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Warming stimulates sediment denitrification at the expense of anaerobic ammonium oxidation

Ehui Tan[®]¹, Wenbin Zou¹, Zhenzhen Zheng¹, Xiuli Yan², Moge Du¹, Ting-Chang Hsu³, Li Tian¹, Jack J. Middelburg⁴, Thomas W. Trull^{®⁵} and Shuh-ji Kao[®]^{1⊠}

Temperature is one of the fundamental environmental variables governing microbially mediated denitrification and anaerobic ammonium oxidation (anammox) in sediments. The GHG nitrous oxide (N_2O) is produced during denitrification, but not by anammox, and knowledge of how these pathways respond to global warming remains limited. Here, we show that warming directly stimulates denitrification-derived N_2O production and that the warming response for N_2O production is slightly higher than the response for denitrification in subtropical sediments. Moreover, denitrification had a higher optimal temperature than anammox. Integrating our data into a global compilation indicates that denitrifiers are more thermotolerant, whereas anammox bacteria are relatively psychrotolerant. Crucially, recent summer temperatures in low-latitude sediments have exceeded the optimal temperature of anammox, implying that further warming may suppress anammox and direct more of the nitrogen flow towards denitrification and associated N_2O production, leading to a positive climate feedback at low latitudes.

he anthropogenic production of reactive nitrogen (Nr) has increased tenfold over the past century and is expected to continue to increase to keep pace with the growing world population¹. The impact of anthropogenic Nr on ecosystems and biogeochemical cycles has become the second most serious environmental issue at the global scale^{2–4}. In parallel, climate change is predicted to be a major cause of species extinction and biodiversity degradation in the coming century^{5,6}, and species that occupy narrow thermal niches are likely to be most vulnerable to global warming⁷. Since almost all nitrogen-associated processes are microbially driven, rising temperature will inevitably promote or suppress specific microbial activities^{8,9}, and thus affect global biogeochemical cycles with climate feedbacks¹⁰.

The land-ocean interface receives excess Nr and serves as a hotspot for Nr removal. This occurs through denitrification in anoxic environments¹¹ and the relatively newly discovered anaerobic ammonium oxidation (anammox) process¹², both of which reduce Nr to biologically unavailable dinitrogen (N₂). Both denitrification and anammox are widely observed in anoxic habitats such as paddy soils¹³, salt marshes¹⁴, mud flats¹⁵ and aquatic sediments¹⁶. In contrast to denitrification, anammox does not involve release of N_2O , which is about 300 times more potent than CO_2 in terms of greenhouse warming potential. Thus, anammox is recognized as a more climate-friendly Nr-removal pathway relative to denitrification. These two processes and their partitioning largely determine the size of the global Nr sink and N₂O source. Previous studies have shown that anammox could be crucial in some deepsea sediments^{17,18}, although Nr removal by benthic denitrification generally exceeds removal by anammox¹⁹⁻²¹ in such environments. Devol concluded that the fraction of total N2 production by anammox (ra%) ranged from 0-79% in marine sediments and typically constitutes approximately 10-40% of the total Nr removal¹⁶. The variance in ra% has been attributed to a number of environmental factors such as temperature, the quantity and quality of organic matter, and nitrate concentration^{19,22-24}. Among these potential environmental variables, temperature is one of the most fundamental factors governing microbial metabolism of denitrification and anammox, but the temperature effect has not been sufficiently resolved. Global ra% and temperature data²⁵ show that ra% has an inverse relationship with the temperature at which samples were collected or incubated. Despite decades of study, the factors that control the partitioning between sedimentary denitrification and anammox remain underexplored¹⁶, specifically, the knowledge of how the two Nr-removal processes respond to future warming remains limited due to lack of field measurements^{19,26-29}. Understanding the environmental factors of Nr-removal pathways and associated N₂O production, particularly temperature, will be essential for Earth system models¹¹ to accurately quantify the Nr-removal flux and to predict future biogeochemical cycles and associated climate feedbacks driven by reactive nitrogen.

A few studies have examined the temperature dependence of sedimentary denitrification and anammox in aquatic ecosystems, mostly focused on sediments in continental shelves and estuaries in temperate or polar regions^{19,26-29}, but there is a lack of reports from subtropical regions, especially for sediments deposited in shallow freshwater creeks, estuaries, tidal flats, mangroves and aquaculture ponds. These coastal systems receive anthropogenic Nr and will directly experience future temperature rise. In this study, we applied the ¹⁵N-isotope pairing technique to sediments collected from various coastal aquatic systems in southern China, a typical subtropical region (Supplementary Fig. 1), to identify how temperature change may affect the potential rates of Nr-removal pathways, the denitrification-driven N₂O production rate, and the partitioning between denitrification and anammox in subtropical sediments. By adding our experimental data for subtropical regions to existing data for various ecosystems, we evaluate the latitudinal differences in thermal sensitivity of microbial communities involved in Nr removal to global warming in relation to N₂ and N₂O production rates at an ecosystem level.

¹State Key Laboratory of Marine Environmental Science, College of Ocean and Earth Science, Xiamen University, Xiamen, China. ²Institute of Marine Science and Guangdong Provincial Key Laboratory of Marine Biotechnology, Shantou University, Shantou, China. ³School of Urban and Environmental Science, Huaiyin Normal University, Huaian, China. ⁴Department of Earth Sciences, Utrecht University, Utrecht, The Netherlands. ⁵Commonwealth Scientific and Industrial Research Organization, Hobart, Tasmania, Australia. ⁵Me-mail: sjkao@xmu.edu.cn



Fig. 1 [Temperature responses of sedimentary denitrification, anammox and related N₂O production potential rates. a–I, Temperature manipulation experiments for samples collected during summer (**a–f**) and winter (**g–I**) at different sampling sites. Sampling points are indicated in the top left of each graph. Error bars denote the s.d. of triplicates (most are smaller than the symbols).

Temperature response of Nr-removal processes and $N_2 O \mbox{ production}$

We observed consistently distinct temperature response patterns of denitrification and anammox (Fig. 1). From these thermal performance patterns, we obtained different optimum temperature (T_{opt}) values for both denitrification and anammox for various habitats and seasons. In the brackish-water ecosystems (estuary, aquaculture pond and tidal flat; see Supplementary Table 1), denitrification potential continuously increased with increasing incubation temperature up to 35 °C in both seasons (Fig. 1a–e,g–k), except for site S4 in winter, where T_{opt} for denitrification was approximately 30 °C (Fig. 1j). This suggests that the T_{opt} for denitrification in our

study system should be higher than 35 °C in summer and higher than 30 °C in winter. T_{opt} values for anammox were determined to be 25–30 °C in summer and 20–30 °C in winter (Fig. 1). In the freshwater ecosystem (site S6 in both seasons), T_{opt} values for denitrification and anammox were found to be 30 °C in summer (Fig. 1f), whereas in winter, the T_{opt} values for denitrification and anammox were 25 °C and 20 °C, respectively (Fig. 1l). Overall, T_{opt} for denitrification was consistently higher than for the corresponding anammox in both seasons (*P* values in Supplementary Table 3); leaving aside the seasonal variation in T_{opt} for denitrification due to the uncertainty, the T_{opt} for anammox was significantly higher in summer than in winter (*P*=0.017; Supplementary Table 4). In addition, the T_{opt} values



Fig. 2 | Temperature responses of the relative contribution of anammox to total gas ($N_2 + N_2O$) production (ra%). a,b, Temperature responses at different sampling sites in summer (a) and winter (b). Error bars represent the s.d. of triplicates.

for both Nr-removal processes in brackish-water ecosystems were higher than those in freshwater ecosystems regardless of season. Remarkably, the habitat temperature was higher than the observed T_{opt} for the anammox community during summer in the subtropical regions (Fig. 1a–f).

Sedimentary production of N_2O , an intermediate during denitrification, was observed to increase with warming in both summer and winter samples. The response of N_2O production to temperature change showed a consistent pattern, with potential rates increasing exponentially as the incubation temperature increased, regardless of sites and seasons (Fig. 1).

Temperature effects on relative contribution of anammox

The relative contribution of anammox to total Nr removal (ra%) showed a large spatial variation among stations, with increasing contributions at lower incubation temperature for all except two cases (S1 in summer and S4 in winter; Fig. 2). These two exceptions exhibited relatively high ra% at moderate temperatures, with low ra% at both high and low temperature extremes (Fig. 2). Additionally, four ra% data points in winter exceeded 40% due to the extremely low ${}^{30}N_2$ production at low temperature (see gas production rates of S1 and S2 in winter in Supplementary Fig. 8), thus leading to an exceptionally high contribution of anammox to total Nr removal (Fig. 2b). The average ra% at the habitat temperature across our study sites was approximately 7% in summer and 5% in winter.

Q₁₀ values for Nr-removal processes and N₂O production

To investigate whether the warming effect on the intermediate product (N_2O) and the end product (N_2) is invariable during denitrification, the Q_{10} values for N₂ and N₂O production were calculated separately (Supplementary Figs. 2 and 3). The Q_{10} of denitrification-related N₂ production was in the range 1.9–2.7 (2.2 ± 0.3 , mean \pm s.d.) in summer and 2.0–3.5 (2.6 \pm 0.6) in winter (Fig. 3); the denitrification-derived N_2O production had slightly higher Q_{10} in both summer (2.9 ± 0.9) and winter (3.1 ± 0.6) , with exclusion of two extreme measurements at sites S4 and S6). The higher Q_{10} values for the denitrification-derived N2O production imply that warming might direct more nitrogen flow towards N₂O during nitrate removal. For denitrification as a whole, the Q_{10} was in the range $1.9-2.7~(2.3\pm0.4)$ and $2.2-3.5~(2.9\pm0.5)$ in summer and winter, respectively (Fig. 3). For anammox, the Q_{10} was in the range 1.3–2.5 (1.9 ± 0.5) in summer and 2.1-4.8 (2.4 ± 0.2) in winter (excluding site S4 in winter) (Fig. 3). No significant spatial and seasonal differences were observed for Q₁₀ values of denitrification-derived N₂O and N₂ production (see P values in Supplementary Tables 5 and 6).

Combining the Q_{10} values from both seasons (Fig. 3), the Q_{10} values for denitrification-derived N₂O were slightly higher than those for denitrification-derived N₂ (*P*=0.042; Supplementary Table 6); moreover, the Q_{10} values for denitrification were consistently higher than those for anammox (*P*=0.021; Supplementary Table 7).

The temperature dependence of denitrification-associated N₂O production in aquatic sediments is rarely reported. Our results in subtropical coastal sediments show that warming directly stimulates N₂O production by increasing sedimentary denitrification (Fig. 1). There are very few estimates in coastal sediments available for comparison of the Q₁₀ values of sedimentary N₂O production. Our results (1.2-7.1) in subtropical aquatic sediments were within the range of the few available studies, which reported a Q_{10} of 1.8 in lake sediments³⁰ and 3.7–12 in grassland or soils^{31,32}. For the Q_{10} of denitrification, our estimates (1.9-3.5) were slightly lower than two estimates (3.8 and 5.0) for the subtropical region, but were in line with the measurements in a few studies from temperate (1.7-2.5)and polar (1.6–2.3) regions (Fig. 4a). Q_{10} values for anammox in our study (1.3-4.9) were similar to those from the limited anammox studies from temperate (1.6-3.1) and polar (1.4-2.7) regions (Supplementary Table 8). Accordingly, the Q_{10} values of denitrification-derived N2O production, denitrification and anammox in this study are comparable with values reported in the literature, supporting the suggestion that our results may be applicable to natural environments more broadly at a global scale, even though the slurry incubation methodology provides potential rates. Both denitrification and anammox are responsive to temperature change in aquatic ecosystems. Although other site-specific and season-specific factors, such as content of organic compounds, substrate concentration and community structure, may have roles in the variability of Q_{10} , the results in Fig. 3 indicate statistically different warming response patterns among denitrification-derived N2O production, denitrification and anammox. This coherent pattern implies that warming may favour denitrification more than anammox, and consequently, direct more of the nitrogen flow towards denitrification, in which the nitrogen bifurcation to N₂O is even higher, leading to a positive climate feedback.

To obtain a more comprehensive picture of the temperature response of Nr-removal pathways at the global scale, we assembled a cross-system comparison among studies from other regions, including pure culture bacteria³³, polar continental shelf^{28,29}, polar intertidal flats²⁷, polar fjords²², temperate inner continental shelf²⁶, temperate intertidal flats²⁷, temperate estuary²⁶, subtropical tidal zones²⁷, rivers³⁴, streams³⁵, anoxic groundwater aquifer³⁶, riparian soil³⁷, mountain lakes³⁸ and denitrification bed³⁹. Both denitrification and anammox showed wide ranges of T_{out} values (Fig. 4b and



Fig. 3 | Variation of Q_{10} **values for denitrification-related gas production and Nr-removal processes from both seasons.** Q_{10} values for denitrificationrelated gas production and Nr-removal processes for both seasons are shown. D_{N_20} , potential rate of denitrification-derived N₂O production; D_{N_27} , potential rate of denitrification-derived N₂ production; D, potential rate of denitrification; A, potential rate of anammox. The mid lines and triangles in the boxes indicate the median and the mean of Q_{10} values, respectively, from two seasons; the lower and upper lines of the boxes are the first and third quartiles; the lower and upper whiskers represent the minimum and maximum values, respectively. The black crosses and circles denote the data from summer and winter, respectively. Statistical significance is tested between D_{N_20} and D_{N_2} , and between D and A; *P < 0.05. See specific Q_{10} values for different processes in Supplementary Figs. 2 and 3.

Supplementary Table 8). Combined with our results from the subtropical region, the enriched database reveals how community-level Nr-removal processes vary with temperature change latitudinally (Fig. 4b). The high T_{opt} values of more than 35 °C in summer and more than 30°C in winter for denitrification found in our brackish-water ecosystems were consistent with those from a subtropical tidal zone in the Gulf of Mexico²², where denitrification T_{opt} values were observed to be 35-37 °C (sites 4 and 8 in Fig. 4b). Our T_{opt} values were significantly higher than those found in the temperate and polar sediments (21.0-31.1 °C; Fig. 4b) except for two sites reported by Canion et al.²⁷ and Brin et al.²⁶ (sites 16 and 24 in Fig. 4b), which had comparable $T_{\rm opt}$ to those in our study. For anammox, the observed T_{opt} values (20–30 °C in both seasons) in our study were similar to the T_{opt} of 22–33 °C found in temperate sediments (Fig. 4b) and to the T_{opt} of 25–30 °C in a culture of marine anammox bacteria (Supplementary Table 8). However, our T_{opt} were higher than the anammox optimum of 12°C and 15°C found in continental shelf sediments from polar regions by Dalsgaard and Thamdrup²⁹ and Rysgaard et al.²⁸, respectively, where habitat temperatures were markedly lower than those in our study (sites 29 and 34 in Fig. 4b). Relative to the anammox optimum of 37 °C found in wastewater treatment systems^{40,41}, almost all measurements from field observations were much lower.

 $T_{\rm opt}$ has been widely used to study the temperature tolerance of microbes in laboratory studies^{12,42}. In Fig. 4b, the denitrification $T_{\rm opt}$ values were generally higher than 25 °C, which agrees well with the mesophilic response to temperature for denitrifying communities, as reported in most laboratory studies^{22,43}. Additionally, we find that except for the three data points in ref. ²⁶, $T_{\rm opt}$ values for denitrification were consistently higher than those for anammox regardless of season or region (Fig. 4b), suggesting that denitrifiers have a higher thermal tolerance than anammox bacteria. Notably, $T_{\rm opt}$ values increase from polar to low-latitude regions for both denitrification and anammox, corresponding to the latitudinal pattern of habitat temperature (Fig. 4b). In addition, the temperature differences between $T_{\rm opt}$ and habitat temperature significantly decrease towards low latitudes for both Nr-removal processes. Moreover,

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the projected temperature after warming by 2100 in the representative concentration pathway (RCP) emission scenario 8.5 is close to the denitrification T_{opt} and even exceeds the upper limit of anammox T_{opt} in the subtropical summer (Fig. 4b). Such a pattern suggests that habitat temperature will remain in a tolerable range for denitrification and anammox in mid-high latitudes, whereas in low latitudes, a small temperature increase may have large effects on the growth of anammox bacteria, although low latitudes in the Northern Hemisphere are likely to experience less warming in the future relative to the mid-high latitudes⁴⁴, and thus result in a shift in the community activity of Nr-removal microbes. On the basis of our observations, we hypothesize that anammox at low latitudes is vulnerable to global warming, with the consequence that the proportional contribution of anammox to total Nr removal declines. Considering the ability of microbes to adapt to different temperatures and the large timescale at which climate change occurs, however, anammox communities may shift in composition towards microbial species that are more adapted to warming temperatures.

In this study, we address the direct impact of temperature on sedimentary Nr-removal processes and associated N2O production. However, environmental regulators often co-vary across global ecosystems, and the variation in T_{opt} values for denitrification and anammox might be due to other environmental factors, such as microbial assemblage, organic carbon availability or nitrate and oxygen concentration^{27,38,45-47}. The latitudinal variation pattern of $T_{\rm opt}$ in the two tested pathways (Fig. 4b) indicates that habitat temperature has a significant role in regulating thermal dependence, even though the T_{opt} for anammox may respond in a different way to that for denitrification in relation to habitat temperature due to varying thermal tolerance between these two microbial communities. In our case, T_{opt} values for denitrification and anammox in the freshwater ecosystem were lower than those in brackish-water ecosystems in both seasons (Fig. 1); such a difference might be caused by the different compositions of microbial assemblages^{45,47}. Future studies combining isotopic and molecular techniques are needed to clarify the relative influence of physicochemical and biological factors²⁷. Additionally, substrate availability is a known regulator of the temperature performance of denitrifiers^{27,38,46}. Substratedependent temperature tolerance can also be found in freshwater phytoplankton species48. In general, warming tends to stimulate organic carbon- and nitrogen-transformation processes⁴⁹. GeoChip hybridization-based analysis revealed that the abundance of genes involved in organic carbon decomposition and denitrification increased significantly in response to warming^{50,51}. Warming can facilitate the degradation of recalcitrant carbon to labile and dissolved organic carbon, and this additional supply of electron donors can subsequently stimulate heterotrophic denitrifiers^{50,52}. However, the specific effect of warming-derived carbon decomposition on denitrification and anammox requires further investigation. The studies mentioned here indicate that various environmental factors may synergistically influence the temperature response of microbial species; nevertheless, our analysis clearly shows that denitrification and anammox communities respond sensitively but dissimilarly to warming at the ecosystem level.

According to the fifth assessment report of the IPCC, the projected warming at the global scale by the end of the 21st century relative to 1986–2005 is likely to be between 0.3 °C and 1.7 °C and between 2.6 °C to 4.8 °C in scenarios of low GHG emissions (RCP 2.6) and high GHG emissions (RCP 8.5) scenarios, respectively⁴⁴. Intertidal sediments are expected to directly experience such warming due to the increasing air and water temperatures; thus, the sedimentary denitrification and associated N₂O production potential may be intensified with increasing temperature, as shown by the immediate response of sedimentary Nr removal and N₂O production in our study. Encouraged by the globally consistent pattern of experimentally derived Q_{10} for denitrification (Fig. 4a), and given



Fig. 4 | Compiled literature values for denitrification and anammox in subtropical, temperate and polar sediments globally. a,b, The Q_{10} values for denitrification (**a**) and habitat temperatures, projected temperature range and T_{opt} values for denitrification and anammox (**b**) are shown at different latitudes. The projected temperature range is the sum of habitat temperature and projected warming range by 2100 in RCP scenario 8.5. The numbers indicate sites and latitudes from different studies, as shown in Supplementary Table 8. Note that the error bars for T_{opt} in our study were taken to be ± 2.5 °C and show only positive error bars for denitrification when T_{opt} is higher than 35 °C. The specific values are listed in Supplementary Table 8.



Fig. 5 | The projected increase of sedimentary denitrification and associated N_2O production in coastal environments. a,b, The increase of denitrification (a) and N_2O production (b), as caused by the potential warming across latitudes under RCP 2.6 and RCP 8.5 scenarios. In both plots, the lines show the mean and the shaded areas show the s.d.

the lack of estimates of Q_{10} for coastal sedimentary N₂O production in mid-high latitudes, we applied our results from low latitudes to predict future global Nr removal and associated N₂O production under warming climate conditions, assuming that all environmental parameters other than temperature are invariable. However, these predictions are subject to large uncertainties because of the underlying assumptions and the differential effect of warming on multiple factors governing carbon and nitrogen cycling in marine systems. Considering only the changing temperature and applying the average Q_{10} value of 2.6±0.5 for denitrification and 2.9±0.8 for N₂O production, the projected increase suggests that by 2100, sedimentary Nr removal in marine environments will increase significantly by up to 18% under the RCP 2.6 scenario and 116% under the RCP 8.5 scenario, whereas N₂O production will increase by up to 21% under the RCP 2.6 scenario and 134% under the RCP 8.5 scenario, with the highest increases in the northern high-latitude ecosystems

(Fig. 5). We refrain from presenting a global projection for anammox, because such an approach may not apply because of its lower thermo-tolerance (that is, conditions are likely to exceed T_{opt} values in low-latitude regions). Thus, we believe that while future warming may promote both Nr-removal processes at mid-high latitudes, particularly in the Northern Hemisphere, at low latitudes, warming may suppress anammox (the more climate-friendly community involving no GHG production), with the overall result that denitrification and N₂O production directly stimulated by warming becomes of higher relative importance, thus leading to enhanced warming feedback to the global climate system.

Online content

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Methods

Seasonal sampling and pretreatment. The study area is in the Fujian Province in southern China (Supplementary Fig. 1), which experiences a typical subtropical climate. The monthly average air temperature ranges from 10°C to 32°C, while the highest temperature can reach 37°C. The average annual rainfall is ~1,200 mm⁵³, of which ~75% occurs in the wet season from April to October⁴⁴. The excessive use of fertilizers in the Jiulong River watershed has led to intensive Nr pollution in downstream regions; Chen et al.⁵⁵ quantified the annual Nr input to the Jiulong River watershed to be ~129 kg Nha⁻¹ yr⁻¹, resulting in high ammonium and nitrate concentrations in the downstream estuary⁻⁶. Our samples were collected from diverse habitats in Jiulong River estuary and its nearby region, where nitrogen is highly saturated (Supplementary Fig. 1), including the brackish-water ecosystems (estuary, aquaculture pond and tidal flat) and the freshwater ecosystem (freshwater creek).

We conducted campaigns in both summer (20 July 2018) and winter (26 November 2017 and 18 December 2017). Six sites were sampled in each campaign (Supplementary Fig. 1), including estuarine regions, tidal flats, an aquaculture pond and a freshwater creek (Supplementary Fig. 1 and Supplementary Table 1). Both surface sediments and its overlying water were collected. The surface 2 cm of sediments were collected using a grab sampler and immediately transferred into a hermetic bag, whereas the overlying water was collected by using a 21 brown bottle. During sampling, the temperature of sediments and the salinity in overlying water were measured on-site using a thermometer and WTW analyser, respectively. After collection, the sediments and water samples were stored in an ice chest in the field and transported back to laboratory within 4 h for chemical analyses and incubation experiments.

In the laboratory, the overlying water was filtered using a 0.22 μm membrane, and the filtrate was then prepared for later temperature-gradient incubation experiments and nutrients analysis, including ammonium (NH₄⁺) and nitrite plus nitrate (NO_x⁻). Meanwhile, the sediments collected from each site were divided into two parts; one part was stored at -20 °C for later analysis of physiochemical characteristics of sediments, including the porosity and contents of organic carbon (OC) and organic nitrogen (ON). The remaining sediments were applied to conduct temperature-gradient incubation experiments.

Temperature-gradient incubation assays. The temperature dependence of denitrification and anammox were investigated by applying temperature-gradient incubation experiments. The experiments were conducted in thermostat incubators with a temperature control accuracy of ± 1 °C. Temperature was set between 1 and 35 °C with 5 °C intervals.

Potential rates of denitrification and anammox were measured using slurry incubations, a simplified version of the technique described by Thamdrup and Dalsgaard⁵⁷. In brief, the surface sediments were mixed and vigorously stirred with the filtered water in a ratio of 1:4 (v/v) to make homogenized slurries. The slurries were then purged with helium to ensure anoxic conditions. Dissolved oxygen concentration in slurries was monitored by using an oxygen sensor (OX 50, Unisense) with continuous helium flushing until the oxygen was undetectable⁵⁸. Subsequently, 3 ml of anoxic slurries were transferred into gas-tight borosilicate vials (12 ml, Labco Exetainer), and the vials were then sealed with butyl rubber septa and screw caps. The headspace in the vials was again purged with helium. A total of 15 vials for each temperature treatment were prepared at each sampling site. After the preparation, all the vials were pre-incubated for ~24 h at room temperature to eliminate residual oxygen and background NO_x⁻. To acclimate the microbes to the designed temperature, another 24 h pre-incubation was performed at this temperature before ¹⁵N-tracer addition.

After complete pre-incubations, three of the 15 vials in each experimental set were used for residual NO_x⁻ concentration analysis. In the remaining 12 vials, the slurries were spiked with ¹⁵NO₃⁻ (Sigma-Aldrich, 98% ¹⁵N atoms) to a final concentration of 100 µmol ¹⁵N l⁻¹, and were then thoroughly mixed. In each temperature set, 3 of the vials were immediately fixed with 100 µl of saturated ZnCl₂ solution after the ¹⁵N-tracer addition and assigned as initial zero-time samples. The remaining nine vials were then incubated at the designed temperature in the dark. During the incubation, the samples were collected after about 0.5, 1 and 2 h. Three replicates were sacrificed for each time interval by adding 100 µl of saturated ZnCl₂ solution. After the incubation, all of the fixed samples were kept upside down at room temperature in the dark before ¹⁵N-labelled N₂ and N₂O measurement. Detailed time-course results for summer and winter can be seen in Supplementary Figs. 6 and 7, respectively, which show the data quality of our rate measurements.

 $T_{\rm opt}$ values for denitrification and an ammox were estimated from the thermal sensitivity curves, and assigned an uncertainty of $2.5\,^{\circ}{\rm C}$ to reflect the 5 $^{\circ}{\rm C}$ spacing of the incubation temperatures. Denitrification rates did not always show a clear down turn with increasing temperature, and for those experiments, $T_{\rm opt}$ values were set to the top incubation value of $35\,^{\circ}{\rm C}$ and are thus minimum estimates. This means the difference in thermal sensitivity of the two processes may be even larger than shown here.

Chemical analysis. Sediment porosity was measured from the weight loss of a known amount of wet sediments dried at $60\,^{\circ}$ C to a constant value. The dry

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sediments were then acidified with 1 N HCl to remove inorganic carbon and freeze dried, and the OC and ON contents in sediments were determined with a FLASH 2000 elemental analyzer⁵⁹. The concentration of NO_x^- was analysed by vanadium (III) reduction with chemiluminescence detector⁶⁰. The NH₄⁺ concentration was measured by the indophenol blue spectrophotometric method⁶¹ with a detection limit of 0.5 µmoll⁻¹. The environmental characteristics of sampling sites are provided in the Supplementary Information. The ¹⁵N-labelled gas products in fixed samples, including ²⁹N₂, ³⁰N₂, ⁴⁵N₂O and ⁴⁶N₂O, were quantified by using a Thermo Finnigan Delta V Plus isotope-ratio mass spectrometer equipped with a Gas-Bench II and a PoraPlot Q gas chromatography column (see details in ref. ⁶²).

Rate calculation and temperature-sensitivity evaluation. We applied the quantification technique as described⁶² for potential rate calculation. The potential rates of denitrification, anammox and associated N₂O production were quantified according to the ${}^{29}N_2$, ${}^{30}N_2$, ${}^{45}N_2O$ and ${}^{46}N_2O$ production rates by using the following equations:

$$D_{N_2O} = (P_{45} + 2P_{46}) \times (r_{14} + 1) = 2P_{46} \times (r_{14} + 1)^2$$
(1)

$$D = 2 \times P_{30} \times (r_{14} + 1)^2 + (P_{45} + 2P_{46}) \times (r_{14} + 1)$$
(2)

$$A = (P_{29} - 2 \times r_{14} \times P_{30}) \times (2 \times r_{14} + 1)$$
(3)

$$D_{N_2} = D - D_{N_2O}$$
 (4)

where D_{N_2O} , D, A and D_{N_2} represent the potential rates of N_2O production, denitrification, anammox and denitrification-derived N_2 , respectively. P_{29} , P_{30} , P_{45} and P_{46} denote the ²⁹ N_2 , ³⁰ N_2 , ⁴⁵ N_2O and ⁴⁶ N_2O production rates, respectively, which were determined from the time-series incubation experiments (see Supplementary Figs. 7 and 8); and r_{14} is the ratio between ¹⁴ NO_3^- and ¹⁵ NO_3^- undergoing nitrate reduction, which is derived from the following equation:

$$r_{14} = \frac{C_{14} + 0.02 \times C_{15}}{0.98 \times C_{15}} \tag{5}$$

where C_{14} is the residual NO_x⁻ concentration after the secondary pre-incubation. C_{15} is the added ¹⁵N-labelled tracer during the incubation, here referred to ¹⁵NO₃⁻ (100 µmoll⁻¹). The numbers 0.02 and 0.98 represent the proportion of ¹⁴N and ¹⁵N in the added ¹⁵NO₄⁻, respectively.

The Q_{10} value, which reflects the biological rate in response to a temperature increase of 10 °C, was calculated to evaluate the temperature sensitivity of denitrification, anammox and N₂O production⁶¹. The relation between biogeochemical rates and temperature (below the optimum temperature) was simply expressed as an exponential function:

$$R(T) = R(T_0) \times e^{kT} \tag{6}$$

where *k* is the temperature coefficient. R(T) and $R(t_0)$ represent the denitrification, anammox or N₂O production rate measured at temperature *T* and *T*₀, respectively. Q_{10} was then calculated using the following equation, and the standard error in Q_{10} was estimated from the regression line:

$$Q_{10} = e^{10k}$$
 (7)

In addition, because the temperature sensitivity of microbes can be described in terms of activation energy (E_a), and many related studies have used this approach, we provide E_a results and discussions in the Supplementary Information. The responses of biological denitrification and anammox activity to temperature change were modelled using the integrated form of the Arrhenius equation^{63,64} to evaluate the thermal sensitivity of Nr-removal microorganisms.

$$\ln[R(T)] = -E'_{a}\left(\frac{1}{kT} - \frac{1}{kT_{c}}\right) + \ln[R(T_{c})]$$
(8)

$$E_a = E'_a \times 96.485 \tag{9}$$

where *T* is the absolute temperature (below optimum temperature, unit in Kelvin) and T_c is the standardized temperature (here $T_c = 15$ °C = 288.15 K, adopted from Perkins et al.^(a)). R(T) and $R(T_c)$ represent the potential rates of denitrification or anammox at temperature *T* and T_c , respectively; *k* is the Boltzmann constant $(8.62 \times 10^{-5} \text{ eV K}^{-1})$ and $\ln[R(T_c)]$ is the Arrhenius constant in the traditional equation. The E'_a and E_a represent the apparent activation energy with units of eV and KJ mol⁻¹, respectively. The value of 96.485 is a constant since 1 eV is equal to 96.485 KJ mol⁻¹. The standard error in E_a was estimated from the regression line.

Prediction of sedimentary denitrification and N₂O production increase. Based on the Q_{10} values for sedimentary denitrification and the associated N₂O production, we modelled the increase of Nr removal and N₂O production (ΔR)

as a result of global warming by the end of the 21st century according to the following equation,

$$\Delta R(\%) = 100 \times \frac{\Delta T \times Q_{10}}{10} \tag{10}$$

where ΔT is the projected warming (as reported by Pinsky et al.⁴⁴), which represents the global warming by the end of the 21st century relative to 1986–2005 under RCP 2.6 and RCP 8.5 scenarios. Note that the average Q_{10} value of 2.6 \pm 0.5 for denitrification and 2.9 \pm 0.8 for N₂O production from two seasons in our study was applied in this equation. The standard deviation of ΔR is calculated using the error propagation.

Statistical analyses. Exponential and linear regressions in this study were conducted using SigmaPlot (v.12.5). The effects of season and process on the Q_{10} and optimum temperature values were examined using two-way ANOVA; the difference of optimum temperature values for anammox between summer and winter was tested using one-way ANOVA. The spatial difference of different parameters were analysed separately using one-sample *t*-test. All statistical analyses were conducted using SPSS (IBM, v.19) at a 0.05 significance level, the *n*, *F*, *t*, degree of freedom (df) and *P* values are provided in Supplementary Tables 3–7.

Reporting Summary. Further information on research design is available in the Nature Research Reporting Summary linked to this article.

Data availability

The data that support the findings of this study can be requested from the corresponding author upon request.

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Author contributions

E.T. and S.-j.K. conceived the study and designed the experiment; E.T., W.Z., M.D. and L.T. performed the experiment and measured the samples; E.T., M.D., T.-C.H., L.T. and S.-j.K. analysed the data; E.T., Z.Z., X.Y., T.-C.H., J.J.M., T.W.T. and S.-j.K. contributed to the discussion of the results and wrote the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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Correspondence and requests for materials should be addressed to S.-j.K.

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Software and code

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Data collection	We collected our data from the published literatures. No software was used.						
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Study description	This study mainly focused on the effect of temperature on sedimentary nitrogen removal rates and associated nitrous oxide production in subtropical regions. This experiment includes 8 temperature sets from 1 to 35°C with a 5°C interval at each sampling site and three replicates for each temperature set.				
Research sample	The sediment samples were collected from different subtropical ecosystems with different salinity and habitat. Only the natural nitrogen removal organisms were included in the study. We also compiled the existing datasets which conducted in temperate and polar regions at the global scale.				
Sampling strategy	The surface 2 cm of sediments were collected using a grab and immediately transferred into a hermetic bag, whereas the overlying water was collected by using a 2 L brown bottle.				
Data collection	The rate-related data was measured using a Thermo Finnigan Delta V Plus isotope ratio mass spectrometer.				
Timing and spatial scale	We conducted campaigns in both summer (20 July 2018) and winter (26 November 2017 and 18 December 2017). Six sites were sampled in each campaign, including estuarine regions, tidal flats, an aquaculture pond and a freshwater creek. All of the sampling sites were included in an area with a range of latitude from 24.3-24.7°N and a range of longitude from 117.7-118.5°E.				
Data exclusions	No data was excluded from the analysis in this study.				
Reproducibility	All attempts to repeat the experiment were successful.				
Randomization	The sediments collected from one site were mixed with the overlying water and then the slurry was added into 15 vials, these vials were allocated into three groups randomly.				
Blinding	We collected the historical data on the temperature dependence of nitrogen removal processes in temperate and polar regions, so blinding was not possible.				
Did the study involve field work? 🛛 📉 Yes 🔄 No					

Field work, collection and transport

Field conditions	The temperature during field sampling was 16-24°C and 30-37°C in winter and summer, respectively, and no rainfall was experienced.			
Location	The samples were collected from Jiulong River Estuary and the adjacent coastal bays, a typical subtropical region in South China, with water depth of 0-11.2 m, latitude of 24.3-24.7°N and longitude of 117.7-118.5°E.			
Access and import/export	The sediments and water samples were stored in an ice chest in the field and transported back to laboratory within 4 h for chemical analyses and incubation experiments. Our permits were obtained from Xiamen University.			
Disturbance	Nearly no disturbance was caused by the study since only ~500 mL sediments and water were collected from each site.			

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Laboratory animals	This study did not involve any laboratory animals.			
Wild animals	We collected sediment samples from aquatic ecosystems, involving different bacteria under natural conditions.			
Field-collected samples	The field samples were collected from natural ecosystems, including estuary, tidal flat and mangroves, with temperature of 16-24°C and 30-37°C in winter and summer, respectively.			
Ethics oversight	No ethical approval or guidance was required since our study just include natural bacteria.			

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