

Unified prediction of organic matter preservation and degradation

Jack J. Middelburg ¹✉

A mechanistic and predictive hierarchical model explains observed organic carbon burial efficiencies and degradation kinetics, and reveals that upper and lower limits of organic matter activation energies are key to predicting organic matter preservation and degradation rates.

Organic matter cycling forms an important part of the global carbon cycle and has been studied for many decades across multiple environments. Organic matter is consumed by microbes and animals for energy and biomass production, and through this process of mineralization, inorganic carbon and nutrients are released that sustain primary production and biogeochemical cycles. Although typically less than 1% of organic matter ultimately avoids mineralization, its accumulation in soils and sediments is an important long-term carbon sink. The reactivity and persistence of organic matter vary greatly and understanding what controls preservation and degradation rates has proven a challenge. Although degradation and preservation are two sides of the same coin, many studies focus on one or the other and rarely consider both dissolved and particulate forms. Writing in *Communications Earth & Environment*, Shang¹ presents a novel unified model that simultaneously predicts bulk organic matter preservation and degradation in aquatic systems.

Organic matter (OM) preservation and degradation is governed by multiple factors relating to OM chemistry, environmental conditions and the biological actors causing OM breakdown². Natural OM is a heterogeneous mixture of thousands of organic compounds that differ in composition, structure, origin, thermodynamic properties, degradation history, and how they interact with other molecules and minerals. Such factors influence degradability, with different molecules being preferentially consumed or preserved. This chemical view has its shortcomings, however, as OM preservation and degradation rates and pathways are also governed by multiple physical and chemical properties of the environment, including oxygen and nutrient availability, and temperature. In addition, organisms have specific traits and machinery to produce and consume OM and so the number, diversity and identity of consumers also play a major role. Accordingly, any generic model for OM processing should incorporate not only the heterogeneity in OM composition, but also the environmental conditions and heterotrophic community composition and functioning^{3,4}.

In their article, Shang¹ presents a new kinetic model of OM degradation and preservation, inspired by (quantum) statistical theory, that is mechanistic and predictive, which sets it apart from previous empirical and phenological models. Crucially, Shang's model quantitatively predicts both the observed logarithmic relation between organic carbon burial in sediments and oxygen exposure time⁵ as well as empirical power-law degradation kinetics⁶. As such, Shang's hierarchical model unites OM preservation efficiencies and degradation rate patterns in a single model (Fig. 1). The model also predicts that more recalcitrant OM has a higher activation energy, which is consistent with observations⁷. Moreover, the upper and lower limits of OM activation energies are identified as key parameters to predict OM preservation and degradation rates over a wide range. This mechanistic incorporation of temperature sensitivity in the hierarchical model will be instrumental to further our knowledge on how OM preservation and degradation will respond to projected changes in temperature.

¹Department of Earth Sciences, Utrecht University, Princetonlaan 8A, 3584 CD Utrecht, The Netherlands. ✉email: j.b.middelburg@uu.nl

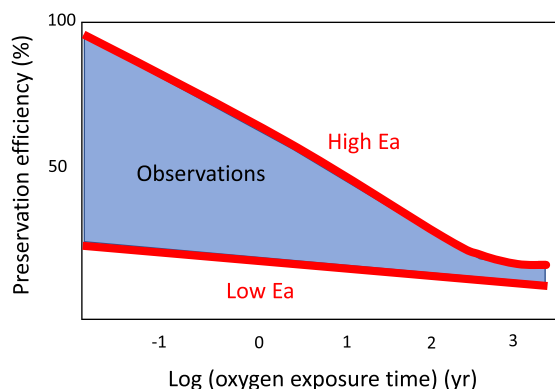


Fig. 1 Predicting organic matter preservation and degradation.

Theoretical predictions (red lines) of the relationship between organic matter preservation efficiency and oxygen-exposure time encompass observations. The upper and lower limit are set by the organic matter activation energies (adapted from ref. ¹).

The hierarchical model predictions should be tested with measurable OM reactivity proxies such as the amino acid degradation index⁸ and serial oxidation inferred activation energies⁹. Shang's model is predictive but rooted in traditional first-order kinetic models that do not explicitly resolve consumer communities and their metabolism. In contrast, a recently proposed quantitative model for OM accumulation that explicitly resolved the ecological dynamics of the consumers combined the stochastic release of multiple pools of OM with microbial uptake by a community of specialist and generalist consumers⁴. Although both models are generic and largely consistent with observations, they differ in basic design, that is, geochemical kinetics versus ecological dynamics as well as in their overall goals: one seeks to predict OM degradation rates and preservation whereas the other was built to explain the accumulation of dissolved OM. Integration of these two approaches would be the logical next step towards a unified theory and predictive model of OM degradation and preservation across ecosystems and covering multiple timescales.

The unified model of Shang¹ shows that organic matter in aquatic systems is broken down in a hierarchical ensemble, involving a decrease in effective kinetic barriers. Through incorporating the temperature sensitivity of organic matter degradation, we can better understand and predict future changes to this globally important carbon pool under future warming.

Received: 5 January 2023; Accepted: 12 January 2023;

Published online: 23 January 2023

References

- Shang, H. A generic hierarchical model of organic matter degradation and preservation in aquatic systems. *Commun. Earth Environ.* <https://doi.org/10.1038/s43247-022-00667-4> (2023).
- Middelburg, J. J. & Meysman, F. J. R. Burial at sea. *Science* **316**, 1294–1295 (2007).
- Kothawala, D. N. et al. Organic matter degradation across ecosystem boundaries: the need for a unified conceptualization. *Trends Ecol. Evol.* **36**, 113–122 (2021).
- Zakem, E. J. et al. A unified theory for organic matter accumulation. *Proc. Natl. Acad. Sci. USA* **118**, 1–8 (2021).
- Hartnett, H. E. et al. Influence of oxygen exposure time on organic carbon preservation in continental margin sediments. *Nature* **391**, 572–575 (1998).
- Middelburg, J. J. A simple rate model for organic matter decomposition in marine sediments. *Geochim. Cosmochim. Acta* **53**, 1577–1581 (1989).
- Westrich, J. T. & Berner, R. A. The effect of temperature on rates of sulfate reduction in marine sediments. *Geomicrobiol. J.* **6**, 99–117 (1988).
- Dauwe, B. et al. Linking diagenetic alteration of amino acids and bulk organic matter reactivity. *Limnol. Oceanogr.* **44**, 1809–1814 (1999).
- Hemingway, J. D. et al. Technical note: An inverse method to relate organic carbon reactivity to isotope composition from serial oxidation. *Biogeochemistry* **14**, 5099–5114 (2017).

Acknowledgements

This research contributes to the Netherlands Earth System Science Centre, financially supported by Ministry of Education, Culture and Sciences in the Netherlands.

Competing interests

The author declares no competing interests.

Additional information

Correspondence and requests for materials should be addressed to Jack J. Middelburg.

Reprints and permission information is available at <http://www.nature.com/reprints>

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2023