
- [25] H. Lang, G. Rheinwald, *J. Prakt. Chem.* 341 (1999) 1 and Refs. therein.
- [26] W. Frosch, S. Back, G. Rheinwald, H. Lang, *J. Organomet. Chem.* (1999) submitted for publication.