

Production of NO and N₂O by Pure Cultures of Nitrifying and Denitrifying Bacteria during Changes in Aeration†

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Peak emissions of NO and N₂O are often observed after wetting of soil. The reactions to sudden changes in the aeration of cultures of nitrifying and denitrifying bacteria with respect to NO and N₂O emissions were compared to obtain more information about the microbiological aspects of peak emissions. In continuous culture, the nitrifier *Nitrosomonas europaea* and the denitrifiers *Alcaligenes eutrophus* and *Pseudomonas stutzeri* were cultured at different levels of aeration (80 to 0% air saturation) and subjected to changes in aeration. The relative production of NO and N₂O by *N. europaea*, as a percentage of the ammonium conversion, increased from 0.87 and 0.17%, respectively, at 80% air saturation to 2.32 and 0.78%, respectively, at 1% air saturation. At 0% air saturation, ammonium oxidation and N₂O production ceased but NO production was enhanced. Coculturing of *N. europaea* with the nitrite oxidizer *Nitrobacter winogradskyi* strongly reduced the relative levels of NO and N₂O production, probably as an effect of the lowered nitrite concentration. After lowering the aeration, *N. europaea* produced large short-lasting peaks of NO and N₂O emissions in the presence but not in the absence of nitrite. *A. eutrophus* and *P. stutzeri* began to denitrify below 1% air saturation, with the former accumulating nitrite and N₂O and the latter reducing nitrate almost completely to N₂. Transition of *A. eutrophus* and *P. stutzeri* from 80 to 0% air saturation resulted in transient maxima of denitrification intermediates. Such transient maxima were not observed after transition from 1 to 0%. Reduction of nitrate by *A. eutrophus* continued 48 h after the onset of the aeration, whereas N₂O emission by *P. stutzeri* increased for only a short period. It was concluded that only in the presence of nitrite are nitrifiers able to dominate the NO and N₂O emissions of soils shortly after a rainfall event.

Nitrous oxide (N₂O) and nitric oxide (NO) are trace gases contributing to the greenhouse effect, ozone depletion of the stratosphere, and photochemical air pollution (6, 20). Nitrifying bacteria, specifically the chemolithotrophic ammonia oxidizers, and denitrifying bacteria appear to be the main biological sources of NO and N₂O (11). Emission of NO and N₂O is highly variable throughout the year, and peak emissions are reported after (artificial) heavy rainfall (5, 7, 8, 13, 14, 16). Weekly N₂O emission measurements in an acid oak-beech forest soil revealed that three individual peaks resulted in 63% of the total annual emission (28). The importance ascribed to nitrification and denitrification in relation to the NO and N₂O emissions after wetting varies in the above-cited studies. On the one hand, denitrifying enzyme activity can persist during dry periods (9, 24, 25) facilitating a rapid response of denitrification upon wetting, and in addition, de novo synthesis of denitrifying enzymes begins within 4 to 8 h after wetting (24). On the other hand, the activity of nitrifying organisms has also been reported to result in NO and N₂O peaks shortly after wetting of soil (8, 9, 16). In soil, the release of nitrifier and denitrifier substrates after wetting may selectively enhance the effect of the rapid decrease of oxygen availability on NO and N₂O emissions by nitrifiers or denitrifiers. In this study we compared the reaction of nitrifying and denitrifying bacteria to sudden changes in aeration and related the reactions to the peak emissions of NO and N₂O observed in soil studies. By

using a continuous culture technique, aeration could be controlled at a constant rate of substrate supply.

MATERIALS AND METHODS

Organisms. The obligate chemolithotrophic ammonia oxidizer *Nitrosomonas europaea* ATCC 19718, the facultative chemolithotrophic nitrite oxidizer *Nitrobacter winogradskyi* ATCC 14123, and the chemoorganotrophic denitrifiers *Pseudomonas stutzeri* LMAU P12 and *Alcaligenes eutrophus* LMD 82.41 were the organisms used in this study. *N. winogradskyi* was used to reduce the nitrite level in one of the *N. europaea* experiments.

Experimental conditions. Organisms were grown separately in continuous culture (1.2 liter culture volume) (Biostat M; Braun, Melsungen, Germany) except for *N. winogradskyi*, which was cocultured with *N. europaea*. The dilution rate was set at 0.030 h⁻¹, 0.023 h⁻¹, or 0.083 h⁻¹ for the *N. europaea* cultures, the *N. europaea* and *N. winogradskyi* mixed culture, or the denitrifier cultures, respectively. The experiments were performed in duplicate except for the *N. europaea* and *N. winogradskyi* mixed culture.

The medium for nitrifiers was as described by Laanbroek and Gerards (19), as this medium prevents wallgrowth which is commonly found with mineral media. The medium contained 5 mM (NH₄)₂SO₄, 10 mM NaCl, 1 mM KH₂PO₄, 0.75 mM CaCl₂, 0.2 mM MgSO₄, 5 mM sodium pyruvate, 0.15% (wt/vol) yeast extract (Difco, Detroit, Mich.), 0.15% (wt/vol) special peptone (Oxoid, Basingstoke, United Kingdom), and 0.1% (vol/vol) trace element solution (described by Verhagen and Laanbroek [30]). Denitrifiers were cultured in acetate-limited medium which contained 6 mM potassium acetate, 5 mM NH₄NO₃, 10 mM NaNO₃, 1 mM KH₂PO₄, 0.2 mM CaCl₂, 1 mM MgSO₄, and 0.1% (vol/vol) trace element solution.

Gas was sparged through the culture and allowed to escape through an exhaust in the top lid of the culture vessel. The level of air saturation and the gas-sparging rate were controlled by a gas flow ratio controller (type B210; Braun). The unit consisted of an air and a nitrogen mass flow controller and was coupled to an oxygen electrode (Ingold, Frankfurt, Germany) in the culture vessel. The gas-sparging rates (around 0.15 liter min⁻¹) were always monitored when gas samples were taken and did not change during transient state. pH was maintained automatically at 7.5 with 0.5 M HCl and 5% (wt/vol) Na₂CO₃ for the nitrifiers and 0.5 M HCl and 0.5 M NaOH for the denitrifiers. Temperature was kept at 25°C, and the medium was stirred at 400 rpm.

N. europaea was successively cultured at 80 (210 μM O₂), 50, 20, 10, 5, 1, 0, 80, and 0% air saturation. Except for the 0% air saturation level, steady-state

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conditions were reached at every air saturation level. At the transition to 0% air saturation, the dilution of the culture by fresh medium was stopped to prevent washout of *N. europaea*. After 3 days at 0% air saturation, the air saturation level was set at 80% and the dilution was resumed after the culture started to grow again. The *N. europaea* and *N. winogradskyi* mixed culture was successively grown to steady state at 80, 10, and 5% air saturation. Steady state was not reached at 1% air saturation. *P. stutzeri* and *A. eutrophus* were successively grown to steady state at 80, 10, 5, 1, 0, 80, and 0% air saturation.

Samples were taken almost daily to monitor acetate, ammonium, nitrate, and nitrite concentrations; optical density and pH of the culture medium; and the NO and N₂O concentrations of the headspace gas. Headspace gas was sampled aseptically through a Teflon-coated rubber septum. Steady state was usually reached after five volume changes after which the above-mentioned parameters remained constant. At steady state, samples for protein content (only *A. eutrophus* and *P. stutzeri*) were taken, cell numbers were determined, and the culture was checked for contamination in addition to the routinely measured parameters. After transition to another air saturation level, the NO and N₂O concentrations in the headspace gas were monitored.

Analytical procedures. Ammonium and nitrate concentrations were determined in the supernatant (15 min at 15,000 × g) of samples stored at -20°C by using a Traacs 800 autoanalyzer (Technicon Instruments Corp., Tarrytown, N.Y.). Directly after sampling, the nitrite concentration was determined colorimetrically (Griess-Ilosvay reaction) at 540 nm with a spectrophotometer. The optical density was measured spectrophotometrically at 660 nm. Acetate concentrations were measured with a gas chromatograph (HRGC Mega 2 Series; Fisons Instruments, Milan, Italy) containing a formic acid-saturated Chromosorb 101 column at 180°C and a flame ionization detector. Cell numbers were counted microscopically in samples stored on ice by using a Bürker-Türk counting chamber on the day of sampling. Tryptone soy broth (1% [wt/vol]; Oxoid) agar plates were used to check for contamination. The protein contents of NaOH-boiled cells were determined by Lowry's method with bovine serum albumin (Sigma, St. Louis, Mo.) as standard.

N₂O concentrations in the headspace gas were measured with a gas chromatograph (GC 6000; Carlo Erba, Milan, Italy) equipped with an electron capture detector. Gases were separated on a Haysep Q column operated at 80°C with helium as carrier gas. A soda lime precolumn was used to absorb carbon dioxide. N₂O (48 ppm [vol]) in nitrogen standard gas (HoekLoos, Dieren, The Netherlands) was used to calibrate the electron capture detector. NO was detected with a NO_x analyzer (Model 42S; Thermo Environmental Instruments Inc., Franklin Mass.) adapted with a sample-mixing unit for small-volume samples (17). Chemically produced NO was used to make standards (by a modification of the method of Goretski et al. [15]). NO was generated by quantitative reduction of nitrite in anoxic vials containing 12 mM potassium iodide in 0.87 M acetic acid. The concentration of NO in the headspace of the vials was calculated by using the Bunsen absorption coefficient (27).

Statistical procedures. Statistical differences were determined by one-way analyses of variance combined with Tukey tests ($P < 0.05$) by using the Statistix 4.0 software package (Analytical Software, St. Paul, Minn.).

RESULTS

Nitrification. The steady-state ammonium and nitrite concentrations were between 1 and 2 mM and between 7.8 and 8.8 mM, respectively, in the *N. europaea* cultures at 80, 50, 20, 10, and 5% air saturation. At 1% air saturation, the ammonium concentration increased to 2.65 mM and the nitrite concentration decreased to 7.34 mM. The effects of aeration on the ammonium and nitrite concentrations, however, were not significant. Cell numbers at steady state, on average 1.99×10^8 cells ml⁻¹ (standard error, 0.42×10^8), were not significantly influenced by aeration. The fraction of ammonium converted into NO increased with the decreasing air saturation level from below 1% at 80% air saturation to 2.3% at 1% air saturation (Table 1). At 0% air saturation and a dilution rate of 0, NO was the single nitrogen oxide released by the culture (data not shown). The fraction of N₂O was lower than the fraction of NO at every level of air saturation. The relative N₂O emission from the cultures was around 0.15% in the range of 80 to 5% air saturation but increased considerably at 1% air saturation. At 0% air saturation (culture dilution rate, 0) the N₂O emission by the culture stopped (data not shown).

The nitrite concentrations in a mixed culture of *N. europaea* and *N. winogradskyi* were below 0.1 mM at steady state at 80, 10, and 5% air saturation. At 1% air saturation, steady state of the mixed culture was not reached, probably due to washout of

TABLE 1. The steady-state productions of nitric oxide and nitrous oxide by *N. europaea* in a chemostat ($D = 0.030 \text{ h}^{-1}$) at different levels of air saturation

% Air saturation	Mean (SE) % production ^a	
	NO	N ₂ O
80	0.87 a (0.16)	0.17 a (0.02)
50	1.18 ab (0.11)	0.17 a (0.04)
20	1.35 ab (0.21)	0.13 a (0.01)
10	1.59 ab (0.39)	0.16 a (0.16)
5	1.42 ab (0.08)	0.15 a (0.01)
1	2.32 b (0.13)	0.78 b (0.19)

^a The productions of nitric oxide and nitrous oxide were calculated as mean percentages of the total amount of converted ammonium for two individual continuous cultures. Within a column, nonsignificant differences are marked with the same letter (Tukey test, $P < 0.05$).

N. winogradskyi. The ammonium concentrations in the mixed culture were more or less the same as the concentrations observed in the pure cultures of *N. europaea*. The fraction of NO was below 0.2% at every air saturation level tested, and the fraction of N₂O was below 0.1% (Table 2). Washout of *N. winogradskyi* at 1% air saturation resulted in increased emissions of NO and N₂O. Nonbiological NO and N₂O production, tested in aseptic growth medium containing 10 mM nitrite instead of ammonium, was insignificant above pH 6.5 (data not shown).

The NO and N₂O emissions for the *N. europaea* cultures were monitored directly after the air saturation level was lowered. Transitions from 10 to 5% and from 5 to 1% air saturation resulted in increased NO and N₂O fluxes (Fig. 1A and B). Increased NO and N₂O fluxes were immediately visible upon the shift in air saturation with maxima after approximately 15 min, which subsequently subsided over periods of 1 h and 4 h, respectively. Transition from 1 to 0% air saturation (dilution rate, 0) enhanced the NO flux but not the N₂O flux (Fig. 1C). The enhanced NO flux continued throughout the 3-day anoxic period (data not shown), in contrast to the short-lasting peaks after the 10 to 5% and 5 to 1% transitions. The switch from 80 to 0% air saturation (dilution rate, 0) resulted in NO and N₂O peaks during the first hour and an increased level of the NO flux later on, whereas the N₂O flux stopped after 2 h (Fig. 1D). In the *N. europaea* cultures, transition from 20 to 10% air saturation was followed by small NO and N₂O peaks, and after transitions from 80 to 50% and 50 to 20% air saturation, no NO and N₂O peaks were detected (data not shown).

Like the *N. europaea* pure cultures, a mixed culture of *N. europaea* and *N. winogradskyi* was subjected to changes in air saturation level. Transition from 80 to 10% resulted in a N₂O peak (Fig. 2A) which was, however, very small compared to the peaks found in the *N. europaea* pure cultures. No peaks were

TABLE 2. The steady-state productions of nitric oxide and nitrous oxide by a *N. europaea* and *N. winogradskyi* mixed culture in a chemostat ($D = 0.023 \text{ h}^{-1}$) at different levels of air saturation

% Air saturation	% Production ^a	
	NO	N ₂ O
80	0.14	0.08
10	0.18	0.04
5	0.20	0.06

^a The productions of nitric oxide and nitrous oxide were calculated as percentages of the total amount of converted ammonium.

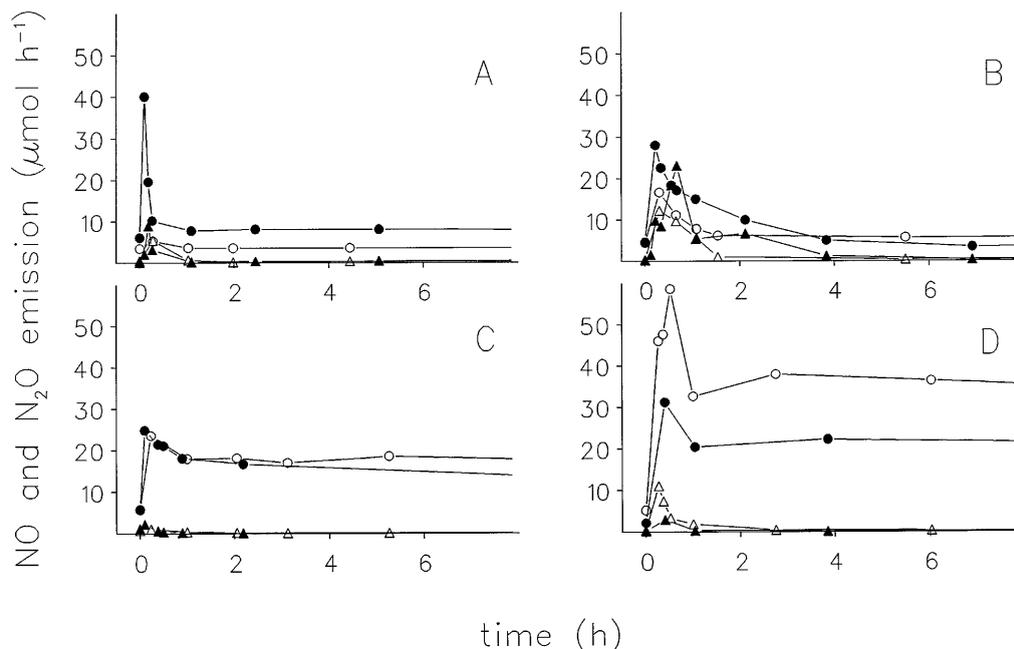


FIG. 1. Emission of nitric oxide and nitrous oxide by two continuous cultures of *N. europaea* during transient state after a change in the level of air saturation. At time zero air saturation was switched from 10 to 5% (A), 5 to 1% (B), 1 to 0% (C), and 80 to 0% (D). At 0% air saturation the dilution rate was 0. Circles and triangles represent nitric oxide and nitrous oxide emissions, respectively. Open and closed symbols represent the nitric oxide and nitrous oxide emissions of the two individual continuous cultures.

detected after transition from 10 to 5% air saturation (Fig. 2B), although the N_2O flux of the mixed culture showed some increase 4 h after the transition. A small and short-lasting N_2O peak was found after transition from 5 to 1% air saturation, which coincided with a small dip in the NO flux of the mixed culture (Fig. 2C). As reported above, *N. winogradskyi* washed

out at 1% air saturation, which coincided with the slow increase of the NO and N_2O fluxes from that mixed culture.

Denitrification. At steady state, the acetate concentrations in the *A. eutrophus* cultures were below the detection limit (less than 0.1 mM acetate) and between 0.66 and 0.78 mM for the *P. stutzeri* cultures. The protein yield of the *A. eutrophus* cultures

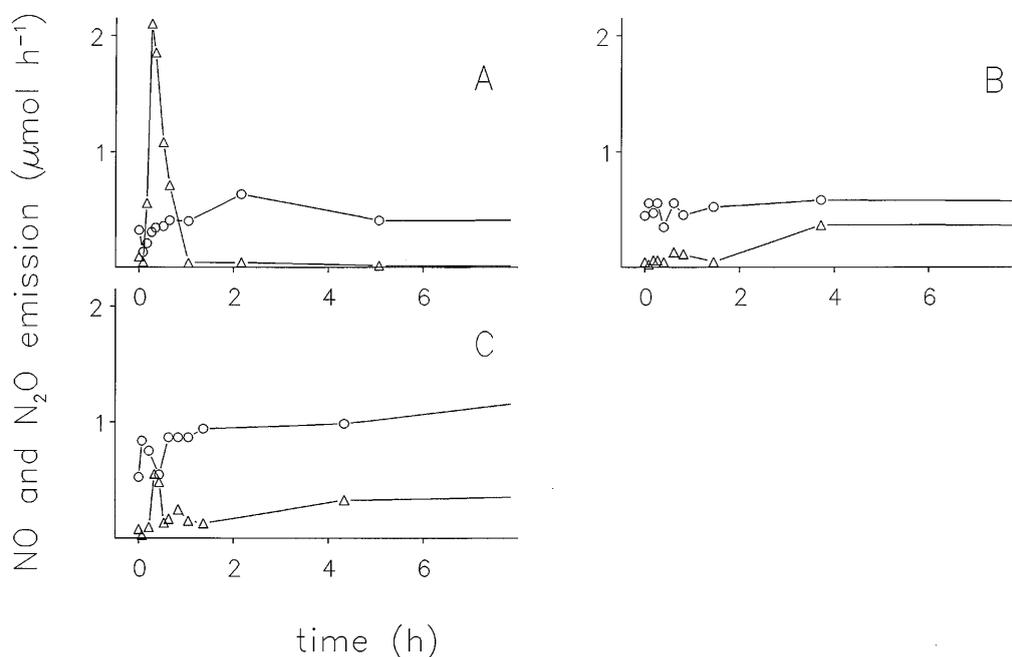


FIG. 2. Emission of nitric oxide and nitrous oxide by a continuous culture of *N. europaea* and *N. winogradskyi* during transient state after a change in the level of air saturation. At time zero air saturation was switched from 80 to 10% (A), 10 to 5% (B), and 5 to 1% (C). Circles and triangles represent nitric oxide and nitrous oxide emissions, respectively.

TABLE 3. The steady-state productions of nitrite, nitric oxide, and nitrous oxide by *A. eutrophus* and *P. stutzeri* in acetate-limited chemostats ($D = 0.083 \text{ h}^{-1}$) at different levels of air saturation

Strain	% Air saturation	Mean (SE) production ($\mu\text{mol liter}^{-1}$ of culture h^{-1}) ^a		
		NO ₂ ⁻	NO	N ₂ O
<i>A. eutrophus</i>	80	5.5 (3.8) a	0.006 (0.006) a	0.07 (0.04) a
	10	4.3 (2.6) a	0.006 (0.002) a	0.02 (0.00) a
	5	19.0 (9.6) a	0.004 (0.004) a	0.23 (0.02) a
	1	41.5 (11.6) a	0.006 (0.002) a	0.21 (0.02) a
	0	588.9 (55.6) b	2.062 (0.391) b	79.99 (7.78) b
<i>P. stutzeri</i>	80	0.4 (0.4) a	0.000 (0.000) a	0.06 (0.06) a
	10	10.1 (10.1) a	0.132 (0.132) a	0.28 (0.20) ab
	5	5.3 (4.4) a	0.136 (0.127) a	0.27 (0.15) ab
	1	3.1 (3.1) a	0.140 (0.114) a	0.31 (0.28) ab
	0	18.9 (0.4) b	0.702 (0.009) b	1.29 (0.22) b

^a Data are mean percent production amounts for two individual continuous cultures for each species. Within a column (per species), nonsignificant differences are marked with the same letter (Tukey test, $P < 0.05$).

decreased from 13.9 g at 80% air saturation to 7.63 g of protein per mol of acetate at 0% air saturation. The protein yield of *P. stutzeri* was lower than the yield of *A. eutrophus* and ranged between 9.8 and 7.1 g of protein per mol of acetate at 80 and 0% air saturation, respectively. Both cultures were acetate limited as was shown by elevations in cell number, optical density, and protein content upon doubling the acetate concentration in the growth medium (data not shown).

Anoxity strongly enhanced the production of nitrite, NO,

and N₂O by *A. eutrophus* (Table 3). Between 80 and 1% air saturation, there were no significant effects of the level of air saturation on the production of nitrite, NO, and N₂O. The effects of the air saturation level on the production of nitrite, NO, and N₂O by *P. stutzeri* were similar to the effects found with *A. eutrophus* but less pronounced. The nitrite and N₂O production by *A. eutrophus* at anoxia was significantly higher than that of *P. stutzeri*. As calculated from the nitrogen balance under acetate limitation, *P. stutzeri* tended to reduce nitrate completely to N₂ (96% of reduced nitrate), whereas *A. eutrophus* accumulated the intermediates nitrite and N₂O besides N₂ (56, 15, and 29% of reduced nitrate, respectively).

The effects of changed aeration on NO and N₂O emissions and nitrite accumulation by *A. eutrophus* and *P. stutzeri* are shown in Fig. 3 for one replicate. The patterns of NO and N₂O emission and nitrite accumulation were the same between replicates. For both denitrifiers, it took less time to reach steady state after the 1 to 0% air saturation transition than after the 80 to 0% air saturation transition. Distinct transient maxima of NO and nitrite were observed after the 80 to 0% air saturation transition.

NO and N₂O emissions ceased almost immediately when oxygen was supplied to the anoxic *A. eutrophus* culture (Fig. 3C), whereas N₂O emission of *P. stutzeri* was shortly but strongly enhanced under these conditions (Fig. 3F). The nitrite concentration in the *A. eutrophus* culture decreased gradually. Calculated from the dilution rate, substantial residual nitrite production continued for up to 48 h after the aeration switch. In the *P. stutzeri* culture, the rate of decrease of the nitrite concentration after the 0 to 80% air saturation transition indicated some residual nitrite production for up to 29 h. Tran-

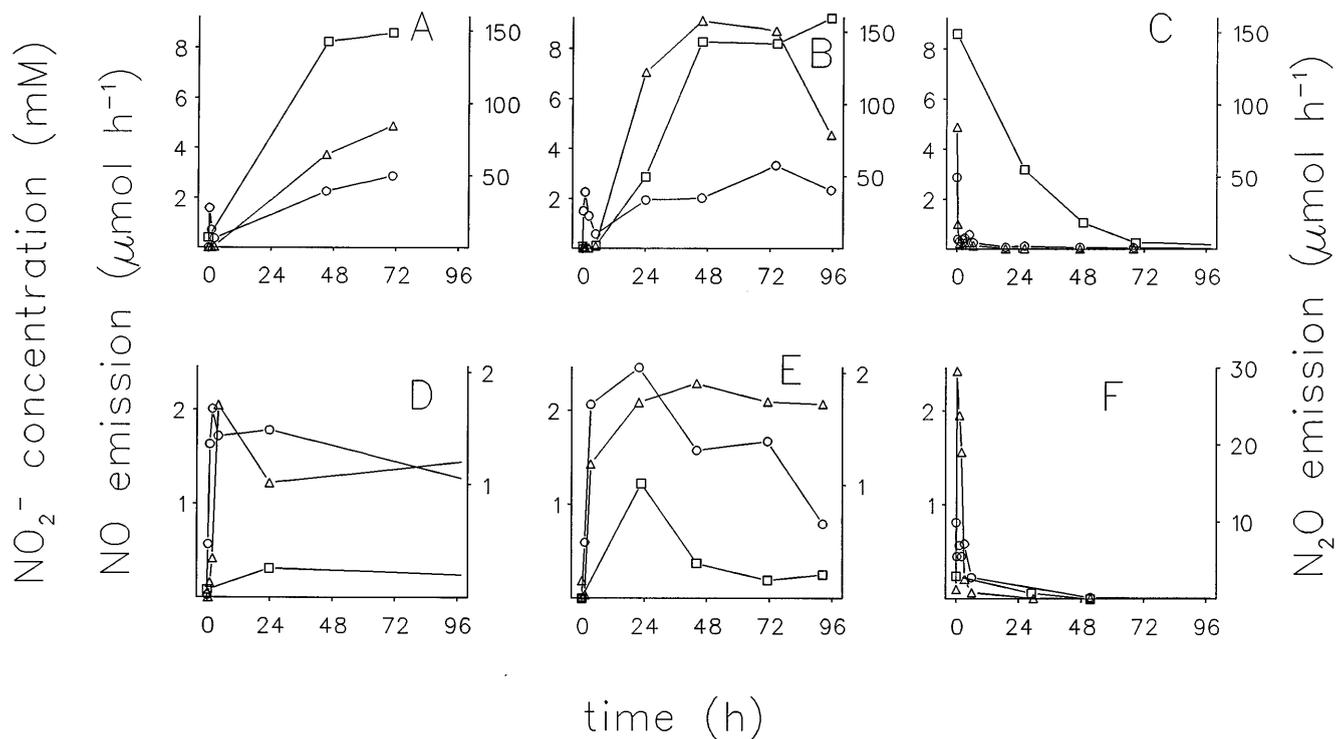


FIG. 3. Emission of nitric oxide and nitrous oxide and the accumulation of nitrite in representative continuous cultures of *A. eutrophus* (A to C) and *P. stutzeri* (D to F) during transient state after a change in the level of air saturation. At time zero air saturation was switched from 1 to 0% (A and D), 80 to 0% (B and E), and 0 to 80% (C and F). Circles, triangles, and squares represent nitric oxide emission (left axis), nitrous oxide emission (right axis), and nitrite accumulation (left axis), respectively.

sitions from 80 to 10%, 10 to 5% and 5 to 1% air saturation did not influence the emission of NO and N₂O by either denitrifier (data not shown).

DISCUSSION

Nitrification. Down to 1% air saturation, oxygen concentration did not significantly influence the growth characteristics of *N. europaea*, which is in accordance with the results of Goreau et al. (14). The relatively high amounts of ammonium in the culture medium at steady state are explained by the high K_m for ammonia of *N. europaea* (19). The fraction of nitrogen converted to NO and N₂O increases with a decreasing air saturation level. Although the fractions of NO and N₂O at oxygen-poor steady states are low compared to the fractions found in some other studies (e.g., 19% N loss [26] and nearly 10% N₂O-N [14]), they are in good agreement with the results of Remde and Conrad (23), 1.4 to 2.7% NO and 0.1 to 3.9% N₂O (depending on cell density), and Anderson et al. (1), 2.6% NO and 1% N₂O. The discrepancies with the high fractions of NO and N₂O reported in other studies might be due to dissimilar behaviors of the different strains of *N. europaea* used or to the use of different incubation systems (i.e., batch versus continuous culture). The strain used in this study was the same as that used by Anderson et al. (1).

Anderson et al. (1) found an optimum for NO and N₂O production by *N. europaea* between 0.2 kPa and 0.4 kPa pO₂ (approximately 1 and 2% air saturation, respectively) at low nitrite concentrations. The authors stated that at high nitrite concentrations the influence of pO₂ is limited, perhaps because of competition between the terminal oxidase and nitrite reductase for electrons. Reduction of nitrite is thought to be the prime source of NO and N₂O emissions by nitrifiers. The weak enhancement of NO and N₂O production at reduced oxygen concentrations in the *N. europaea* cultures in our study is probably also related to the high nitrite concentrations.

The strongly reduced NO and N₂O emissions by the *N. europaea* and *N. winogradskyi* mixed culture was probably due to the low nitrite concentration in the mixed culture. It is possible that NO consumption by *N. winogradskyi* may have occurred as well, as reported by Freitag and Bock (12).

The peak emissions of NO and N₂O after changes in aeration have, to the best of our knowledge, never been reported before for *N. europaea* cultures. However, Bock et al. (3) described a large loss of nitrogen on the same day for a *N. europaea* culture whose oxygen concentration was decreased from 0.4 to 0.2 mg/liter, but this was not recognized as a peak emission by the authors. The switch in oxygen supply in our study apparently disturbed the metabolism of *N. europaea* at 10 and 5% air saturation, resulting in the short-lasting peaks of NO and N₂O emission. These peaks were not observed in the *N. europaea* and *N. winogradskyi* mixed culture, suggesting again that nitrite concentration is an important factor for the production of NO and N₂O. The high gas-sparging rate in the culture makes immediate NO utilization by *N. winogradskyi* an unlikely explanation for the decreased NO production.

The patterns of NO and N₂O emission by *N. europaea* after transition from 80 to 0% air saturation resemble a combination of the 5 to 1% and 1 to 0% patterns. This is probably the result of the relatively slow decrease of the oxygen concentration (it took 9 min to go from 80 to 0% air saturation). In contrast to the nitrite reductase of *N. europaea* studied by Miller and Nicholas (21), which was only induced at low oxygen tension, our results suggest that nitrite reductase must have been present in large amounts at 80% air saturation in order to allow for peaks in NO and N₂O immediately after transition.

The mechanism of NO and N₂O peak production by *N. europaea* and the role of nitrite has yet to be elucidated.

Denitrification. *A. eutrophus* and *P. stutzeri* cultures started to denitrify below 1% air saturation. In the *A. eutrophus* cultures, transition from 1 to 0% air saturation resulted in gradual increases of NO and N₂O, whereas a transient N₂O maximum was observed after the transition from 80 to 0% air saturation. In addition, it took more time to reach steady state in the latter case, suggesting that the N reductases were already present at 1% air saturation but not at 80% air saturation. Measurements of the N reductase activity of the related *Alcaligenes faecalis* at different aeration levels support this explanation (22).

In the *P. stutzeri* cultures switched from 1 to 0% air saturation, steady-state levels of NO and N₂O emission and nitrite accumulation were reached within 24 h. After the transition from 80 to 0% air saturation, transient maxima were observed and it took more time to reach steady-state levels. Körner and Zumft (18) observed elevated expressions of nitrate reductase, nitrite reductase, and nitrous oxide reductase (nitric oxide reductase was not tested) at 17% air saturation and below in *P. stutzeri*. The results in the present study indicate that at 1% air saturation *P. stutzeri* cells contained the complete denitrifying pathway as denitrification commenced immediately and with relatively low transient accumulation of intermediates. The unexpected rapid increases of NO and N₂O after transition from 80 to 0% air saturation suggests that *P. stutzeri* cells are able to induce these reductases within 1 h. Baumann et al. (2) showed that *Paracoccus denitrificans* expressed enhanced levels of NaR and NoS mRNA immediately after an oxic to anoxic transition, and they detected denitrification metabolites after 30 min.

Nitrifier versus denitrifier NO and N₂O emissions. Although it is difficult to directly relate quantitative NO and N₂O emission data from simple chemostat systems to complex natural environments, some of the key underlying mechanisms may hold true. With the organisms used in this study, it was confirmed that denitrifier NO and N₂O emissions are restricted to anoxic environments, whereas nitrifier NO and N₂O emissions are highest at oxygen-poor sites and N₂O emission is absent at 0% oxygen. Significant NO emission by *N. europaea* was found at anoxic conditions and probably requires nitrite and an endogenous substrate or simple organic substrates.

As conditions in soil constantly change, a steady-state-like situation is rarely obtained. Therefore, the dynamics of NO and N₂O emission by nitrifiers and denitrifiers in reaction to changes in the microenvironment of these organisms are probably at least as important as steady-state emission data. The results of this study suggest that rapid decreases in oxygen availability cause larger emissions of NO and N₂O by denitrifiers than do slow decreases. A slow decrease of oxygen availability allows time for adjustment in the activity of the N reductases in the denitrifying pathway. The persistence of denitrifying enzyme activity during drought has not been addressed in this study. Nitrite reductase and nitrous oxide reductase are, however, less persistent than nitrate reductase (10), which suggests that the "old" denitrifying enzyme pool enhances accumulation and emission of denitrification intermediates immediately after rapid decrease in oxygen availability.

This study shows that ammonium-oxidizing bacteria may be able to dominate the emission of N₂O, and especially that of NO, shortly after decreased oxygen availability. Furthermore, substantial NO emission persists under anoxic conditions. The presence of nitrite seems to play an important role in NO and N₂O emissions. Nitrite accumulation in soil is rare, with occasional exceptions such as after the application of alkaline-hydrolyzing ammonium fertilizers or at low moisture content,

and is associated with alkaline conditions (4, 29). Davidson et al. (8) reported accumulation of nitrite in a dry deciduous forest soil and observed pulses of nitrifier NO and N₂O emissions after wetting of neutral soils, whereas Hutchinson et al. (16) observed NO peaks after wetting alkaline soil. Low nitrite concentrations of the bulk soil were detected in the latter case, but drying may concentrate nitrite in thin waterfilms and contribute to nitrite accumulation in microsites (11). NO and N₂O peaks described in this study fit well with the above-mentioned field observations. Accumulation of nitrite in neutral to acidic soils is less likely in temperate regions, except shortly after fertilization, and NO and N₂O peaks after rainfall events are therefore probably dominated by denitrification emission. In agricultural soils, the type of ammonium fertilizer, i.e., acid or alkaline hydrolyzing, may influence the magnitude of the nitrifier NO and N₂O peaks after rainfall events.

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