

Note

Synthesis of a sialic acid ketoside by an intramolecular oxymercuration-demercuration reaction

JEAN-MARIE BEAU*, ROLAND SCHAUER,

Biochemisches Institut im Fachbereich Medizin der Christian-Albrechts Universität, 23 Kiel (Germany)

JOHAN HAVERKAMP**, LAMBERTUS DORLAND, JOHANNES F. G. Vliegenthart,

Laboratory of Organic Chemistry, University of Utrecht, Croesestraat 79, Utrecht (The Netherlands)

AND PIERRE SINAY†

Laboratoire de Biochimie Structurale, U.E.R. de Sciences Fondamentales et Appliquées, E.R.A. 739, 45046 Orléans Cédex (France)

(Received September 10th, 1979; accepted for publication, October 19th, 1979)

Oxymercuration-borohydride demercuration of glycal acetates in alcohol solution gives alkyl 2-deoxyglycoside acetates¹⁻⁵. The method has been applied to the synthesis of 2'-deoxy disaccharides⁶ and in a synthesis of 4-*O*-methyl-*N*-acetylneuraminic acid⁷. All of these reactions involve intermolecular processes.

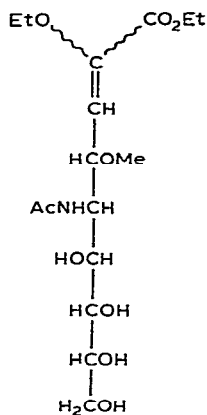
It was considered that mercuration of ethyl 5-acetamido-3,5-dideoxy-2-*O*-ethyl-4-*O*-methyl-*D*-glycero-*D*-galacto-non-2-enonate (**1**) in a non-hydroxylic solvent might give, after demercuration, an ethyl glycoside (*e.g.*, **2**) *via* an intramolecular process. Related cyclisation reactions have been reported⁸.

Hydrolysis of ethyl 5-acetamido-3,5-dideoxy-2-*O*-ethyl-6,7:8,9-di-*O*-isopropylidene-4-*O*-methyl-*D*-glycero-*D*-galacto-non-2-enonate⁷ to give **1** is effected better with toluene-*p*-sulphonic acid in aqueous ethanol than with aqueous trifluoroacetic acid. Treatment of **1** with 2 equiv. of mercury(II) trifluoroacetate in anhydrous tetrahydrofuran followed by demercuration *in situ* with alkaline sodium borohydride gave the ketoside **2** (88%). Compound **2** reacted with acetone or 2-butanone, used as crystallisation solvents, to give variable amounts of ketals **3** or **4**, so that only moderate yields of crystalline **2** were obtained; crystallisation did not occur from non-ketonic solvents. In aqueous solution at room temperature, **3** or **4** was autohydrolysed into **2** within 10 h, and acid hydrolysis of **2** gave the known⁷ 4-*O*-methyl-*N*-acetylneuraminic acid.

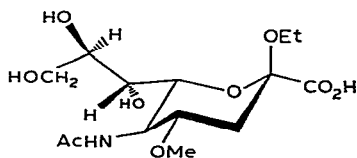
*Present address: Laboratoire de Biochimie Structurale, U.E.R. de Sciences Fondamentales et Appliquées, 45046 Orléans Cédex, France.

**Present address: FOM Institute for Atomic Molecular Physics, Kruislaan 407, Amsterdam, The Netherlands.

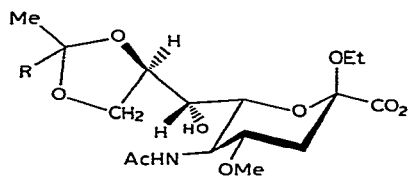
†To whom requests for reprints should be sent.



1



2



3 R = Me

4 R = Et

$^1\text{H-N.m.r.}$ data (360 MHz) on **2**, **3**, and the reference compounds NeuAc, 4-*O*-methyl-NeuAc, and methyl α - and β -D-NeuAc, are given in Table I. The coupling constants for **2** are similar to those of NeuAc and 4-*O*-Me-NeuAc, and indicate the $1C(D)$ conformation⁷. The chemical shifts of H-3eq and H-4 are strongly indicative of the anomeric configuration⁹. Taking into account the shift increments due to alkyl glycoside formation and methylation of HO-4, the chemical shifts for H-3eq (2.158 p.p.m.) and H-4 (3.820 p.p.m.) of **2** indicate the configuration. The n.m.r. parameters of **2** and **3** are similar, except for the chemical shifts of H-6,7,8,9,9' and the values of $J_{7,8}$, $J_{8,9}$, and $J_{9,9'}$, which reflect the effect of the 8,9-acetal in **3**.

The stereospecificity in the formation of the above glycoside is noteworthy and the scope of this reaction is being explored.

EXPERIMENTAL

General methods. — Melting points were determined in capillary tubes with a Büchi apparatus and are uncorrected. Optical rotations were measured with a Perkin-Elmer Model 141 polarimeter and i.r. spectra with a Perkin-Elmer Model 457 spectrophotometer. $^1\text{H-N.m.r.}$ (360-MHz) spectra were obtained for solutions in neutral D_2O at 25° with a Bruker HX-360 spectrometer operating in the Fourier-transform mode. Chemical shifts are given relative to that of sodium 2,2-dimethyl-2-silapentane-5-sulphonate. Methyl α - and β -NeuAc were prepared by saponification of the corresponding methyl esters¹⁰ at 40° in D_2O at pD ~ 11 in the presence of triethylamine. The reaction was monitored by n.m.r. spectroscopy; after disappearance of the ester methyl signal, the solution was lyophilised. Purity of products was determined by t.l.c., either on Silica Gel 60 F 254 (Merck) with detection by charring with sulphuric acid-methanol (1:1) or on cellulose (Merck) with detection of sialic acid by staining with the Bial reagent¹¹. Solvent systems: *A*, chloroform-methanol (85:15); *B*, ethyl acetate-acetic acid-water (5:2:2); *C*, 1-butanol-acetic acid-water (5:2:3); *D*, 1-

TABLE I

¹H-N.M.R. DATA (360 MHz) FOR 2 AND 3 AND REFERENCE SUBSTANCES^a

Compound	H-3 _{eq}	H-3 _{ax}	H-4	H-5	H-6	H-7	H-8	H-9	H-9'	NAc	Aglycon		
											CH ₂	Me	4-OMe Me-C
2	2.518	1.514	~3.820	3.931	~3.834	3.525	3.859	~3.834	3.653	2.041	3.332; 3.585 ^c	1.175	3.383
3 NeuAc	2.512	1.516	3.809	3.914	3.705	3.587	4.352	4.165	3.990	2.037	3.340; 3.543 ^c	1.191	3.380
	2.208	1.827	4.024	3.899	3.984	3.514	3.753	3.835	3.608	2.050	—	—	—
Methyl β-D-NeuAc	(2.730) ^b	(1.621) ^b								(2.030) ^b	—	—	—
Methyl α-D-NeuAc	2.337	1.645	4.009	~3.88	3.785	3.532	~3.88	~3.85	3.662	2.047	—	3.200	—
4-O-Me-NeuAc	2.718	1.626	3.675	3.803	3.689	3.586	3.886	3.869	3.641	2.033	—	3.341	—
	2.382	1.711	~3.77	~3.96	~3.99	3.505	3.756	3.835	3.608	2.039	—	—	3.386
	(2.901) ^b	(1.507) ^b								(2.023) ^b	—	—	—

	J _{3ax,3eq}	J _{3ax,4}	J _{3eq,4}	J _{4,5}	J _{5,6}	J _{6,7}	J _{7,8}	J _{8,9}	J _{8,9'}	J _{9,9'}	Aglycon	
											J _{Me}	J _{gem}
2	-12.9	11.6	4.6	10.4	10.4	~1.0	9.6	2.7	6.0	-12.3	7.0	-9.0
3 NeuAc	-13.0	11.2	4.6	10.4	10.5	≤1.0	8.1	6.0	5.9	-8.7	7.1	-9.1
	-13.2	11.8	5.0	10.4	10.7	1.2	9.4	2.8	6.4	-12.4	—	—
4-O-Me-NeuAc	-13.1	11.7	4.7	10.6	10.6	~1.0	9.1	2.6	6.5	-12.0	—	—

^aNeutral solutions in D₂O at 25°, p.p.m. relative to the signal for sodium 2,2-dimethyl-2-silapentane-5-sulphonate. ^bSignals for α anomer, which represents 7% of the anomeric mixture. ^cThe methylene protons of the ethoxyl group are not equivalent.

propanol-1-butanol-0.1M HCl (2:1:1). Column chromatography was performed on Silica Gel 60 (Merck 0.063–0.200 mm).

Ethyl 5-acetamido-3,5-dideoxy-4-O-methyl-D-glycero-β-D-galacto-2-nonulopyranosidonic acid (2). — A solution of ethyl 5-acetamido-3,5-dideoxy-2-O-ethyl-6,7:8,9-di-O-isopropylidene-4-O-methyl-D-glycero-D-galacto-non-2-enonate⁷ (505 mg) in ethanol-water (7:1, 25 ml) containing toluene-*p*-sulphonic acid (620 mg) was heated for 2 h at 80°, and then cooled to 0°, neutralised (PbCO₃), filtered, and concentrated. The residue was extracted with ethanol-ethyl acetate (1:1, 30 ml), and the extracted material was eluted from a column of silica gel (15 g) with chloroform-methanol (9:1), to give ethyl 5-acetamido-3,5-dideoxy-2-O-ethyl-4-O-methyl-D-glycero-D-galacto-non-2-enonate⁷ (**1**; 292 mg, 70%), m.p. 136°, $[\alpha]_D^{20} -9^\circ$ (*c* 0.6, methanol-water, 4:1, v/v).

To a solution of **1** (207 mg) in anhydrous tetrahydrofuran (20 ml) at 0° was added mercury(II) trifluoroacetate (465 mg) in three portions at intervals of 15 min. After 6 h, the pH of the mixture was adjusted to 10 with M sodium hydroxide, and M sodium borohydride in M sodium hydroxide (0.6 ml) was added. The mixture was stirred at 0° for 1 h, neutralised with M acetic acid, filtered, and concentrated. A solution of the residue in tetrahydrofuran-water (3:2, 10 ml) at 4° was treated with Dowex-50 (H⁺) resin (10 ml) and eluted from a column of Dowex-2 X8 (AcO[−]) resin (100–200 mesh; 30 ml, equilibrated in water at 4°) with 0.05M pyridinium acetate (pH 6). Bial-positive fractions¹¹ were combined and concentrated to give the ethyl glycoside (**2**) of 4-O-methyl-β-D-N-acetylneuraminic acid as a colourless syrup (168 mg, 88%). Crystallisation from acetone-1-propanol gave material (45 mg, 23%) having m.p. 156–157°, $[\alpha]_D^{20} -31.5^\circ$ (*c* 0.7, methanol); $\nu_{\max}^{\text{Nujol}}$ 3420–3250 (NH, OH), 1680 (COOH), 1635 (Amide I), and 1570 cm^{−1} (Amide II). N.m.r. data are given in Table I.

Anal. Calc. for C₁₄H₂₅NO₉ · H₂O: C, 45.24; H, 6.99; N, 3.54. Found: C, 45.52; H, 7.37; N, 3.79.

Ethyl glycoside (3) of 8,9-O-isopropylidene-4-O-methyl-β-D-N-acetylneuraminic acid. — (a) The mother liquors from the crystallisation of **2** were concentrated and the residue was crystallised from acetone, to give the ketal **3** (94 mg, 44% from **4**), m.p. 191–192°, $[\alpha]_D^{20} -23^\circ$ (*c* 0.2, methanol); $\nu_{\max}^{\text{Nujol}}$ 3265 (NH, OH), 1750 (COOH), 1640 (Amide I), and 1560 cm^{−1} (Amide II). N.m.r. data are given in Table I.

Anal. Calc. for C₁₇H₂₉NO₉: C, 52.17; H, 7.47; N, 3.58. Found: C, 52.32; H, 7.60; N, 3.38.

(b) To a solution of **2** (3 mg) in acetone (3 ml) at −20° were added a few beads of dried Dowex-50 (H⁺) resin. T.l.c. (solvents *B* and *C*) showed reaction to be complete within 15 min. The resin was collected, and washed with acetone, and the combined washings and filtrate were concentrated to give **3**, which was identical with the compound described in (a).

Ethyl glycoside (4) of 8,9-O-sec-butylidene-4-O-methyl-β-D-N-acetylneuraminic acid. — Crystallisation of **2** from 2-butanone-2-propanol-ether gave **4** as small needles (30% from **2**), m.p. 187.5–188.5°, $[\alpha]_D^{20} -26^\circ$ (*c* 0.35, methanol); $\nu_{\max}^{\text{Nujol}}$

3270 (NH, OH), 1750 (COOH), 1640 (Amide I), and 1560 cm^{-1} (Amide II).

Anal. Calc. for $\text{C}_{18}\text{H}_{31}\text{NO}_9$: C, 52.32; H, 7.71; N, 3.45. Found: C, 52.97; H, 7.36; N, 3.15.

Autohydrolysis of 3 and 4. — When a solution of 3 or 4 (0.5 mg) in water was kept for 10 h at room temperature, t.l.c. (solvents *B* and *C*) showed complete conversion into 2.

4-O-Methyl-N-acetylneuraminic acid. — A solution of 2 (0.5 mg) in 0.1M hydrochloric acid was heated for 4 h at 80°. Conversion into known 4-*O*-methyl-*N*-acetylneuraminic acid⁷ was monitored by t.l.c. on cellulose (solvent *D*) and silica gel (solvent *B*).

ACKNOWLEDGMENTS

One of us (J.M.B.) thanks the Humboldt Foundation for the award of a Research Fellowship. This investigation was supported by the Netherlands Foundation for Chemical Research (SON) with financial aid from the Netherlands Organization for the Advancement of Pure Research (ZWO).

REFERENCES

- 1 G. R. INGLIS, J. C. P. SCHWARZ, AND L. McLAREN, *J. Chem. Soc.*, (1962) 1014–1019.
- 2 P. T. MANOLOPOULOS, M. MEDNICK, AND N. N. LICHTIN, *J. Am. Chem. Soc.*, 84 (1962) 2203–2210.
- 3 K. TAKIURA AND S. HONDA, *Carbohydr. Res.*, 21 (1972) 379–391.
- 4 S. HONDA, K. IZUMI, AND T. TAKIURA, *Carbohydr. Res.*, 23 (1972) 427–429.
- 5 J. H. LEFTIN AND N. N. LICHTIN, *Isr. J. Chem.*, 3 (1965) 107–111.
- 6 S. HONDA, K. KAKEHI, H. TAKAI, AND K. TAKIURA, *Carbohydr. Res.*, 29 (1973) 477–487.
- 7 J.-M. BEAU, P. SINAÏ, J. P. KAMERLING, AND J. F. G. Vliegenthart, *Carbohydr. Res.*, 67 (1978) 65–77.
- 8 J. SAND AND F. SINGER, *Justus Liebigs Ann. Chem.*, 329 (1903) 166–189; R. ADAMS, F. L. ROMAN, AND W. N. SPERRY, *J. Am. Chem. Soc.*, 44 (1922) 1781–1792; L. E. MILLS AND R. ADAMS, *ibid.*, 45 (1923) 1842–1854; V. SPEZIALE, J. ROUSSEL, AND A. LATTES, *J. Heterocycl. Chem.*, 11 (1974) 771–775, and references therein; V. SPEZIALE, M. AMAT, AND A. LATTES, *ibid.*, 13 (1976) 349–355; V. SPEZIALE, H. G. DAO, AND A. LATTES, *ibid.*, 15 (1978) 225–231.
- 9 J. HAVERKAMP, L. DORLAND, J. F. G. Vliegenthart, J. MONTREUIL, AND R. SCHAUER, *Abstr. Int. Symp. Carbohydr. Chem.*, 9th, London, (1978) D7.
- 10 R. K. YU AND R. LEDEEN, *J. Biol. Chem.*, 244 (1969) 1306–1313.
- 11 P. BÖHM, S. DAUBER, AND L. BAUMEISTER, *Klin. Wochenschr.*, 32 (1954) 289–292.