

Announcing

Gold Open Access

Preprints welcome

flagship journal

Publishing charges waived

Edited by active scientists

our new

WILEY VCH

Excellence in Chemistry Research





Meet the Editors of ChemistryEurope



Luisa De Cola Università degli Studi di Milano Statale, Italy



Ive Hermans University of Wisconsin-Madison, USA



Ken Tanaka Tokyo Institute of Technology, Japan

Specific Lipid Studies in Complex Membranes by Solid-State NMR Spectroscopy

Roy A. M. van Beekveld⁺,^[a] Maik G. N. Derks⁺,^[a, b] Raj Kumar,^[a] Leanna Smid,^[a] Thorben Maass,^[a] João Medeiros-Silva,^[a, c] Eefjan Breukink,^[b] and Markus Weingarth^{*[a]}

Abstract: Specific interactions with phospholipids are often critical for the function of proteins or drugs, but studying these interactions at high resolution remains difficult, especially in complex membranes that mimic biological conditions. In principle, molecular interactions with phospholipids could be directly probed by solid-state NMR (ssNMR). However, due to the challenge to detect specific lipids in mixed liposomes and limited spectral sensitivity, ssNMR studies of specific lipids in complex membranes are scarce.

Introduction

Specific interactions with phospholipids are essential for the function of living cells, for example, by modulating membrane proteins^[1] or by shaping membrane properties.^[2] Furthermore, an increasingly large number of drugs target specific phospholipids.^[3] However, the study of specific interactions with phospholipids at atomic resolution remains a challenge, particularly in complex fluid membranes that mimic biological conditions. This is due to the general difficulty to work in membranes, and due to the dynamic nature of small lipid molecules that evade most structural biology methods.

- [a] R. A. M. van Beekveld,⁺ M. G. N. Derks,⁺ R. Kumar, L. Smid, T. Maass, Dr. J. Medeiros-Silva, Dr. M. Weingarth NMR Spectroscopy Department of Chemistry Faculty of Science Utrecht University Padualaan 8, 3584 CH Utrecht (The Netherlands) E-mail: m.h.weingarth@uu.nl
 [b] M. G. N. Derks,⁺ Dr. E. Breukink Membrane Biochemistry and Biophysics Department of Chemistry
- Faculty of Science Utrecht University Padualaan 8, 3584 CH Utrecht (The Netherlands)
- [c] Dr. J. Medeiros-Silva Present address: Department of Chemistry Massachusetts Institute of Technology 170 Albany Street, Cambridge, MA 02139 (USA)
- [⁺] These authors contributed equally to this work.
- Supporting information for this article is available on the WWW under https://doi.org/10.1002/chem.202202472
- © © 2022 The Authors. Chemistry A European Journal published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

Here, by using purified biological ¹³C,¹⁵N-labeled phospholipids, we show that we can selectively detect traces of specific lipids in complex membranes. In combination with ¹Hdetected ssNMR, we show that our approach provides unprecedented high-resolution insights into the mechanisms of drugs that target specific lipids. This broadly applicable approach opens new opportunities for the molecular characterization of specific lipid interactions with proteins or drugs in complex fluid membranes.

Solid-state NMR (ssNMR) could be an ideal method to investigate specific phospholipids^[4] in complex membranes, but such studies are scarce. This is due to the challenge to distinguish specific phospholipids in mixed membranes and due to insufficient ssNMR signal sensitivity. These problems could be overcome with ¹³C,¹⁵N-labeled phospholipids that could be selectively detected with high sensitivity in mixed membranes. However, uniformly ¹³C,¹⁵N-labeled phospholipids are not commercially available, and their purification has hitherto not been reported, severely curtailing the scope of ssNMR studies of phospholipids. What is more, next to the problem of isotope-labeling, synthetic phospholipids sometimes lack critical properties of biological phospholipids, such as headgroup stereochemistry and lipid-tail features. Synthetic phospholipids usually contain defined uniform lipid tails, while biological lipids have complex tails.^[5]

Here, we report the purification of uniformly ${}^{13}C, {}^{15}N$ -labeled phospholipids (henceforth called ${}^{13}C, {}^{15}N$ -lipids) from cells, including lipids of high biological importance. We show that we can detect traces of ${}^{13}C, {}^{15}N$ -lipids in mixed membranes. Furthermore, we show that, in combination with modern ${}^{1}H$ -detected ssNMR,^[6] our approach provides unprecedented molecular insights into the mechanisms of drugs that target specific lipids in complex membranes.

While our approach is broadly applicable, in this manuscript, we show the isolation of ¹³C,¹⁵N-lipids with an *anionic* cardiolipin, an *anionic* phosphatidylglycerol (PG), a *zwitterionic* phosphatidylethanolamine (PE), and a *zwitterionic* phosphatidyl-choline (PC) headgroup (Figure 1A). All these lipids are involved in critical biological functions, with cardiolipin playing a central role in the respiratory chain,^[7] ATP synthesis,^[8] apoptosis,^[9] and mitochondrial transport,^[10] to name only a few examples.



Figure 1. ssNMR characterization of phospholipid headgroups. A) Chemical structures of PE, PG, and cardiolipin lipid-headgroups. B–D) Zoom into the lipid headgroup region of 2D CC PARISxy ssNMR spectra (200 ms CC magnetization transfer time) acquired with 5 mol % of B) ¹³C, ¹⁵N-PE, C) ¹³C-PG, and D) ¹³C-cardiolipin in DOPC liposomes at 1200 MHz magnetic field and a sample temperature of 275 K. E) 2D CC (150 ms CC transfer time) of ¹³C, ¹⁵N-PE in DOPC liposomes acquired at 500 MHz at 255 K temperature. F) 2D CC (200 ms CC transfer time) of ¹³C, ¹⁵N-PG in DOPC liposomes acquired at 700 MHz at 245 K temperature. G) Cross-polarization build-up curves measured at 1200 MHz and 17 kHz MAS using 275 K temperature show that the ¹³C-PG headgroup is more mobile than its glycerol backbone. See Figures S6-S8 for full assignments of labeled lipids.

Results and Discussion

We isolated ¹³C,¹⁵N-lipids from bacteria or yeast grown in isotope-enriched tailored media (see Experimental Section and Supporting Information). Total lipids from bacteria were extracted according to the method of Bligh-Dyer^[11] in the presence of lysozyme to increase cardiolipin extraction efficiency,^[12] while yeast lipids were extracted under mechanical disruption.^[13] Next, we developed a simple and fast preparative purification method, based on a single isocratic elution on silica, to purify multiple phospholipid classes from complex mixtures (see Experimental Section and Supporting Information), yielding pure fractions of different lipid classes that we analyzed by solution state NMR and gas chromatography to establish purity, lipid-headgroup identity, and lipid-tail composition. By comparison to synthetic standards, solution NMR data unambiguously confirmed the headgroup identity and the high purity (based on headgroup type) of our isolations (Figures S1-4). The nature of the lipid tails varied according to the organism and is discussed further below.

Next, we performed ssNMR experiments using mixed liposomes that contained $5 \text{ mol }\%^{13}\text{C},^{15}\text{N}$ -lipids and 95 mol% DOPC (Figure 1). Since the assignments of lipid headgroups and the glycerol backbone in membranes are heavily compromised by signal overlap, we first acquired 2D $^{13}\text{C}-^{13}\text{C}$ PARISxy^[14] spectra at ultra-high magnetic field of 1200 MHz (¹H-frequency) and a



Research Article doi.org/10.1002/chem.202202472

sample temperature of 275 K at which the investigated membranes were in the liquid-crystalline phase with a high degree of mobility. At these experimental conditions, we could resolve all headgroup and backbone signals. This is the first time that the complete PG headgroup could be assigned in membranes. While the chemical shifts of the glycerol backbone (g1-g3) are similar for all measured lipids, we measured substantial chemical shift differences in the headgroups. Strikingly, in the liquid-crystalline phase, most cross-peaks in the PG and PE headgroups were not observable in dipolar spectra, which implies that the headgroups are highly flexible (Figure 1B,C,G). Indeed, by lowering the sample temperature to 255 or 245 K, at which the ¹³C,¹⁵N-lipids were in the gel phase in which the mobility of individual lipids decreases sharply, we observed clear cross-peaks for the entire PE and PG headgroups (Figure 1E,F). The unusual cardiolipin headgroup, whose mobility is severely restricted, is already fully visible in the liquidcrystalline phase at 275 K (Figure 1D). Note that in contrast to commercially available PG, which contains a racemic glycerol headgroup,^[15] our purified bacterial PG-lipids contain a glycerol headgroup in enantiomeric excess, as verified by CD spectroscopy (Figure S5).

Organisms feature a broad spectrum of lipid tails that are vital to shape membrane properties. The most well-known example is lipid unsaturation that keeps membranes of eukaryotes fluid, but especially bacterial membranes feature a large diversity of lipid tails that include branching and cyclic moieties. This broad spectrum of lipid tails is important to finetune membrane properties,^[16] and likely also for specific interactions with membrane proteins.^[17] First, we comprehensively characterized the nature of the tails using solution NMR and gas chromatography (Figure 2A,B and Figure S9). The lipid tails that were isolated from Escherichia coli (with a PE headgroup) were mostly straight and saturated, with 20% of the tails containing cyclopropyl-groups,^[5b] whereas lipids isolated from Staphylococcus simulans or Micrococcus flavus (with PG or cardiolipin headgroups) had tails that were fully saturated and branched (so-called iso or anteiso methyl-branched) at the termini.^[5a] In Saccharomyces cerevisiae (yeast), lipid tails were straight and unsaturated (Table S1 and Figure S6). We note that the composition of lipid tails can readily be altered by adjusting culture conditions such as temperature, salt concentrations, addition of detergents, or choosing a different species (Figure S10).^[5,18]

In complex liposomes, we could obtain well-resolved 2D CC ssNMR spectra that enabled us to assign large parts of the tails with the exception of congested bulk signals (Figure 2C,D). In dipolar spectra acquired in the liquid-crystalline phase, most correlations of the lipid-tail termini could not be observed as the lipid mobility increases towards the membrane center. Conversely, these missing correlations involving mobile tailatoms could be observed in a scalar 2D CC TOBSY⁽¹⁹⁾ ssNMR spectrum (Figure S11). At lower temperature in the gel phase, we could observe all cross-peaks for the tails, including for the termini, in dipolar spectra (Figures S12–13).

The available high-resolution knowledge on the structure and the dynamics of lipid headgroups in complex liposomes is



Figure 2. Characterization of phospholipid tails. A) Gas chromatography derived tail compositions of isolated biological ¹³C, ¹⁵N-lipids. B) Structures and nomenclatures of lipid tails. C) Aliphatic region of 2D CC PARISxy spectra of ¹³C-PG-lipids (5 mol %) from *S. simulans* in DOPC liposomes. The red spectrum was acquired at 1200 MHz and 275 K sample temperature (liquid-crystalline phase), the blue spectrum at 500 MHz and 255 K (gel phase). D) Similar to C) but for ¹³C, ¹⁵N-PE-lipids from *E. coli*. The red spectrum was acquired at 1200 MHz and 275 K, the blue spectrum at 500 MHz and 255 K.

© 2022 The Authors. Chemistry - A European Journal published by Wiley-VCH GmbH

still largely derived from computational studies (see, for example, ref.^[20]). In a similar vein, this is also the case for the molecular behavior of lipid tails in complex membranes. Thereby, ¹³C,¹⁵N-lipids provide unprecedented access to molecular parameters of lipids, which we demonstrate below on the example of a lipid-targeting drug.

An increasing number of drugs target specific phospholipids. This includes antibiotics,^[3a,b] such as the clinically used daptomycin,^[21] and also antimycotics.^[3c] For all these drugs, due to the above-mentioned technical challenges, the molecular details of their lipid interactions are scarce and usually elusive in membranes. As a test case, we studied cinnamycin, a 19residue lanthipeptide (Figure S14A) that is well-known to target the headgroup of PE-lipids.^[22] Cinnamycin is a naturally produced drug that is, in addition, used to monitor PE domains in bacterial and cancer cells,^[23] and has potential antiviral applications.^[24] A previous solution NMR structural model^[25] determined in DMSO using C12-lysoPE (a truncated, soluble mimic of PE-lipids) suggested that cinnamycin forms a binding pocket that specifically accommodates the small cationic PE headgroup, but not bulkier headgroups such as that of PC. Contacts to the phosphate group, the glycerol backbone, or even the lipid tails were not observed in DMSO. However, as outlined by a recent molecular dynamics study that observed a deep membrane penetration of cinnamycin,^[26] this binding model is difficult to align with experimental observations that i) cinnamycin has no affinity to PE-sphingolipids, which have a PE headgroup but a ceramide backbone instead of a glycerol backbone,^[23] ii) cinnamycin is not antagonized by phosphopropanolamine^[27] suggesting that the distance between phosphatidylethanolamine headgroup and glycerol backbone matters, and iii) all members of the cinnamycin-like peptide family such as duramycin have a hydrophobic stretch in common (Phe7-Ala14), suggesting interactions with hydrophobic moieties (Figure S14B). Hence, it appears unlikely that only the PE headgroup is involved in the binding interface.

We prepared DOPC liposomes containing 5 mol% of ${}^{13}C$, ${}^{15}N$ -PE-lipids in the presence or the absence of 5 mol% cinnamycin and acquired 2D CC PARISxy, ${}^{[14]}$ as well as ${}^{1}H$ -detected dipolarbased 2D NH and 2D CH ssNMR spectra. Upon addition of cinnamycin, we observed large chemical shift perturbations (CSPs) for the headgroup carbons, as well as a massive (-7.4 ${}^{15}N$ ppm) CSP for the NH₃ ammonium group (Figure 3A,B,D and Figure S15). This indeed shows that the PE headgroup is tightly coordinated by cinnamycin in membranes. Furthermore, ${}^{31}P$ ssNMR studies show a CSP (+1.1 ppm, Figure S16) in the presence of cinnamycin, which is in line with MD simulations that suggested hydrogen bonding with the PE-lipid phosphate group.^[26]

Strikingly, we also observed stark CSPs for the carbons of the glycerol backbone and the beginning of the lipid-tail (Figure 3A). Furthermore, cross-peaks between the beginning of the lipid tails (*sn*-1 and *sn*-2 carbonyl signals) and the glycerol backbone could only be observed in the presence of the drug, but not in the absence of the drug. This demonstrates that the glycerol backbone and the lipid tails stiffen in the presence of the drug. Finally, the shifts of the *sn*-1 and *sn*-2 carbonyl signals imply a structural change in the lipid tails. Together, these data clearly demonstrate that cinnamycin does not only bind to the PE-headgroup, but also to the glycerol backbone and the lipid tails, i.e., an interface that strongly diverges from the structural model obtained in DMSO with truncated PE-like lipids (Figure 3G).^[28] This is yet another example that a proper membrane environment is essential to obtain relevant binding modes for small molecules that target lipids in membranes.^[29]

Finally, to quantify the observed stiffening of PE-lipids upon target binding, we probed their dynamics by measuring atomresolved ¹³C and ¹⁵N rotating frame relaxation (R₁₀) data using a series of ¹H-detected 2D CH and 1D NH experiments (Figure 3C,E,F and Figure S17).^{[30] 15}N relaxation data revealed a substantial stiffening of the free ammonium-group of the PEheadgroup (~20.1 ms ^{15}N R₁₀ in the absence of cinnamycin; ~14.4 ms ${}^{15}N$ R₁₀ in the presence of cinnamycin). Strikingly, in the presence of cinnamycin, ¹³C relaxation data show a marked stiffening of the entire PE-headgroup (~20 ms ^{13}C R₁₀) that is substantially more rigid than the glycerol backbone (~40 ms¹³C R_{10} (Figure 3H). This is in stark contrast to the dynamics of free PE-lipids, where the headgroup is highly mobile (Figure 1B). Note that the headgroup of free PE-lipids gave only weak and broad dipolar signals in 2D CH spectra at 60 kHz MAS and a sample temperature of approximately 310 K, presumably because of strong dynamics that interfered with magnetization transfer (data not shown).

Conclusions

Here, we have introduced an ssNMR approach that enables detailed high-resolution insights into phospholipids in complex fluid membranes. As shown on the example of the peptide drug cinnamycin, this broadly applicable approach provides new vistas for the characterization of small molecules, drugs, or proteins that specifically interact with phospholipids. In particular, the combined sensitivity gains of ¹³C,¹⁵N-lipids and modern ¹H-detected methods at fast sample spinning enable detailed multi-dimensional relaxation studies in mixed liposomes with traces (~ 5 mol%) of the lipid of interest. Thereby, our approach opens new opportunities to characterize the molecular properties of phospholipids and their interactions in biologically relevant membranes. Previously, ssNMR studies of specific lipids in complex membranes were usually limited to either the phosphorus as the sole reporter^[31] or to rare, serendipitous events of co-purified endogenous ¹³C, ¹⁵N-lipids.^[4g,h, 32] The here introduced approach is also complementary to the use of deuterated phospholipids, which are commercially available and are widely used to study microsecond motion of lipids using static ssNMR spectroscopy.[33] We note that our labelling strategy can be further finetuned by using specifically labeled precursors to reduce spectral crowding, which seems especially useful to resolve spectral congestion of the lipid headgroup/ glycerol backbone signals. As proof-of-principle we grew bacteria on 1-13C1-D-Glc or 2-13C1-D-Glc, [34] yielding a specifically labeled PE-lipids (Figure S18). Such a sparse labelling strategy Research Article doi.org/10.1002/chem.202202472 European Chemical Societies Publishing





Figure 3. Specific interactions between the peptide-drug cinnamycin and PE-lipids in mixed liposomes. A) 2D CC PARISxy spectra of ¹³C,¹⁵N PE-lipids measured in the presence (in magenta) and absence (in blue) of cinnamycin. Spectra were obtained at 500 MHz and 260 K temperature using 150 ms CC magnetization transfer. B) ¹H-detected 2D NH spectra acquired in the presence (in magenta) and absence (blue) of cinnamycin, obtained at 700 MHz. C) ¹⁵N R_{1p} relaxation rates for the ammonium-group of the PE-headgroup, with (magenta) and without (blue) cinnamycin. D) ¹H-detected 2D CH spectra of cinnamycin-bound ¹³C, ¹⁵N PE-lipids (in magenta), superimposed on the solution NMR spectrum of free PE-lipids (in green). E) ¹³C R_{1p} relaxation rates for relipids bound to cinnamycin. F) ¹³C R_{1p} signal decay curves for selected headgroup and glycerol backbone atoms. G) ssNMR studies show that not only the PE-headgroup, but also the phosphate, the glycerol backbone, and the beginning of the lipid-tail are part of the interface with cinnamycin. A to conserved hydrophobic residues point towards the lipid tails. H) Illustration of the site-resolved dynamics. The size of the spheres represents the R_{1p} relaxation rates. For all measurements, we used DOPC membranes with a small (5 mol%) fraction of ¹³C, ¹⁵N PE-lipids, and added an equimolar, stoichiometric amount of cinnamycin.

could be valuable to probe specific lipid binding to membrane proteins, as recently shown on the example of sterol-lipids.^[35]

Experimental Section

Strains

Staphylococcus simulans 22, Micrococcus flavus DSM 1790, Escherichia coli BL21(DE3), Bacillus cereus ATCC 10987, and Saccharomyces cerevisiae BY4742 were used in this work. **Cell growth**: Bacteria were revived from glycerol stocks on fresh plates of LB 1.8 w/v% agar (*S. simulans, E. coli, B. cereus*) or TSB 1.8 w/v% agar (*M. flavus*) for 24 h at either 30°C (*M. flavus, B. cereus*) or 37°C (*S. simulans, E. coli*). From this fresh plate, a single colony was inoculated in an overnight culture (50 mL in a 250 mL unbaffled flask) in the same medium without agar. The medium was removed by centrifugation (4,000×g, 10 min, RT), and the bacterial pellets resuspended to $OD_{600} = 0.05$ in 1 L of medium A (Table S2) in a 2 L baffled flask. Bacteria were grown at either 30°C or 37°C as before, with 200 or 220 RPM shaking respectively and harvested by centrifugation (4,000×g, 4°C, 15–30 min) at late-log phase.

Sa. cerevisiae was grown in Standard Defined (SD) medium (Table S3). Glycerol stocks were revived on a SD 1.8 w/v% agar plate for 48 h at 30 °C. Single colonies were inoculated into liquid SD medium (50 mL in a 250 mL unbaffled flask) and grown overnight. Finally, precultures were diluted in 1 L medium B (Table S4) in a 2 L baffled flask to $OD_{600} = 0.05$ and grown for 40 h at 30 °C before harvesting by centrifugation.

Lipid isolation: Bacteria were resuspended in 250 mM phosphate buffer (pH = 7.4) to an OD₆₀₀ of ~ 50, after which they were treated with lysozyme and incubated at 37 °C for 2 h with shaking. Subsequently, two volumes of MeOH and one volume of CHCl₃ were added and the resulting suspension was stirred vigorously for 30 minutes at RT. Then, one volume of CHCl₃ and 0.8 volumes of H₂O were added, and the organic phase was collected and evaporated to dryness. Dried lipid films were resuspended in CHCl₃ assisted by sonication and loaded on a silica column equilibrated in CHCl₃ (silica 60, 0.040-0.063 mm). The column was washed with CHCl₃ and phospholipids were eluted with CHCl₃/EtOH/25%NH₃/ H_2O 48:48:1:3 $(v/v/v) + 0.2 \; g/L \; NH_4Ac.$ Fractions were analyzed on TLC using the same eluent (HPTLC Nano-ADAMANT® 0.20 mm silica 60, stained by iodine vapor and/or molybdenum blue^[36]). Pure fractions were combined, evaporated and redissolved in 2:1 CHCl₃/ MeOH (v/v). Small quantities of phospholipids were isolated using preparative TLC. 5 mm horizontal lanes of lipid extracts in 2:1 CHCl₃/MeOH (v/v) were spotted using a CAMAG® Linomat 5 semiautomatic sample application system and developed in CHCl₃/ EtOH/25%NH₃/H₂O 48:48:1:3 (v/v/v/v) + 0.2 g/L NH₄Ac. TLC plates were dried under a flow of N₂, after which a small vertical lane was cut and stained with iodine vapor to identify lipid spots. Horizontal lanes of interest were scratched off and lipids were extracted from the silica by extensive vortexing in 2:1 CHCl₃/MeOH (v/v) followed by centrifugation $(4,700 \times g)$. Silica was extracted thrice to ensure quantitative transfer.

For Saccharomyces cerevisiae lipids a different extraction procedure was used. Lyophilized cells were resuspended in 2:1 CHCl₃/50 mM HCl (v/v) and subjected to multiple sonication (Branson 3800) and stirring cycles, followed by filtration of the suspension over a glass filter (porosity index 0.3). The organic phase was then collected, evaporated to dryness, and the residue purified as described above for bacteria.

We obtained typical yields of > 10 mg of the highest abundant lipid class of the microorganism of choice at > 99% purity based on solution NMR analysis.

Solid state NMR sample preparation: Appropriate volumes of lipid stocks in $CHCl_3$ or $CHCl_3/MeOH$ were mixed, evaporated under a stream of N₂ gas and exposed to high vacuum for 20 minutes. The resulting lipid films were resuspended in buffer (50 mM HEPES, 150 mM NaCl, pH=7.2) by extensive vortexing. The formed lipid vesicles were then collected by ultra-centrifugation ($100k \times g$, $4^{\circ}C$, 30 min) and spun down into 3.2 or 1.3 mm solid state NMR rotors. Samples containing cinnamycin were made by including cinnamy-

cin (at 1:1 U-[^{13}C , ^{15}N]-PE to cinnamycin stoichiometry) in the resuspension buffer.

NMR spectroscopy: Solution state NMR spectra of isolated phospholipid and appropriate commercial standards were recorded in 2:1 CD₃OD/CDCl₃ (v/v) at 298 K and at 14.1 T (600 MHz ¹H frequency). Chemical shifts were referenced to DSS using internal TMS. Assignments were done using standard experiments such as HSQC, COSY, TOCSY, HMBC on the lipid standards and compared to the literature (e.g. BMRB^[37] entry bmse001105). For purified U-[¹³C,¹⁵N] enriched lipids assignments were copied from the standards or literature^[38] and confirmed with 2D ¹H-¹³C-CT-HSQC-TOCSY experiments.

Solid state NMR spectra were recorded at 11.7, 16.4, or 28.2 T (500, 700, and 1200 MHz ¹H frequency, respectively). Dipolar based 2D ¹³C-¹³C spin diffusion experiments were acquired with PARISxy^[14] recoupling (m = 1) and SPINAL64^[39] decoupling. At 1200 MHz we used 17 kHz MAS and a mixing time of 200 ms, at 700 MHz we used 10 kHz MAS and a mixing time of 200 ms, and at 500 MHz we used 12 kHz MAS and a mixing time of 150 ms. TOBSY^[19] scalar couplingbased 2D ¹³C¹³C experiments were conducted at 305 K and 8 kHz MAS with a mixing time of 6 ms. An INEPT block was used for the initial ¹H-¹³C magnetization transfer and WALTZ16 decoupling was applied in both dimensions. ³¹P spectra were recorded at 11.7 T (500 MHz ¹H frequency) and 10 kHz MAS and a contact time of 1.5 ms using SPINAL64^[39] decoupling. ¹H-detected experiments were performed at 60 kHz MAS and 16.4 T (700 MHz ¹H frequency) using low-power $\mathsf{PISSARRO}^{[40]}$ decoupling in all dimensions and a sample temperature of 305 K. ¹H-detected R_{1rho} relaxation experiments were acquired with a ¹⁵N/¹³C spin-lock field of 18 kHz and durations of 0, 10, 25, 50 and 75 ms for $^{15}\mathrm{N}$ and 0, 10, 30, 60 and 120 ms for ¹³C. Chemical shifts were referenced externally using adamantane (CH at 31.48 ppm). Effective sample temperatures were calibrated using the ⁷⁹Br resonance of KBr as previously described.[41]

Gas chromatography (GC): Dried phospholipids (0.2-0.5 mg) were redissolved in 1 mL n-hexane by extensive shaking, followed by the addition of 200 µL 100 g/L KOH in MeOH. Samples were vigorously shaken for at least 1 min to mix both layers well. The n-hexane layer was then taken and dried under a flow of N_{2} at 40 $^{\circ}\text{C}.$ The obtained fatty acid methyl esters (FAMEs) were redissolved in 50 µL n-hexane and analyzed using gas chromatography with flame-ionization detection on a Trace GC Ultra (Thermo Fisher Scientific) equipped with a 30 m long biscyanopropyl polysiloxane column (internal diameter 0.25 mm; Restek) and N₂ as the carrier gas. A temperature gradient was applied, starting at 40 °C, held for one minute, then increasing linearly to 160°C in 4 min and finally to 220°C in 15 min. Peaks were identified using FAME standards Mixture BR2 (Larodan; 90-1052) for branched species and certain straight chain fatty acids or 63-B (Nu-Chek-Prep) for various unsaturated and straight chain fatty acids. Assignments not included in these mixtures were made based on expected elution patterns that could be deduced from the standards. FAME-species that are noted as "unidentified" in this work elute after nC18:0.

Circular dichroism (CD) spectroscopy: CD spectroscopy was performed on a Jasco J-810-150S spectropolarimeter, using a 1 mm quartz cuvette at 20 °C. Lipids were dissolved to 1 g/L in MeOH. Obtained spectra were corrected for a MeOH blank. Spectra were averaged over five scans and represent data for which the absorbance was \leq 2.0.



5213765,

2, 70, Downloaded from https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/chem.202202472 by Cochrane Netherlands, Wiley Online Library on [02/02/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/rems-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons License

Acknowledgements

This work was funded by the Dutch Research Council (NWO, grant numbers 723.014.003 & 711.018.001 to MW). Experiments at the 1200 MHz instrument were supported by uNMR-NL, an NWO-funded Roadmap NMR facility (no. 184.032.207). The present work was part of the research programme of the Netherlands Centre for One Health (www.ncoh.nl).

Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request. The solution and solid-state NMR assignments of lipids have been deposited in the BMRbig (https://bmrbig.bmrb.io/) Data Bank (accession codes BMRbig75, BMRbig76, BMRbig77, BMRbig78, BMRbig79, BMRbig80, BMRbig81).

Keywords: antibiotics · cinnamycin · lipids · membranes · solid state NMR

- a) A. Laganowsky, E. Reading, T. M. Allison, M. B. Ulmschneider, M. T. Degiacomi, A. J. Baldwin, C. V. Robinson, *Nature* **2014**, *510*, 172–175;
 b) R. Amani, C. G. Borcik, N. H. Khan, D. B. Versteeg, M. Yekefallah, H. Q. Do, H. R. Coats, B. J. Wylie, *Proc. Natl. Acad. Sci. USA* **2020**, *117*, 2938–2947; c) S. B. Hansen, X. Tao, R. MacKinnon, *Nature* **2011**, *477*, 495–498;
 d) J. E. Harlan, P. J. Hajduk, H. S. Yoon, S. W. Fesik, *Nature* **1994**, *371*, 168–170.
- [2] a) A. G. Lee, *Biochim. Biophys. Acta Biomembr.* 2003, *1612*, 1–40; b) H. I.
 Ingólfsson, M. N. Melo, F. J. van Eerden, C. Arnarez, C. A. Lopez, T. A.
 Wassenaar, X. Periole, A. H. de Vries, D. P. Tieleman, S. J. Marrink, *J. Am. Chem. Soc.* 2014, *136*, 14554–14559; c) E. Mileykovskaya, W. Dowhan,
 Biochim. Biophys. Acta Biomembr. 2009, *1788*, 2084–2091; d) M. P. Czech,
 Cell 2000, *100*, 603–606.
- [3] a) C. W. Johnston, M. A. Skinnider, C. A. Dejong, P. N. Rees, G. M. Chen, C. G. Walker, S. French, E. D. Brown, J. Berdy, D. Y. Liu, N. A. Magarvey, *Nat. Chem. Biol.* **2016**, *12*, 233–239; b) M. Song, Y. Liu, X. Huang, S. Ding, Y. Wang, J. Shen, K. Zhu, *Nat. Microbiol.* **2020**, *5*, 1040–1050; c) M. Järvå, F. T. Lay, T. K. Phan, C. Humble, I. K. H. Poon, M. R. Bleackley, M. A. Anderson, M. D. Hulett, M. Kvansakul, *Nat. Commun.* **2018**, *9*.
- [4] a) J. M. Boettcher, R. L. Davis-Harrison, M. C. Clay, A. J. Nieuwkoop, Y. Z. Ohkubo, E. Tajkhorshid, J. H. Morrissey, C. M. Rienstra, Biochemistry 2011, 50, 2264–2273; b) A. A. Smith, A. Vogel, O. Engberg, P. W. Hildebrand, D. Huster, Nat. Commun. 2022, 13; c) M. Li, A. Mandal, V. A. Tyurin, M. DeLucia, J. Ahn, V. E. Kagan, P. C. A. van der Wel, Structure 2019, 27, 806-815.e804; d) V. S. Mandala, J. K. Williams, M. Hong, Annu. Rev. Biophys. 2018, 47, 201-222; e) E. A. W. van der Cruijsen, D. Nand, M. Weingarth, A. Prokofyev, S. Hornig, A. A. Cukkemane, A. M. J. J. Bonvin, S. Becker, R. E. Hulse, E. Perozo, O. Pongs, M. Baldus, Proc. Natl. Acad. Sci. USA 2013, 110, 13008–13013; f) M. T. Eddy, T.-C. Ong, L. Clark, O. Teijido, P. C. A. van der Wel, R. Garces, G. Wagner, T. K. Rostovtseva, R. G. Griffin, J. Am. Chem. Soc. 2012, 134, 6375-6387; g) J. E. de Vlugt, P. Xiao, R. Munro, A. Charchoglyan, D. Brewer, M. S. Al-Abdul-Wahid, L. S. Brown, V. Ladizhansky, Biochim. Biophys. Acta Biomembr. 2020, 1862; h) P. Hariharan, E. Tikhonova, J. Medeiros-Silva, A. Jeucken, M. V. Bogdanov, W. Dowhan, J. F. Brouwers, M. Weingarth, L. Guan, BMC Biol. 2018, 16.
- [5] a) T. Kaneda, Microbiol. Rev. 1991, 55, 288–302; b) V. A. Knivett, J. Cullen, Biochem. J. 1965, 96, 771–776.

- [6] a) V. Chevelkov, K. Rehbein, A. Diehl, B. Reif, Angew. Chem. Int. Ed. 2006, 45, 3878–3881; b) D. H. Zhou, G. Shah, M. Cormos, C. Mullen, D. Sandoz, C. M. Rienstra, J. Am. Chem. Soc. 2007, 129, 11791–11801; c) V. Agarwal, S. Penzel, K. Szekely, R. Cadalbert, E. Testori, A. Oss, J. Past, A. Samoson, M. Ernst, A. Böckmann, B. H. Meier, Angew. Chem. Int. Ed. 2014, 53, 12253–12256; d) L. B. Andreas, K. Jaudzems, J. Stanek, D. Lalli, A. Bertarello, T. Le Marchand, D. Cala-De Paepe, S. Kotelovica, I. Akopjana, B. Knott, S. Wegner, F. Engelke, A. Lesage, L. Emsley, K. Tars, T. Herrmann, G. Pintacuda, Proc. Natl. Acad. Sci. USA 2016, 113, 9187–9192; e) J. Medeiros-Silva, D. Mance, M. Daniëls, S. Jekhmane, K. Houben, M. Baldus, M. Weingarth, Angew. Chem. Int. Ed. 2016, 55, 13606–13610; f) E. E. Najbauer, K. Tekwani Movellan, K. Giller, R. Benz, S. Becker, C. Griesinger, L. B. Andreas, J. Am. Chem. Soc. 2022, 144, 2953–2967.
- [7] K. Pfeiffer, V. Gohil, R. A. Stuart, C. Hunte, U. Brandt, M. L. Greenberg, H. Schägger, J. Biol. Chem. 2003, 278, 52873–52880.
- [8] A. Mühleip, S. E. McComas, A. Amunts, *eLife* **2019**, *8*.
- [9] J. B. McMillin, W. Dowhan, *Biochim. Biophys. Acta Mol. Cell Biol. Lipids* 2002, *1585*, 97–107.
 [10] a) S. Ghosh, W. Basu Ball, T. R. Madaris, S. Srikantan, M. Madesh, V. K.
- [10] a) S. Gnosn, W. Basu Bail, I. K. Madaris, S. Srikantan, M. Madesh, V. K. Mootha, V. M. Gohil, Proc. Natl. Acad. Sci. USA 2020, 117, 16383–16390;
 b) K. Malhotra, A. Modak, S. Nangia, T. H. Daman, U. Gunsel, V. L. Robinson, D. Mokranjac, E. R. May, N. N. Alder, Sci. Adv. 2017, 3.
- [11] E. G. Bligh, W. J. Dyer, Can. J. Biochem. Physiol. 1959, 37, 911–917.
- [12] M. H. Filgueiras, J. A. F. O. den Kamp, Biochim. Biophys. Acta, Lipids Lipid Metab. 1980, 620, 332–337.
- [13] R. Letters, *Biochim. Biophys. Acta, Lipids Lipid Metab.* 1966, *116*, 489–499.
 [14] M. Weingarth, G. Bodenhausen, P. Tekely, *Chem. Phys. Lett.* 2010, *488*,
- 10-16.
- [15] P. D'Arrigo, L. de Ferra, G. Pedrocchi-Fantoni, D. Scarcelli, S. Servi, A. Strini, J. Chem. Soc. Perkin Trans. 1 1996, 2657–2660.
- [16] K. Pluhackova, A. Horner, BMC Biol. 2021, 19.
- [17] a) C. Pliotas, A. C. E. Dahl, T. Rasmussen, K. R. Mahendran, T. K. Smith, P. Marius, J. Gault, T. Banda, A. Rasmussen, S. Miller, C. V. Robinson, H. Bayley, M. S. P. Sansom, I. R. Booth, J. H. Naismith, *Nat. Struct. Mol. Biol.* **2015**, *22*, 991–998; b) K. Gupta, J. A. C. Donlan, J. T. S. Hopper, P. Uzdavinys, M. Landreh, W. B. Struwe, D. Drew, A. J. Baldwin, P. J. Stansfeld, C. V. Robinson, *Nature* **2017**, *541*, 421–424.
- [18] W. M. O'Leary, S. G. Wilkinson, in *Microbial Lipids, Vol. 2* (Eds.: C. Ratledge, S. G. Wilkinson), Academic Press, **1988**, pp. 117–201.
- [19] M. Baldus, B. H. Meier, J. Magn. Reson. Ser. A 1996, 121, 65-69.
- [20] G. Hedger, M. S. P. Sansom, Biochim. Biophys. Acta Biomembr. 2016, 1858, 2390–2400.
- [21] I. Kotsogianni, T. M. Wood, F. M. Alexander, S. A. Cochrane, N. I. Martin, ACS Infect. Dis. 2021, 7, 2612–2619.
 [22] C. M. L. L. L. C. L. District Control of Control o
- [22] G. Machaidze, J. Seelig, *Biochemistry* 2003, 42, 12570–12576.
- [23] F. Hullin-Matsuda, A. Makino, M. Murate, T. Kobayashi, Biochimie 2016, 130, 81–90.
- [24] N. Naruse, O. Tenmyo, K. Tomita, M. Konishi, T. Miyaki, H. Kawaguchi, K. Fukase, T. Wakamiya, T. Shiba, J. Antibiot. 1989, 42, 837–845.
- [25] K. Hosoda, M. Ohya, T. Kohno, T. Maeda, S. Endo, K. Wakamatsu, Journal of Biochemistry 1996, 119, 226–230.
- [26] M. Vestergaard, N. A. Berglund, P.-C. Hsu, C. Song, H. Koldsø, B. Schiøtt, M. S. P. Sansom, ACS Omega 2019, 4, 18889–18899.
- [27] S.-Y. Choung, T. Kobayashi, J.-i. Inoue, K. Takemoto, H. Ishitsuka, K. Inoue, Biochim. Biophys. Acta Biomembr. 1988, 940, 171–179.
- [28] K. Wakamatsu, S. Y. Choung, T. Kobayashi, K. Inoue, T. Higashijima, T. Miyazawa, *Biochemistry* 1990, 29, 113–118.
- [29] a) J. Medeiros-Silva, S. Jekhmane, A. L. Paioni, K. Gawarecka, M. Baldus, E. Swiezewska, E. Breukink, M. Weingarth, *Nat. Commun.* 2018, 9; b) R. Shukla, J. Medeiros-Silva, A. Parmar, B. J. A. Vermeulen, S. Das, A. L. Paioni, S. Jekhmane, J. Lorent, A. M. J. J. Bonvin, M. Baldus, M. Lelli, E. J. A. Veldhuizen, E. Breukink, I. Singh, M. Weingarth, *Nat. Commun.* 2020, 11; c) R. Shukla, F. Lavore, S. Maity, M. G. N. Derks, C. R. Jones, B. J. A. Vermeulen, A. Melcrová, M. A. Morris, L. M. Becker, X. Wang, R. Kumar, J. Medeiros-Silva, R. A. M. van Beekveld, A. M. J. J. Bonvin, J. H. Lorent, M. Lelli, J. S. Nowick, H. D. MacGillavry, A. J. Peoples, A. L. Spoering, L. L. Ling, D. E. Hughes, W. H. Roos, E. Breukink, K. Lewis, M. Weingarth, *Nature* 2022, 608, 390–396.
- [30] a) J. R. Lewandowski, H. J. Sass, S. Grzesiek, M. Blackledge, L. Emsley, J. Am. Chem. Soc. 2011, 133, 16762–16765; b) S. Jekhmane, J. Medeiros-Silva, J. Li, F. Kümmerer, C. Müller-Hermes, M. Baldus, B. Roux, M. Weingarth, Nat. Commun. 2019, 10.
- [31] D. E. Warschawski, A. A. Arnold, I. Marcotte, *Biophys. J.* 2018, 114, 1368– 1376.

Chem. Eur. J. 2022, 28, e202202472 (7 of 8)

© 2022 The Authors. Chemistry - A European Journal published by Wiley-VCH GmbH



5213765,

- [32] a) S. Laage, Y. Tao, A. E. McDermott, Biochim. Biophys. Acta Biomembr. 2015, 1848, 260-265; b) M. Weingarth, A. Prokofyev, E. A. W. van der Cruijsen, D. Nand, A. M. J. J. Bonvin, O. Pongs, M. Baldus, J. Am. Chem. Soc. 2013, 135, 3983-3988; c) M. E. Ward, E. Ritz, M. A. M. Ahmed, V. V. Bamm, G. Harauz, L. S. Brown, V. Ladizhansky, J. Biomol. NMR 2015, 63, 375-388.
- [33] a) A. Legrand, D. Martinez, A. Grélard, M. Berbon, E. Morvan, A. Tawani, A. Loquet, S. Mongrand, B. Habenstein, Front. Mol. Biosci. 2019, 6; b) J. J. Kinnun, K. J. Mallikarjunaiah, H. I. Petrache, M. F. Brown, Biochim. Biophys. Acta - Biomembranes 2015, 1848, 246-259; c) R. Dazzoni, A. Grélard, E. Morvan, A. Bouter, C. J. Applebee, A. Loquet, B. Larijani, E. J. Dufourc, Sci. Rep. 2020, 10.
- [34] A. Loquet, K. Giller, S. Becker, A. Lange, J. Am. Chem. Soc. 2010, 132, 15164-15166.
- [35] a) C. G. Borcik, I. R. Eason, M. Yekefallah, R. Amani, R. Han, B. H. Vanderloop, B. J. Wylie, Angew. Chem. Int. Ed. 2022, 61; b) M. R. Elkins, I.V. Sergeyev, M. Hong, J. Am. Chem. Soc. 2018, 140, 15437-15449; c) M. R. Elkins, J. K. Williams, M. D. Gelenter, P. Dai, B. Kwon, I. V. Sergeyev, B. L. Pentelute, M. Hong, Proc. Natl. Acad. Sci. USA 2017, 114, 12946-12951; d) T. M. Anderson, M. C. Clay, A. G. Cioffi, K. A. Diaz, G. S.

Hisao, M. D. Tuttle, A. J. Nieuwkoop, G. Comellas, N. Maryum, S. Wang, B. E. Uno, E. L. Wildeman, T. Gonen, C. M. Rienstra, M. D. Burke, Nat. Chem. Biol. 2014, 10, 400-406.

- [36] J. C. Dittmer, R. L. Lester, J. Lipid Res. 1964, 5, 126-127.
- [37] E. L. Ulrich, H. Akutsu, J. F. Doreleijers, Y. Harano, Y. E. Ioannidis, J. Lin, M. Livny, S. Mading, D. Maziuk, Z. Miller, E. Nakatani, C. F. Schulte, D. E. Tolmie, R. Kent Wenger, H. Yao, J. L. Markley, Nucleic Acids Res. 2008, 36, D402-408.
- [38] a) G. Knothe, Lipids 2006, 41, 393–396; b) F. D. Gunstone, Chem. Phys. Lipids 1993, 65, 155-163.
- [39] B. M. Fung, A. K. Khitrin, K. Ermolaev, J. Magn. Reson. 2000, 142, 97-101. [40] M. Weingarth, G. Bodenhausen, P. Tekely, J. Magn. Reson. 2009, 199, 238-241.
- [41] K. R. Thurber, R. Tycko, J. Magn. Reson. 2009, 196, 84-87.

Manuscript received: August 9, 2022 Accepted manuscript online: September 13, 2022 Version of record online: October 25, 2022