

Primary structure of the glycans of porcine pancreatic lipase

Bernard FOURNET¹, Yves LEROY¹, Jean MONTREUIL¹, Josiane DÉCARO², Mireille ROVERY², J. Albert van KUIK³
 and Johannes F. G. VLIEGENTHART³

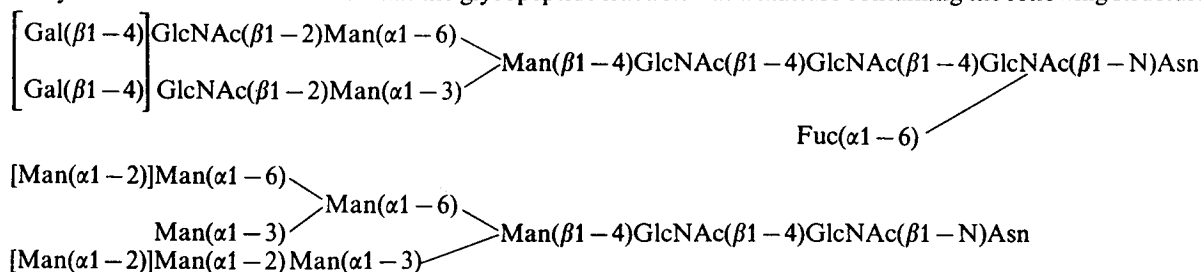
¹ Laboratoire de Chimie Biologique de l'Université des Sciences et Techniques de Lille Flandres-Artois
 et Unité Associée au Centre National de la Recherche Scientifique No 217, Villeneuve d'Ascq

² Centre de Biochimie et de Biologie Moléculaire du Centre National de la Recherche Scientifique, Marseille

³ Department of Bio-Organic Chemistry, University of Utrecht

(Received April 7/July 23, 1987) – EJB 87 0415

The glycan primary structure of the main glycopeptide fraction obtained by pronase and carboxypeptidase A digestions of porcine pancreatic lipase has been investigated by 500-MHz ¹H-NMR spectroscopy and methylation analysis. The results demonstrate that the glycopeptide fraction was a mixture containing the following structures:



Porcine pancreatic lipase (triacylglycerol acylhydrolase) is one of the principal enzymes involved in the intraluminal digestion of fats [1]. It has been shown that porcine [2, 3] and human [4] lipases are glycoproteins whereas ovine, bovine [5] and horse [6] lipases are not glycosylated. In one of the first preparations of porcine pancreatic lipase [7], two isolipases L_A and L_B were characterized (approximate ratio 1:4). Their carbohydrate moieties were found to be made up each of a single glycan chain containing: fucose 0.83, 0.92; galactose 1.05, 0.53; glucose 0.48, 0.16; mannose 4.12, 4.81; glucosamine 3.05, 2.98 and sialic acid 0.31, 0.04 mol/mol lipases L_A and L_B, respectively [8]. The non-stoichiometric value of the sialic acid content in lipase A suggested that the preparation L_A contained more than one kind of isolipase. A few years later it was shown that lipase L_A contained several isoenzymes [9]. Since no allotypic replacement of amino acid has been reported in the course of the lipase primary structure determination [10], the variability in the isolipases could only originate from a microheterogeneity of the glycan chains.

The amino acid sequence of porcine pancreatic lipase has been reported [10]. The glycan chain (CHO) is attached to asparagine at position 166, the sequence previously found around this residue: Thr-Asn-Gly-Thr-Ile-Glu-Arg- was confirmed.



MATERIALS AND METHODS

Porcine pancreatic lipase was prepared from delipidated powder of porcine pancrei by the method of Rovery et al. [9].

Correspondence to B. Fournet, Laboratoire de Chimie Biologique de l'Université des Sciences et Techniques de Lille Flandres-Artois et Unité Associée au CNRS No. 217, F-59655 Villeneuve d'Ascq, France

Enzyme. Porcine pancreatic lipase, triacylglycerol acylhydrolase (EC 3.1.1.3)

The isolipase mixture was digested with *Staphylococcus aureus* protease [11] followed by pronase (twice) and carboxypeptidase treatments [12]. The glycopeptides were purified by gel-filtration on Sephadex G-25 (Superfine) with 60 mM ammonium bicarbonate buffer pH 8.5.

Qualitative and quantitative sugar analyses were carried out after methanolysis (MeOH/0.5 M HCl, 24 h, 80°C) by gas-liquid chromatography of trifluoroacetylated methyl glycosides [13].

Permethylation was performed according to Hakomori [14]. Partially methylated monosaccharides were identified as described by Fournet et al. [15].

For 500-MHz ¹H-NMR spectroscopy, the glycopeptide fraction was repeatedly exchanged in ²H₂O (99.96 atom % ²H, Aldrich) with intermediate lyophilization. ¹H-NMR spectra were recorded on a Bruker WM-500 spectrometer (SON hF-NMR facility; Department of Biophysical Chemistry; University of Nijmegen; The Netherlands) operating at 500 MHz in the Fourier-transform mode at a probe temperature of 27°C. Resolution enhancement of the spectra was achieved by Lorentzian-to-Gaussian transformation [16]. Chemical shifts (δ) are given relative to sodium 4,4-dimethyl-4-silapentane-1-sulfonate, but were actually measured indirectly to acetone (δ = 2.225 ppm) [17].

RESULTS AND DISCUSSION

The digestion of isolipase preparation with *Staphylococcus aureus* protease led to the isolation of two glycopeptide fractions (fractions Sta and Stb) with the same amino acid sequence [11]: Ala¹⁶¹-Gly-Arg-Arg-Thr-Asn¹⁶⁶-Gly-Thr-Ile-Glu¹⁷⁰. Each of the fractions Sta and Stb was resolved into two subfractions, Sta A, Sta B and Stb A, Stb B, by electrophoresis on thin-layer cellulose plates [18]. The major subfractions Sta A (200 nmol) and Stb B (125 nmol) migrated

Table 1. Molar ratios of the monosaccharide methyl ethers obtained by methanolysis of the permethylated glycopeptide Sta AP followed by acetylation of free methyl ethers

Values are calculated on the basis of 2 mol methyl-3,4,6-tri-*O*-methyl-2-mono-*O*-acetylmannoside

Methyl ether	Molar ratio
Methyl 2,3,4-tri- <i>O</i> -methylfucoside	0.9
Methyl 2,3,4,6-tetra- <i>O</i> -methylmannoside	0.4
Methyl 2,3,4,6-tetra- <i>O</i> -methylgalactoside	0.4
Methyl 3,4,6-tri- <i>O</i> -methyl-2-mono- <i>O</i> -acetylmannoside	2
Methyl 2,4-di- <i>O</i> -methyl-3,6-di- <i>O</i> -acetylmannoside	1.3
Methyl 3,4,6-tri- <i>O</i> -methyl- <i>N</i> -methyl-acetylglucosaminide	1.5
Methyl 3,6-di- <i>O</i> -methyl-4-mono- <i>O</i> -acetyl- <i>N</i> -methyl-acetylglucosaminide	1.4
Methyl 3-mono- <i>O</i> -methyl-4,6-di- <i>O</i> -acetyl- <i>N</i> -methyl-acetylglucosaminide	0.6

8.5 cm and 7.5 cm, respectively, towards the cathode and the minor subfractions Sta B and Stb A migrated 7.5 and 8.5 cm, respectively, towards the cathode. In order to shorten the peptide chain, the major component Sta A (280 nmol) was digested with pronase (twice) and carboxypeptidase A. The glycopeptide Sta AP Thr-Asn(CHO)-Gly was isolated.

Determination of the sugar composition of glycopeptide Sta AP by gas-liquid chromatography yielded fucose, galactose, mannose and *N*-acetylglucosamine in a molar ratio 0.7:0.3:3.3:2.9, respectively.

The molar ratios of the partially methylated and acetylated methylglycosides obtained by methanolysis of the permethylated glycopeptide Sta AP followed by acetylation were determined by GLC-MS [15] and are given in Table 1. The identification of 3,4,6-tri-*O*-methyl and 2,4-di-*O*-methyl mannoses is in favour of the presence of a mannotrioso branching core with a biantennary structure. The presence of 0.6 mol 3-mono-*O*-methyl glucosamine indicates that one residue of *N*-acetylglucosamine was substituted by one fucose residue (0.9 mol 2,3,4-tri-*O*-methylfucose) in $\alpha(1-6)$ linkage.

The ¹H-NMR spectrum of Sta AP reveals a mixture of asparagine-linked carbohydrate chains. Relevant NMR parameters are listed in Table 2. The major component (70%, estimated from the relative intensity of the anomeric proton signals) of this mixture is an incomplete biantennary structure with GlcNAc-5 and -5' in terminal position and with Fuc in $\alpha(1-6)$ linkage to GlcNAc-1. For the numbering of the monosaccharide residues, see Fig. 1. The biantennary core structure is evident from the characteristic set of chemical shifts of Man H-2 atoms (Man-3, $\delta = 4.249$ ppm; Man-4, $\delta = 4.189$ ppm; Man-4', $\delta = 4.118$ ppm). The Fuc residue gave rise to a well known set of structural-reporter groups (H-1, $\delta = 4.872$ ppm; H-5, $\delta = 4.12$ ppm; CH₃, $\delta = 1.21$ ppm) which reveals Fuc in $\alpha(1-6)$ linkage to GlcNAc-1. This is corroborated by GlcNAc-2 H-1, which is resonating at 4.684 ppm, and NAc, which is found at 2.089 ppm [17, 19]. The absence of an *N*-acetyl signal at ≈ 2.079 ppm reflects the absence of an afuco analogue of the biantennary structure. The diversity of GlcNAc-1 NAc signals is due to heterogeneity in the peptide moiety. The GlcNAc-5 and -5' anomeric protons are resonating mainly at $\delta = 4.555$ ppm, which is typical for terminal GlcNAc residues [17]. However, a small part (10%) of the GlcNAc-5/5' H-1 signals is shifted downfield to $\delta = 4.580$ ppm. In combination with a signal of equal intensity

Table 2. Relevant ¹H-chemical shifts of structural-reporter groups constituent monosaccharides for glycopeptide Sta AP from porcine pancreatic lipase

Chemical shifts are given in ppm downfield from sodium dimethyl-4-silapentane-1-sulfonate in ²H₂O at 27°C acquired at 500 MHz (but were actually measured relative to internal acetone, $\delta = 2.225$ ppm). For the numbering of the monosaccharide residues and complete structures, see Fig. 1
n.d. = not determined

Reporter group	Residue	Chemical shift in ppm				
		A galacto component	galacto component	Man 6 component	extended Man 6 component	
H-1		1	5.032		5.032	
		2	4.684		n.d.	
		4	5.116		5.345	
		4'	4.914		4.87*	
		5	4.555	4.580		
		5'	4.555	4.580		
		6		4.451		
		6'		4.451		
		Fuc	4.872			
					5.090	
H-2		3	4.249		4.235	
		4	4.189		4.107	
		4'	4.118		4.142	
					4.064	
					4.00*	4.02*
					4.07*	4.10*
					4.064	
					4.064	
H-5	Fuc	4.12*				
CH ₃	Fuc	1.21*				
NAc		1	2.000 ^b		2.000 ^b	
			2.005 ^b		2.005 ^b	
			2.012 ^b		2.012 ^b	
		2	2.089		2.060	
		5	2.051			
		5'	2.051			

* Values could not be determined more accurately (± 0.01 ppm) due to the complexity of the mixture.

^b Signals indicate stemming from the main components with respect to the heterogeneity of the peptide moiety

at $\delta = 4.451$ ppm, which is attributed to Gal-6/6' H-1, this indicates that 10% of the mixture has Gal attached in $\beta(1-4)$ linkage to the biantennary structure. A set of lower-intensity signals in the region $5.4 > \delta > 4.8$ ppm is indicative for minor components of the oligomannoside type. It was estimated from the relative intensities of the α -anomeric protons that 30% of the mixture consists of three minor components, which all share the Man6 moiety consisting of Man-3, -4, -4', -B and -C [20]. This can be concluded from Man-4 H-1 ($\delta = 4.118$ ppm) and Man-4' H-1 ($\delta = 4.142$ ppm) signals.

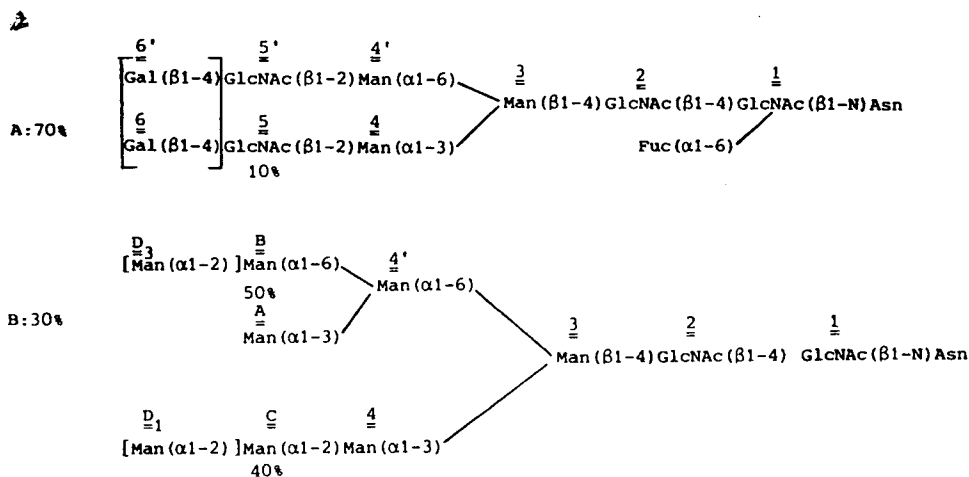


Fig. 1. Primary structures of glycans of glycopeptide Sta AP from pancreatic lipase

5.345 ppm) which is characteristic for Man-C attachment, and Man-4' H-1 ($\delta = 4.87$ ppm), which is indicative for the presence of Man-A and -B. Man-A H-1 is detected at $\delta = 5.090$ ppm. For H-1 of Man-B two signals are observed, firstly at $\delta = 4.91$ ppm, pointing to Man-B in terminal position, and at $\delta = 5.146$ ppm, which is typical for Man-B substituted with Man-D₃. From the intensity ratio of these two signals it was estimated that Man-B is substituted for 50% with Man-D₃. Furthermore two signals, of equal intensity, for Man-C H-1 are present. A signal at $\delta = 5.051$ ppm is indicative for terminal Man-C, and a signal at $\delta = 5.304$ ppm shows (40%) substitution of Man-C with Man-D₁. Both Man-D₁ and Man-D₃ H-1 atoms resonate at $\delta = 5.042$ ppm.

On the basis of carbohydrate composition, permethylation analysis and 500-MHz ¹H-NMR spectroscopy, the primary structure of glycans from glycopeptide fraction Sta AP as given in Fig. 1 was deduced. The fraction is heterogenous and consists of a mixture of two glycans (glycans A and B). On the basis of molar ratios of monosaccharides and monosaccharide methyl ethers and the relative intensity of the anomeric proton signals, the mixture contains 70% glycan A and 30% glycan B. The presence of two different structures, the first one of the N-acetylglucosamine type and the other of the oligomannoside type on the same protein is not unusual and has been previously described in various glycoproteins like thyroglobulin [21], 'Ashwell receptor' of rabbit liver [22] and in bovine transferrin [23].

This investigation was supported by the *Université des Sciences et Techniques de Lille Flandres-Artois*, the *Centre National de la Recherche Scientifique* (UA 217 and C. B. M. Marseille) and by the *Netherlands Foundation for Chemical Research (SON)* with financial aid from the *Netherlands organization for the Advancement of Pure Research (ZWO)*.

REFERENCES

1. Semèriva, M. & Desnuelle, P. (1979) *Adv. Enzymol.* 48, 319–370.
2. Verger, R. (1970) Ph. D. thesis, Université d'Aix-Marseille.

3. Garner, C. W. Jr & Smith, L. C. (1972) *J. Biol. Chem.* 247, 561–565.
4. De Caro, A., Figarella, C., Amic, J., Michel, R. & Guy, O. (1977) *Biochim. Biophys. Acta* 490, 411–419.
5. Canioni, P., Benajiba, A., Julien, R., Rathelot, J., Benabdeljlil, A. & Sarda, L. (1975) *Biochimie (Paris)* 57, 35–41.
6. Rathelot, J., Julien, R., Bosc-Bierne, J., Gargouri, Y., Canioni, P. & Sarda, L. (1981) *Biochimie (Paris)* 63, 227–234.
7. Verger, R., de Haas, G. H., Sarda, L. & Desnuelle, P. (1969) *Biochim. Biophys. Acta* 188, 272–282.
8. Plummer, T. H. & Sarda, L. (1973) *J. Biol. Chem.* 248, 7865–7869.
9. Rovey, M., Boudouard, M. & Bianchetta, J. (1978) *Biochim. Biophys. Acta* 525, 373–379.
10. De Caro, J., Boudouard, M., Bonicel, J., Guidoni, A., Desnuelle, P. & Rovey, M. (1981) *Biochim. Biophys. Acta* 671, 129–138.
11. Bianchetta, J., Bidaud, J., Guidoni, A., Bonicel, J. & Rovey, M. (1979) *Eur. J. Biochem.* 97, 395–405.
12. Charles, M., Erlanson, C., Bianchetta, J., Joffre, J., Guidoni, A. & Rovey, M. (1974) *Biochim. Biophys. Acta* 359, 186–197.
13. Zanetta, J. P., Breckenridge, W. C. & Vincendon, G. (1972) *J. Chromatogr.* 69, 291–304.
14. Hakomori, S. I. (1964) *J. Biochem. (Tokyo)* 55, 205–208.
15. Fournet, B., Strecker, G., Leroy, Y. & Montreuil, J. (1981) *Anal. Biochem.* 116, 498–502.
16. Ernst, R. R. (1966) *Adv. Magn. Resonance* 2, 1–135.
17. Vliegthart, J. F. G., Dorland, L. & Van Halbeek, H. (1983) *Adv. Carbohydr. Chem. Biochem.* 41, 209–374.
18. Benkouka, F., Guidoni, A. A., De Caro, J. D., Bonicel, J. J., Desnuelle, P. A. & Rovey, M. (1982) *Eur. J. Biochem.* 128, 331–341.
19. Van Kuik, J. A., Hoffmann, R. A., Mutsaers, J. H. G. M., Van Halbeek, H., Kamerling, J. P. & Vliegthart, J. F. G. (1986) *Glycoconjugate J.* 3, 27–34.
20. Neeser, J. R., Vedovo, S. D., Mutsaers, J. H. G. M. & Vliegthart, J. F. G. (1985) *Glycoconjugate J.* 2, 355–364.
21. Ito, S., Yamashita, K., Spiro, R. G. & Kobata, A. (1977) *J. Biochem. (Tokyo)* 81, 1621–1631.
22. Kawasaki, T. & Ashwell, G. (1976) *J. Biol. Chem.* 251, 5292–5299.
23. Spik, G., Coddeville, B., Legrand, D., Mazurier, J., Leger, D., Goavec, M. & Montreuil, J. (1985) In *Proteins of iron storage and transport* (Spik, G., Montreuil, J., Grichton, R. R. & Mazurier, J., eds) pp. 47–51, Elsevier, Amsterdam.