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# $[\eta^2$ -8-(tert-Butylimino)-1-naphthyl]bis( $\eta^5$ -cyclopentadienyl)titanium(IV)

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#### **Key indicators**

Single-crystal X-ray study T = 298 KMean  $\sigma(C-C) = 0.006 \text{ Å}$  R factor = 0.056 wR factor = 0.131Data-to-parameter ratio = 18.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound,  $[\mathrm{Ti}(C_5H_5)_2(C_{15}H_{15}N)]$ , displays an iminoacyl fragment in an N-outside conformation, indicating that the C $\equiv$ N vector points away from the  $\sigma$ -bonded C atom. Considering only the geometrical centres of all  $\eta$ -coordinated groups, the  $\mathrm{Ti}^{\mathrm{IV}}$  ion adopts a severely distorted tetrahedral configuration. The crystal packing displays a short C $-\mathrm{H}\cdots\pi$  contact linking the molecules into an infinite chain running along the b-axis direction.

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#### Comment

The structure of the title compound, (I), was determined in the course of our investigations on the synthesis and reactivity of small metallacycles. A similar compound has been reported by Campora *et al.* (1995) and a related compound with Zr by Berg & Petersen (1989).

$$(CH_3)_3C$$
 $N$ 
 $T_1$ 
 $Cp$ 
 $T_1$ 
 $Cp$ 

The structure of (I) is shown in Fig. 1, and selected bond lengths and angles are listed in Table 1. The title compound has two  $\eta^5$ -coordinated cyclopentadienyl (Cp) rings coordinated to the Ti<sup>IV</sup> ion. The distances between the perpendicular projection of the Ti<sup>IV</sup> ion on the Cp rings and their ring centroids (ring slippage) are 0.043 and 0.017 Å for the rings containing C16 and C21, respectively. The title compound further displays an iminoacyl fragment in an N-outside conformation, indicating that the C=N vector points away from the  $\sigma$ -bonded atom C2 (Tatsumi et al.., 1985). When the Cp rings and the C=N group are represented by their geometrical centres, as is suggested in Fig. 1, the Ti<sup>IV</sup> ion displays a severely distorted tetrahedral configuration, with X-Ti-Y angles in the range 88.24 (11)–133.16 (9)° (Table 3). In an alternative description, the Ti<sup>IV</sup> ion takes part in two adjacent titanacycles, the three-membered ring Ti1/C11/N11 and the five-membered ring Ti1/C2/C1/C10/C11. The maximum deviation from the least-squares plane through this last ring system is 0.028 (3) Å for C2. The Ti1-C2 bond is elongated by 0.2 Å compared with its parent compound 1,8naphthalenediyltitanocene (Tinga et al., 1994). The crystal packing displays a short  $C-H \cdot \cdot \pi$  contact (Fig. 2 and Table 2),

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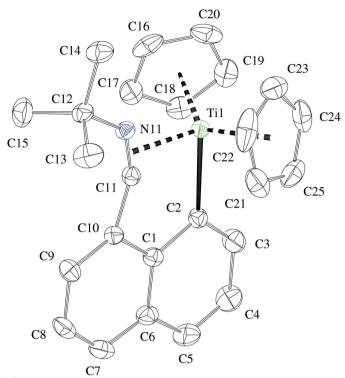


Figure 1 A view of the title compound, showing 30% probability displacement ellipsoids. H atoms have been omitted for clarity.

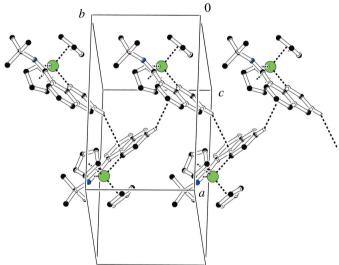


Figure 2 Section of the crystal packing illustrating the C4-H4 $\cdots\pi$  interactions, viewed approximately perpendicular to the  $(10\overline{2})$  planes. H atoms, with the exception of H4, have been omitted.

which links the molecules into an infinite chain running along the b-axis direction.

# **Experimental**

A solution of 1,8-naphthalenediyltitanocene (0.1 mmol), prepared according to Tinga et al. (1994), in pentane (10 ml) was added to a solution of CN-<sup>t</sup>Bu (0.2 mmol) in toluene (2 ml) at room temperature. After stirring for 3 h, the solvents were evaporated and the residue was extracted with cyclopentane. Brown crystals were obtained after a week from a solution in cyclopentane, which was kept at 280 K. Spectroscopic analysis: <sup>1</sup>H NMR (200 MHz, room temperature, C<sub>6</sub>D<sub>6</sub>, reference  $C_6D_5H = 7.17$  p.p.m., p.p.m.):  $\delta 7.85$  (m, 1 H, H3), 7.83 (m, 1 H, H9), 7.68 (m, 1 H, H5), 7.58 (m, 1 H, H4), 7.50 (m, 1 H, H7), 7.37 (m, 1 H, H8), 5.20 (s, 10 H, 2Cp), 1.21 (s, 9 H, <sup>t</sup>Bu) (at 240 K a similar proton NMR spectrum was measured in CD2Cl2); 13C NMR (50.3 MHz, room temperature,  $C_6D_6$ , reference  $C_6D_6 = 128.0$  p.p.m., p.p.m.):  $\delta$  220.2 (d,  $^{3}J$  = 3.3 Hz, C11), 191.2 (m, C2), 158.1 (m, C10), 140.3 (ddd,  ${}^{1}J$  = 154.1 Hz,  ${}^{2}J$  = 1.2 Hz,  ${}^{3}J$  = 7.5 Hz, C3), 137.3 (d,  ${}^{3}J$  = 7.6 Hz, C1 or C6), 135.0 (dd,  $^{3}J$  = 8.0/6.6 Hz, C6 or C1), 132.0 (d,  $^{1}J$  = 164.0 Hz, C9), 126.9 (dd,  ${}^{1}J$  = 150.9 Hz;  ${}^{2}J$  = 2.3 Hz, C4), 124.3 (dd,  ${}^{1}J$ = 158.4 Hz,  ${}^{2}J$  = 1.9 Hz, C8), 121.6 (dd,  ${}^{1}J$  = 157.3 Hz,  ${}^{3}J$  = 8.9 Hz, C7), 121.1 (ddd,  ${}^{1}J = 157.3 \text{ Hz}$ ,  ${}^{3}J = 9.4/8.2 \text{ Hz}$ , C5), 105.5 (dm,  ${}^{1}J =$ 171.0 Hz, 2Cp), 59.2 (d,  ${}^{2}J$  = 4.2 Hz, C12), 29.3 (gsept,  ${}^{1}J$  = 125.4 Hz,  ${}^{3}J$ = 4.7 Hz, <sup>t</sup>Bu); IR (hexane)  $\bar{\nu}$  (cm<sup>-1</sup>) 1726 (s); HRMS (EI) calculated  $C_{21}H_{16}N^{48}Ti$ : 330.0764; found: 330.076; MS m/z (relative intensities): 387 (not observed,  $M^+$ ), 330 (77, [M - $^{t}Bu]^{+}$ ), 303 (17,  $[M - CHN^{-t}Bu]^{+}$ ), 178 (38,  $Cp_{2}Ti^{+}$ ), 128 (100, naphthalene, impurity).

#### Crystal data

$[Ti(C_5H_5)_2(C_{15}H_{15}N)]$	Z = 4
$M_r = 387.33$	$D_x = 1.297 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
a = 15.0891 (12)  Å	$\mu = 0.44 \text{ mm}^{-1}$
b = 8.4475 (12)  Å	T = 298  K
c = 15.6025 (18)  Å	Block, brown
$\beta = 94.200 (13)^{\circ}$	$0.5 \times 0.4 \times 0.2 \text{ mm}$
$V = 1983.4 (4) \text{ Å}^3$	

### Data collection

Enraf-Nonius CAD-4	2344 reflections with $I > 2\sigma(I)$
diffractometer	$R_{\rm int} = 0.200$
$\omega/2\theta$ scans	$\theta_{\text{max}} = 27.5^{\circ}$
Absorption correction: none	3 standard reflections
5661 measured reflections	frequency: 60 min
4533 independent reflections	intensity decay: 1.5%

# Refinement

Refinement on $F^2$	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.056$	$w = 1/[\sigma^2(F_0^2) + (0.0508P)^2]$
$wR(F^2) = 0.131$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} < 0.001$
4533 reflections	$\Delta \rho_{\text{max}} = 0.28 \text{ e Å}^{-3}$
247 parameters	$\Delta \rho_{\min} = -0.28 \text{ e Å}^{-3}$

Table 1 Selected geometric parameters (Å, °).

Ti1-N11	2.197 (3)	Ti1-C11	2.043 (3)
Ti1-C2	2.292 (3)	N11-C11	1.238 (4)
N11-Ti1-C2 N11-Ti1-C11 C2-Ti1-C11	104.45 (11) 33.70 (12) 70.75 (12)	C11-N11-C12 N11-C11-C10	132.5 (3) 150.8 (3)

Table 2 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$C4-H4\cdots Cg4^{i}$	0.93	2.98	3.768 (5)	143

Symmetry code: (i) -x + 1,  $y - \frac{1}{2}$ ,  $-z + \frac{1}{2}$ . Cg4 is the centroid of the C1/C6/C7/C8/C9/C10

**Table 3** Tetrahedral geometry of Ti1 (angles in °).

Cg1 represents the centroid of the Cp ring C16–C20; Cg2 represents the centroid of the Cp ring C21–C25; Cg3 represents the mid-point of the N=C bond.

C2-Ti1-Cg1	103.22 (10)
C2-Ti1-Cg2	101.46 (11)
C2-Ti1-Cg3	88.24 (11)
Cg1-Ti1-Cg2	133.16 (9)
Cg1-Ti1-Cg3	110.35 (9)
Cg2-Ti1- $Cg3$	109.60 (10)

H atoms were introduced in calculated positions, with C—H = 0.93 and 0.96 Å, and refined as riding on their carrier atoms, with  $U_{\rm iso}({\rm H})$  = 1.5 $U_{\rm eq}({\rm methyl~C})$  or 1.2 $U_{\rm eq}({\rm other~C})$ . The methyl groups were refined as rigid groups, allowing for rotation around the C—C bond.

Data collection: locally modified *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *SET4* (de Boer & Duisenberg, 1984); data reduction: *HELENA* (Spek, 1997); program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985); program(s) used to refine structure:

ture: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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