Empirical Relations Predicting Human and Rat Tissue:Air Partition Coefficients of Volatile Organic Compounds

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Based on the hypothesis that tissue partitioning of volatile organic compounds (VOCs) is due to lipophilic and hydrophilic interactions with tissue components, empirical relations are established between olive oil ($P_{oil:air}$), saline ($P_{saline:air}$), and tissue partition coefficients ($P_{tissue:air}$) for human and rat tissues. Reported values of partition coefficients of a wide range of VOCs with distinct chemical structures (n = 137) have been compiled from the literature. Bilinear regression analysis shows that partition coefficients of VOCs in human blood, brain, fat, liver, kidney, and muscle tissues are well described by a linear combination of $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ with tissue-specific regression coefficients. The regression coefficient associated with the hydrophilic component of VOC partitioning in rat tissues is systematically higher than that of human tissues. For the human model, tissue concentrations calculated from predicted partition coefficients are generally within a factor 4 of tissue concentrations calculated from experimentally observed partition coefficients. These results demonstrate that, without prior knowledge of tissue composition, it is possible to obtain estimates of human tissue partition coefficients of VOCs with an accuracy that is in the same range as that commonly used in risk assessment. © 2000 Academic Press

Key Words: Organic solvents; tissue partition coefficients; human PBPK modeling.

Knowledge of the distribution of chemicals over different body compartments contributes to the understanding of the risk of toxic effects. Ambient exposure to volatile organic compounds (VOCs), particularly in the occupational setting, may cause adverse, neurotoxic effects (reviewed by Mikkelsen, 1997; White and Proctor, 1997). Despite their neurotoxic potential, only a few studies have addressed the relation between exposure and brain concentrations of VOCs, which constitute a large, heterogeneous class of chemicals. Alternatively, physiologically based pharmacokinetic (PBPK) models have been used to describe the relation between inhalation exposure to and tissue concentrations of VOCs (Andersen, 1991; Krishnan and Andersen, 1994; Gargas *et al.*, 1995). However, a specific brain compartment is often lacking in PBPK models. The detailed modeling of brain concentrations is hampered by a general lack of knowledge of brain tissue partition coefficients.

Partitioning between blood and a specific tissue depends on the relative affinities of a compound for blood and for the tissue. For volatile substances it is more convenient to determine tissue:air partition coefficients (Sato and Nakajima, 1979a), and blood:tissue partition coefficients are defined as the ratio of the blood:air partition coefficient $(P_{blood:air})$ and the tissue:air partition coefficient ($P_{\text{tissue:air}}$). Since the relative proportions and the basic composition of tissue constituents vary among tissues, the prediction of tissue partitioning on a rational basis requires detailed knowledge of tissue composition (Poulin and Krishnan, 1995a,b). However, a simplified approach supposes that tissue partitioning of nonreactive chemicals is determined completely by lipophilic and hydrophilic interactions of compounds with tissue constituents. It has been shown before that such a simple approach successfully applies to a set of 12 volatile anesthetics. The $P_{\text{tissue:air}}$ of these volatile anesthetics in human tissues can be described as linear combinations of $P_{\text{saline:air}}$ and $P_{\text{olive oil:air}}$ (Droz, 1978).

Demonstration of the applicability of this approach to tissue partitioning of VOCs in general would provide a basis for predicting tissue partitioning without prior knowledge of the tissue composition. Despite the neurotoxic potential of VOCs, human tissue partition coefficients for industrial important organic solvents, e.g., the alkylbenzenes, appear to be lacking in the literature and, in general, brain tissue partition coefficients have been reported for a limited number of compounds only. Since brain concentrations may be a key issue for the risk assessment of VOCs, detailed knowledge on VOC partitioning in brain tissue is required.

Here we have compiled published values of partition coefficients for a large number of VOCs in olive oil and saline, as well as partition coefficients in rat and human blood, fat, brain, liver, muscle, and kidney tissues. Using linear regression analysis, empirical relations are established between the partitioning of VOCs in tissues, olive oil, and saline, based on the



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approach used for volatile anesthetics before (Droz, 1978). The results demonstrate that VOC partitioning into tissues can be described by a linear combination of the saline and oil partition coefficients with tissue- and species-dependent coefficients.

METHODS

Partition coefficients. Values of $P_{\text{olive oil:air}}$, $P_{\text{saline:air}}$, and $P_{\text{tissue:air}}$ for human and rat blood, fat, brain, liver, muscle, and kidney were compiled from various sources. The reported values are compiled in Table 1. Values for $P_{\text{olive oil:air}}$, $P_{\text{saline:air}}$, and human $P_{\text{blood:air}}$ for various organic solvents were first reported by Droz (1978) and by Sato and Nakajima (1979a,b). Steward et al. (1973) compiled human tissue partition coefficients for various anesthetics. Paterson and Mackay (1989) compiled water solubility values for some compounds. Extensive data sets containing partition coefficients for various human tissues and a large number of compounds were reported by Fiserova-Bergerova et al. (1984), Perbellini et al. (1985), and Fiserova-Bergerova and Diaz (1986). Some sources contained values of human P_{blood:air} for a few compounds only. These values were included when the corresponding $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ were reported as well (Johanson and Dynésius, 1988; Järnberg and Johanson, 1995; Nihlén and Johanson, 1995). A large set of rat tissue partition coefficients with corresponding values of Poiliair and Psalineair for various VOCs published by Gargas et al. (1989) was included and extended with rat tissue partition coefficients for ethylbenzene (Tardif et al., 1997). Kaneko et al. (1994) reported rat data for alcohols and esters, which were compiled and supplemented with additional values for ketones by Poulin and Krishnan (1996a,b). Pierce et al. (1996) reported partition coefficients of aromatic hydrocarbons in human and rat fat tissue. Values for human and rat tissue partition coefficients of volatile anesthetics and related compounds, e.g., fluorinated alkanes, are scattered over various references (Eger and Eger, 1985; Coburn and Eger, 1986; Fassoulaki and Eger, 1986; Lerman et al., 1986, 1987; Eger, 1987; Strum and Eger, 1987; Yasuda et al., 1989; Taheri et al., 1993; Chortkoff et al., 1994; Eger et al., 1994; Liu et al., 1994; Fang et al., 1996, 1997a,b). Two compounds with extremely low water, oil, and tissue partition coefficients, perfluoropropane and perfluoropentane (Eger et al., 1994), were not included in Table 1. In general, reported partition coefficients have been determined in vitro by headspace gas chromatography at 37°C in a vial equilibration technique (Sato and Nakajima, 1979a) or by a modification of this method (Gargas et al., 1989). In the compilation of saline partition coefficients it was noted that in many cases partitioning in saline and in water is considered to be identical. Although $P_{\text{saline:air}}$ may be slightly lower than $P_{\text{water:air}}$ (Steward *et al.*, 1973; Lerman et al., 1983), they are considered equivalent here. When values of partition coefficients were available from multiple sources, the mean was calculated and used in this study.

Regression analysis. $P_{\text{tissue-air}}$ is described as a bilinear function of $P_{\text{olive oil:air}}$ and $P_{\text{saline-air}}$ (Droz, 1978) according to:

$$P_{\text{tissue:air}} = \alpha_0 P_{\text{olive oil:air}} + \alpha_s P_{\text{saline:air}} + c. \tag{1}$$

The coefficients α_o and α_s in Eq. (1) represent the tissue-specific contributions of the lipophilic and hydrophilic interactions to the solubility of the compounds in tissue. The constant *c* was included in the equation to avoid errors in the slope of the regression plane, which occurred in fitting the same equation with zero intercept. Although *c* has no specific physical meaning it may be required to compensate for systematic errors in tissue, oil, or saline partition coefficients. Tissue partitioning according to Eq. (1) was fitted using weighted bilinear regression; i.e., each tissue partition coefficient value was divided by its own value to ensure equal weights of individual compounds in the regressions. Estimated values of the regression coefficients α_o and α_s and of the constant *c* are reported with their correlation coefficients (R^2). Cross-correlation between fitted parameters was also monitored to judge the quality of the regressions and possible redundancy of parameters. To analyze the applicability of the estimated regression coefficients for the prediction of tissue partition coefficients, the ratio of predicted and observed values was plotted logarithmically against log $P_{\rm oiltair}$ and against log $P_{\rm saline.air}$, and the mean value and 2.5 and 97.5 percentiles were determined for each tissue. All regression analyses were performed using SigmaPlot 3.02 software (Jandel Scientific Software, SPSS Inc., Chicago, IL).

RESULTS

Oil and Saline Partition Coefficients

All values of $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ of VOCs, obtained from the literature and compiled in Table 1 (n = 137), are graphically presented in Fig. 1. From the clustering of points near the origin in Fig. 1 it is clear that many of the VOCs have relatively small $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ values. Compounds which are highly lipophilic or highly hydrophilic always have small corresponding $P_{\text{saline:air}}$ or $P_{\text{oil:air}}$, respectively. Only a few compounds, e.g., 2-butoxyethanol (Fig. 1, compound 112), combine intermediate lipophilicity and hydrophilicity. Due to the inverse relation between oil and saline partition coefficients, points are found in a region of the plane limited by a hyperbolic curve. Although water-soluble VOCs are underrepresented in the total set of data, the scatter of the data (see Fig. 1, inset) indicates that the correlation between oil and saline partition coefficients is small for the data compiled and that P_{oilair} and $P_{\text{saline:air}}$ are largely independent descriptors of VOC properties.

Human Tissue Partition Coefficients

Experimental values of partition coefficients of VOCs, available from the literature, in human blood (n = 109) and human fat, liver, brain, muscle, and kidney tissues (n = 28-41; see Table 1) were used in regression analysis. For each of the human tissues the relation between $P_{\text{tissue:air}}$ and the corresponding $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ was evaluated by bilinear regression according to Eq. (1). The regressions for $P_{\text{fat:air}}$, $P_{\text{brain:air}}$, and $P_{\text{blood:air}}$ are plotted in three-dimensional graphs in Fig. 2. For fat tissue and blood the slopes of the regression planes are mainly determined by α_0 and α_s , respectively. Although the slope of the regression plane of brain tissue partition coefficients is intermediate between those of blood and fat tissue partition coefficients, α_s appears to be the main regression coefficient for brain tissue partitioning. Predicted $P_{\text{tissue:air}}$ values for human muscle and kidney are within planes with an intermediate orientation similar to that for brain (not shown). The results of the regressions for human tissue partition coefficients (Table 2) show that, except for fat, tissue partitioning is mainly predicted by the $P_{\text{saline;air}}$, according to large values of α_s over small values of α_0 . For all tissues, except liver, the regressions yielded good correlation coefficients ($R^2 = 0.92 - 0.99$), and the crosscorrelation between the estimated values of the parameters α_s and α_0 was very small (<0.04). For human liver, only α_0 has been determined. Partition coefficients available for human liver were strongly biased toward more lipophilic compounds (see Table 1 and Fig. 3). Therefore, a reliable estimate of α_s

TABLE 1	Compilation of Reported and Predicted Partition Coefficients of VOCs in Olive Oil and Saline and in Human and Rat Tissues
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								H	uman P ₁	tissue: air					Ra	t P _{tissue:ai}	ц		
	Compound	$P_{ m oil:air}$	Ref	$P_{ m saline:air}$	Ref	Blood	Fat	Brain	Liver]	Muscle	Kidney	Ref	Blood	Fat	Brain	Liver	Muscle	Kidney	Ref
- 0 %	<i>n</i> -Ethane <i>n</i> -Butane <i>n</i> -Pentane	1.62 19.3 56.1	F F i/w/F	0.025 0.018 0.014	F F WF	0.06 0.18 0.38	7.3 15.2 39.6	0.98 1.33 2.2	0.84 (1.33 2.1	0.97 1.22 0.7	0.72 0.91 0.6	ж	$\begin{array}{c} 0.1037 \\ 0.2925 \\ 1.48 \end{array}$	10.4 20.9 42.7	0.11 1.06 3.04	2.43 2.88 3.83	0.32 0.49 0.86	0.18 1.89 5.47	8/t 8/t
04 v	n-Hexane	157 157	c/m/w/F	0.015	m/v/F	0.8	104	2	2.5	2	33.0		1.72	159	8.48	2.5	2.9	15.3 14.7	g/m/t
96	n-tueptane n-Octane n-Decane	1406 14400	c/F c/F F	0.0051	w V/F F	10.2 104	233 233 6443	16.5 289	25.8 404	8.6 203	8.2 159	1	7.53 17.3	819 844 8563	75.9 778	38.9	14.3 144	137 137 1402	$\frac{g/t}{t}$
×	Ethvlene	1.27	Q	0.090	Q	0.15	0.95	1.00	0.83	0.00	0.74	Q	1.25	10.2	0.14	2.48	0.37	0.20	
6	2,2-Dimethylbutane	71	М	0.0096	v	0.26	99	2.8	35	1	1.4	M	1.55	51.6	3.84	4.22	1.00	6.92	
10	2-Methyl-1,3-butadiene (isoprene)	8.81	m	0.21	m	0.28	10.5 87	1.20	1.04	1.14	0.87		1.87	72 70.6	0.65 5 57	3.12	2.04	1.03 10.0	ш
112	z-ivteury ipentane 3-Methylpentane	118	2 2	0.0019	~ ~	0.43	0/ 102	0.4 4.4	1 0	3.8	2.5 2.5	<i>4 4</i>	1.80	79.5	6.37	5.43	1.47	11.5 11.5	
13	3-Methylhexane	311	м	0.0062	V	1.3	277	10.2	10.6	10.8	7.3	м	2.85	194	16.8	10.5	3.40	30.3	
14 15	2,3,4-Trimethylpentane 2,2,4-Trimethylpentane	662 366	m	0.0024 0.0014	~ ~	4.80 1.6	303 170	14.2 8.26	19.3	10.2 5.06	7.97 4.72	ш	3.75 1.77	443 293	35.7 19.8	18.8	4.41 3.3	64.5 35.6	ш
16	Cyclopropane	11.8	D/D	0.21	D	0.55	×	1.4	0.6	0.4	0.4	D	1.42	16.4	0.81	2.85	0.57	1.32	
17	Cyclopentane	139	į	0.088	į	1.11	68.7	3.75	4.68	2.92	2.25		1.74	92.0	7.58	6.05	1.74	13.6	į
18	Cyclohexane	331	c/i/m/w	0.15	i/v	1.45	260	10.7	10.8	10.5	7.2	W	1.39	235	18.0	7.88	1.03	32.3	ш
19	Cycloheptane	2780	i	0.075	į	20.1	1249	56.6	78.6	39.9	31.3		5.2	1661	150	74.7	28.1	271	į
20	Methylcyclopentane	202	м	0.043	м	0.86	176	7.3	2.8	6	4.7	м	2.29	129	10.9	7.65	2.34	19.7	
21	Benzene	505	c/i/m/B	2.80	c/i/m/v/B	7.37	379.5	18	22.6	16.4	12.1	c/l/m/x/B	14.1	432.5	29.6	17	10.3	51.5	i/m/x
22	Toluene (methylbenzene)	1539	c/i/m/B	2.19	c/i/m/v/B	15.11	962	36.4	48.2	34.9	17.8	c/l/x/B	15.9	066	84.9	83.6	27.7	152	i/m/x
53	o-Xylene (1,2-dimethylbenzene)	6010	c/i/m/B	2.88	c/i/m/B	36.3	2460	122	169	86.2	67.9	c/i/m/x/B	44.3	2404	327	108	51.5	588	m/x
24	m-Xylene (1,3-dimethylbenzene)	4797	c/i/m/B	1.93	c/i/m/B	33.2	1919	97.6	135 (58.8 56.8	54.2	c/i/m/x/B	46 ; 6	2092	261 201	90.9	41.9	469	x/m
22	<i>p</i> -Xylene (1,4-dimethylbenzene)	4369	c/i/m/B	1.86	c/i/m/B	38.9	2019	89.0	123	62.8 22.8	49.5 52.2	c/i/m/x/B	41.3	1748	237	86	38.4	427	<i>u</i>
07	Ethylbenzene Streame (vinvilhenzene)	4621 4050	i/B B	1.78 3 02	i/B 	28.2	1764 3184	94.U	130	50.3 71 7	52.2 56 7	<i>x/B/G</i>	30.45 40.2	1556	251 177	60.3	26	451	5/1
1 00	1 3 5-Trimethylbenzene (mesitylene)	0880	<i>u m n</i>	1.23	<i>מעוווועום</i> ח	43.0	4423	100	- LLC	140	110	מאט מ	55.7	5878	535		1001	-043 063	Y/11
29	1,2,4-Trimethylbenzene (pseudocumene)	10200	u u	1.61	u u	59.1	4566	206	286	144	114	u u	57.7	6068	552	269	104	995	
30	1,2,3-Trimethylbenzene (hemimellitene)	10900	u	2.73	u	66.5	4879	220	306	155	122	u	62.6	6484	591	288	111	1064	
31	Propylbenzene	9775	В	1.30	В	47	4376	197	274	138	601	В	55.2	5816	529	258	66	953	
22	Allylbenzene	8049	В	3.55	В	50.9	3605	163	226	115	91	B	47.9	4791	438	215	83.5 22 - 5	787	
0.6	Uumene (Isopropylbenzene) Mathyletyrene (1. vinyl 3. mathylhenzene)	62150 14706	B 11	1.44	B ""	51 108	C8/7	170 071	C/1	50.5 806	07.0	В	30.1 107	3/01 11051	33/ 706	C01	03.0	000 1434	***
35	<i>p</i> -Methylsytrene (1-vinyl-4-methylbenzene)	13942	m	2.11	m	102	6239	281	391	197	155		234	11281	755	324	183	1359 1359	u u
36	Chloromethane	8.57	ш	0.88	ш	2.48	10.5	1.45	1.03	1.40	1.14	m	2.47	13.5	61.1	3.47	76.0	1.56	ш
37	Dichloromethane	148	c/l/m/C	6.71	c/m/v/C	8.15	85	9	7.2	4.8	5.8	c/l/m/C	19.4	120	13.6	14.2	7.92	19.9	ш
38	Chloroform (trichloromethane)	405	c/l/m/C/D	3.71	c/m/v/C/D	9.04	280	20	17	12	11	c/l/m/C/D	20.8	203	25.0	21.1	13.69	42.5	ш
39	Carbon tetrachloride	382	c/l/m/C	0.37	c/m/C	3.16	177	8.73	11.5 (5.43	5.04	c/m/C	4.52	359	20.9	14.2	4.57	37.5	ш
0 1	Chloroethane	38.9	m	1.15	m/D	2.35	24.1	2.15	1.88	1.92	1.58 2.80	m .	4.08	38.6	3.05	3.61	3.22	4.73	ш
4 ç	1,1-Dichloroethane	194	c/l/m/C	2.65	c/m/C	5.17	93.3 201	5.82	5.21	4.67	3.88	c/m/C	11.2	164	12.7	10.8	5.12	21.0	m
4 4 7 4	1, 2-Dicnloroetnane 1-1-1 Triahlamathana	1 41 245	C/1/m/C	0 87	C/m/C	CC.U2	204	14.1 02	15.1	C.11	10.1	c/m/c	30.4 5 76	344 262	53.3 10 1	33.1	2.5.4 2.1 <i>5</i>	21.4 21.2	ш
4 4 4	1.1.2. Tetrachloroethane	2109	c/l/m/C	15.0	c/m/C	37.67	951	48.8	59.8	36.2	29.9	c/m/C	285	1438	126	73.1	22.9	218	u u
45	1,1,1,2-Tetrachloroethane	3766	Um/C	4.5	m/C	30.3	1690	78.0	106	55.4	43.9	m/C	41.7	2148	207	88.2	39.5	370	m

	ш	<i>u u u u</i>	ш	ш		m m j/m	ш
976 653 489 22.73 19.3 19.3 19.3 19.3 19.6 60.6 60.6 60.6 95.7	293 3894 2641	105 139 5.82 27.7 16.9	208 143	1.55 0.01 0.48 1.00 1.26 0.41 2.23 3.09 0.41 17.8	66.3 85.6 65.7 62.0 33.2 38.9 24.8 143	42.3 267 4.71 62.8 2.68 2.1.3 3.09	16.3 7.00 24.2 25.4
101 72.4 75 2.1 2.05 3.52 10.1 12 2.08 2.08 2.04 11 12 10.6 10.6	34 406 275	40.5 45.6 2.26 4.21 4.12	28.9 29.1	1.44 0.29 0.84 0.84 0.86 0.33 0.71 0.71	8.66 11.2 8.69 7.89 4.19 4.79 3.10 15.3	11.1 55.6 2.46 25.4 1.23 4.46 0.82	5.28 1.97 13.2 8.66
196 260 369 116 115 3518 5518 3315 2272 238.9 238.9 238.9 238.9 287.4	86.1 1048 711	68.1 119 3.33 8.17 4.41	153 62.4	2.75 2.37 2.64 3.06 11.7 3.04 3.29	$\begin{array}{c} 21.6\\ 27.2\\ 27.2\\ 20.1\\ 11.7\\ 13.2\\ 9.24\\ 41.0\end{array}$	29.2 126 3.44 42.8 6.63 3.40	6.82 5.15 18.8 14.7
552 363 271 1.68 1.74 11.4 41.7 109 6.94 4.46 53.3 55.1 7.60 53.3	164 2163 1467	63.7 83.3 3.39 15.9 9.75	163 116	1.35 0.01 0.35 0.80 0.82 0.23 0.23 1.77 1.29	37.6 48.6 37.3 35.0 18.7 18.7 13.9 79.7	26.7 151 38.1 1.65 4.51 1.83	13.8 4.40 19.1 17.2
3767 4118 3321 220 28.6 68.6 554 118 118 118 118 118 101 2217 200	1277 23723 16095	792 1219 49.2 158	506 155	1.43 9.43 11.2 11.2 11.2 38.3 22.4 22.4 88.9	402 516 3388 378 208 159 883	325 1917 15.4 959 182 26.7	47.7 45.2 82.0 123
142 104 62.7 1.68 5 5 21.6 9.58 21.9 21.9 18.7 17.3 3.385 5.21 17.3 3.385 7.09	59.4 225 153	74.1 119 4.05 11.7 5.95	223 183	1.6 1.16 1.39 1.39 1.39 9.98 1.19 1.47 7.33	11.4 9.16 5.78 3.1 3.1 5.46 5.4	41.5 116 5.08 52.7 1.27 5.01	12.2 2.79 16.0 10.0
c/m/C m m/C m/C c/m/C/D m/C m/C m/C C C C	C C C	ш ш	m m	العم السم العم العم العم العم العم العم العم الع		m m b/c/Nq/n/D/H D	D D D z z
120 75.2 56.1 1.13 1.54 1.54 3.41 1.54 1.73 8.78 8.78 8.78 8.78 8.78 8.78 8.78 1.79 1.79	34.8 443 301	17.0 21.6 1.48 4.26 2.93	63.2 47.1	1.27 0.69 0.82 1.78 0.93 0.93 0.93 1.09 1.09	8.88 111.3 8.85 8.24 4.66 5.31 3.61 17.0	8.12 33.1 2.17 10.5 2.1 1.14	10 2 8.09 6.14
149 95.5 71.4 1.97 1.97 1.97 1.97 2.96 2.92 2.92 2.92 10.7 10.7 14.9	43.9 563 382	19.9 25.4 1.89 5.30 3.65	64.6 47.6	1.51 0.94 1.07 1.26 5.14 1.20 1.30 1.43 5.05	11.1 14.1 11.0 10.3 5.91 6.73 4.61 21.7	9.32 9.32 2.43 12.3 2.2 6.9 2.2	10 2 8.46 6.82
275 188 141 1.47 1.47 2.59 3.846 5.7.1 3.93 3.84 10.6 38.1	84.1 1119 759	27.6 36.5 2.36 8.41 5.38	30.5 18.7	$\begin{array}{c} 0.92\\ 0.79\\ 0.87\\ 1.03\\ 2.11\\ 0.91\\ 1.40\\ 1.64\\ 4.51\end{array}$	19.3 24.7 19.1 18.2 10.2 11.8 7.82 42.0	10.9 75.9 1.41 16.7 2.3 6.48 1.02	11 3 4.15 6.11
208 136 1.59 1.59 7.59 5.28 3.59 3.59 3.90 8.27 21.85 3.90 20.7	61.8 803 545	25.6 33.0 2.23 6.93 4.63	70.4 51.1	1.53 0.94 1.29 1.29 1.27 1.43 1.43 5.83 5.83	15.0 19.2 14.9 14.0 7.91 9.08 6.11 30.5	11.4 57.4 15.7 15.7 15.7 15.7 1.8 5.54 1.115	12.5 3.5 9.12 7.93
4382 2997 2748 17.5 25.4 887 905 567 567 55.5 163 163	1337 17852 12112	435 578 31.7 128 80.0	491 300	8.82 6.61 7.96 8.74 10.6 11.2 28.4 28.43 20.2 20.2 66.5	302 388 299 156 182 119 664	169 1206 262 34 168.4	50 40 92.1
114.2 50.3 52.4 1.16 0.87 9.03 12.33 9.01 12.33 9.01 12.33 9.73 7.4	30.4 423 201.4	19.9 24.8 7.08 2.57	187 154	$\begin{array}{c} 1.24 \\ 0.079 \\ 0.56 \\ 0.56 \\ 0.33 \\ 0.33 \\ 0.33 \\ 0.33 \\ 0.33 \\ 0.44 \\ 0.44 \\ 0.26 \\ 0.44 \end{array}$	6.85 8.92 6.90 6.12 3.52 2.20 11.0	10.4 52.7 2.96 29.2 11.5 2.51 0.6	12 2.6 17.7 11.7
с/m/C m m m/C c/m/C/D m/C m/n/C/D m/C m m/C m m/C	C C C	u m m	m m	Ξ 6- 6- 6- 6- 6- 6- 6- 6- 6- 6- Ξ	100 100 100 100 100 100 100	т т т <i>w</i> (s/A/D D	m/v/A/D D u
28.7 2.32 0.66 0.43 0.43 0.43 0.72 0.72 0.72 0.72 0.72 0.86 0.70 0.70	3.46 9.00 5.5	$14.4 \\ 17.3 \\ 0.44 \\ 1.44 \\ 1.08 $	127 98.3	$\begin{array}{c} 1.31\\ 0.0041\\ 0.23\\ 0.66\\ 2.49\\ 0.32\\ 9.20\\ 0.052\\ 0.16\\ 5.86\end{array}$	$\begin{array}{c} 2.29\\ 2.41\\ 1.80\\ 0.72\\ 0.43\\ 0.43\end{array}$	8.65 7.34 3.08 8.91 0.42 0.72 0.32	12.6 1.40 15.2 8.39
m/C m m m m m/C mm/C c/lm/C D mm/C c/lm/C D mm/C t/m V m m/C	C C C	m m m	m m	Ξ 4, 4, 4, 4, 4, 4, 4, 4, 4, 4,	the the the the the the the	m m m m/D V/V/D V/D	Д П п
9785 6689 5015 5015 24.4 750 184 750 750 109 69.9 77 977	2976 39920 37080	957 1276 56 272 164	$1062 \\ 640$	$\begin{array}{c} 4.76\\ 0.052\\ 3.02\\ 3.02\\ 1.71\\ 1.22\\ 1.22\\ 2.1.8\\ 3.0.4\\ 1.33\end{array}$	661 853 654 654 621 335 335 333 333 251 1470	361 2683 2569 24.0 213 29.0	$60.3 \\ 60.0 \\ 120 \\ 190$
1,1,2,2-Tetrachloroethane Pentachloroethane (perchloroethane) Chloroethane (perchloroethane) Thexachloroethylene (vinyl chloride) 1,1-Dichloroethylene <i>trans-1,2-Dichloroethylene</i> <i>trans-1,2-Dichloroethylene</i> <i>trans-1,2-Dichloroethylene</i> <i>trans-1,2-Dichloroethylene</i> <i>trans-1,2-Dichloroethylene</i> <i>1,1-Chloropropane</i> <i>2-Chloropropane</i> <i>2-Chloropropane</i> <i>1,2-Dichloropropane</i> <i>3-Chloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2-Dichloropropane</i> <i>1,2</i>	Chlorobenzene <i>o</i> -Dichlorobenzene (1,2-dichlorobenzene) <i>m</i> -Dichlorobenzene (1,3-dichlorobenzene)	Dibromomethane 1,2-Dibromoethane Bromoethylene (vinyl bromide) 1-Bromopropane (<i>n</i> -propyl bromide) 2-Bromopropane (isopropyl bromide)	1-Nitropropane 2-Nitropropane	CF ₂ H ₂ CF ₄ (perfluoromethane) CF ₃ CFH ₂ CF ₃ CFH ₂ CF ₃ CF ₂ CFH ₂ CF ₃ CH ₂ CFH ₂ CF ₃ CH ₂ CFH ₂ CF ₃ CF ₂ CF ₃ H ₂ CF ₃ CF ₂ CF ₃ H ₂ CF ₃ CFCF ₂ CF ₃ H CF ₃ CFHCFHCF ₃ CF ₃ CFHCFHCF ₃	Fluorobenzene <i>o</i> -Difluorobenzene (1,2-difluorobenzene) <i>p</i> -Difluorobenzene (1,4-difluorobenzene) 1,2,4-Tirifluorobenzene 1,3,5-Tirifluorobenzene Pentafluorobenzene Hexafluorobenzene Methylpentafluorobenzene	CBrCIH ₂ CBr ₂ CIH CCIFH ₂ CBrH ₂ CCIH ₂ CF ₃ CCIH ₂ Halothane (CF ₃ CBrCIH) Tefturane (CF ₃ CBrFH)	Diethyl ether Divinyl ether Methyl 1-butyl ether Ethyl 1-butyl ether
46 47 47 49 55 55 55 55 56 55 56 57 56 57 57 57 57 57 57 57 57 57 57 57 57 57	61 63	65 65 65 68 65 65	69 70	71 72 77 77 77 77 77 80 81 81	82 83 85 88 88 88 89	90 91 92 92 92 94 96 95 95 95 95 95 95 95 95 95 95 95 95 95	97 99 100

	Ref		j/m j		<i>а</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i> <i>д</i>	~~~~~~~~~	y/z y/z y/z y/z
	Kidney	42.6	9.24 2.01 5.36 96.0 5.18	10207 29669 19142 10354 6352	3190 2030 1240 875 1166 717 1160 1883	82.6 87.6 197 142 212 243 229 324	269 222 198 191 228 851
ú	Muscle	12.8	1.6 0.65 1.84 1.41 1.3.0 1.07	9487 27696 17819 9550 5498	3980 1710 11140 850 1100 1140 850 850 814 814	65.1 69.9 70.9 110 230 230	170 182 135 99 148 213
Rat P _{tissue:}	Liver	21.6	3.34 3.05 5.55 4.34 29.41 3.99	10802 31509 20282 10887 6335	3090 1730 1290 880 880 980 980 940 1750 1105	89 107 148 263 355 355 435	239 228 164 185 185 182 182
	Brain	28.1	2.04 1.20 5.88 3.28 24.4 3.01	10255 29872 19245 10362 6161	3470 1870 1220 868 1130 577 1140 614 1080 1385	70.1 80.0 99.9 88.9 138 165 221 240	267 213 172 122 139 194 526
	Fat	211	98.1 20.5 67.4 38.0 39.2	1467 3372 2542 2019 3844	193 226 402 720 160 900 2560 6977	99 153 514 303 1110 1520 2750 3730	78 200 395 372 524 505 4604
	Blood	14.1	1.79 1.47 2.38 2.19 25.02 1.78	11438 33398 21484 11507 6595	3335 2355 1340 880 1290 563 553 553 829 829	100 81.7 76.2 35.1 52 89.4 64.7 96.7	208 191 159 127 79 174 225
	Ref	п	b/c/l/q/r/D/H e/K d/j/l/q/r/D D d/lq/r/D l/E/H	00000	р р р р р р р р р р р р р р р р р р р	<u>a</u> a a a a a a a	k/B k/B B B B B B
	Kidney	9.16	1.75 0.4 3.4 1.3 20.35 1.39	4920 14354 9239 4958 2881	1355 940 686 371 533 558 558 448 448	44.8 31.2 27.4 31.6 53.6 53.6	146 74.0 51.8 49.0 82.0 141
P tissue:air	Muscle	10.2	$\begin{array}{c} 3.01\\ 0.94\\ 3.53\\ 2\\ 2\\ 1.94\end{array}$	4726 13782 8873 4766 2785	11309 850 678 678 343 502 <i>534</i> <i>534</i> <i>534</i> <i>534</i> <i>534</i>	43.6 30.9 18.2 36.2 50.9 65.3	151 73.4 51.8 51.5 81.8 81.8 162
Human	Liver	10.2	3.12 0.55 3.65 2 2.125 2.125	20.3 15.6 27.7 46.0 153	2.63 3.83 9.11 14.0 5.47 5.47 2.9.1 3.9.4 3.9.4	3.19 5.72 14.9 9.22 36.6 83.4 111	2.97 7.99 18.3 14.9 35.1 23.4 211
	Brain	12.2	2.23 0.54 2.77 2 24.12 1.225	4681 13642 8786 4726 2789	1252 917 652 362 362 542 542 542 542 533 544	43.7 31.6 19.9 36.4 68.5 88.9	134.5 76.5 54.4 58.5 85.9 206
	Fat	158	69.68 12 108.4 34 798.2 38.5	1239 2933 2167 1655 2970	231 215 215 231 231 388 388 388 126 <i>126</i> <i>522</i> <i>704</i> <i>704</i>	53.0 90.6 144 581 733 1327 1770	86 162 299 241 561 381 3371
	Blood	17.9	1.42 0.424 2.06 1.4 0.643	12393 32836 22093 14416 7965	2108.5 1352.5 1033.5 515.5 774 677 381 534 903	90.1 76.8 33.1 83.4 83.4 92.4	215.5 163.5 150 106 127 168 199
	Ref	п	m/s/A/D e s/A/D A/D s/A/D E	00000	8 (p/v 9 /v 9 /v 8 /p 8 /b	Q Q Q Q Q Q Q Q Q	v/y/B B y/B B B B/B
	$P_{ m saline:air}$	9.11	$\begin{array}{c} 0.58\\ 0.23\\ 0.74\\ 0.83\\ 0.37\\ 0.37\end{array}$	12280 35869 23069 12349 7051	$\begin{array}{c} 3113\\ 2104\\ 1023\\ 1023\\ 1023\\ 1023\\ 1033\\ 1360\\ 1360\\ 1080\\ 1080\end{array}$	108 53.0 34.1 25.9 32.7 22.6 24.0	316 239 166 114 87.0 181 146
	Ref	п	0/m/D 0/1 0/1 0/1 0/1 0/1 0/1 0/1 0/1 0/1 0/1	00000	d/8 d/8 d/8 d/8 d/8 d/8 d/8 d/8 d/8 d/8	d	y/B B y/B y/B y/B
	$P_{ m oil:air}$	337	90.0 97.5 97.5 951 50.1	696 529 962 5446	65.6 109 297 471 154 167 1010 1380 11600	85.7 176 503 301 1280 1620 3940	78.0 257 626 505 1226 808 808
	Compound	Ethyl <i>t</i> -pentyl ether	Isoflurane I-653 Enflurane Fluroxene Methoxy flurane Sevoflurane	1-Methoxy-2-propanol 2-Methoxyethanol 2-Ethoxyethanol 2-Isopropoxyethanol 2-Butoxyethanol	Methanol Ethanol I-Propanol 2-Propanol 2-Propanol 1-Butanol 3-Methyl-2-propanol (t-butanol) 1-Butanol 3-Methyl-1-butanol (isopentanol) 1-Pentanol 1-Pexanol	Methyl acetate Ethyl acetate <i>n</i> -Propyl acetate Isopropyl acetate Isobutyl acetate <i>n</i> -Butyl acetate Isopentyl acetate <i>n</i> -Pentyl acetate	Acetone Methyl ethyl ketone Diethyl ketone Methyl <i>n</i> -propyl ketone Methyl <i>n</i> -butyl ketone Methyl <i>n</i> -butyl ketone Methyl <i>n</i> -pentyl ketone
		101	$102 \\ 104 \\ 105 \\ 107 \\ 107 \\ 107 \\ 107 \\ 107 \\ 102 $	$^{108}_{110}$	$\begin{array}{c} 1113\\1116\\1117\\1117\\1119\\1119\\1120\\1221\\1221\end{array}$	123 124 125 126 128 128 128 129	131 132 132 133 134 135 137

Eger et al. (1994); "Fang et al. (1973), "Fang et al. (1997b); 'Fang et al. (1997b); 'Fassoulaki and Eger (1986); 'Fiserova-Bergerova and Diaz (1986); 'Fiserova-Bergerova et al. (1984); "Gargas et

al. (1989); "Järnberg and Johanson (1995); "Johanson and Dynésius (1988); "Kaneko et al. (1994); "Lerman et al. (1986); "Lerman et al. (1985); "Lerman et al. (1994); "Sinhén and Nakajima (1979a); ^cSato and Nakajima (1979b); ^bSteward *et al.* (1973); ^fStrum and Eger (1987); ^fTaheri *et al.* (1993); ^cTardif *et al.* (1997); ^{fI}Y asuda *et al.* (1989). In case of multiple sources, the Johanson (1995); "Paterson and Mackay (1989); "Perbellini et al. (1985); "Pierce et al. (1996); "Poulin and Krishnan (1996a); "Poulin and Krishnan (1996b); "Renzi and Waud (1977); "Sato and

mean of the reported values is tabulated. Compound numbering is maintained throughout this paper.

TABLE 1—Continued

210



FIG. 1. Scatter plot of P_{oitlair} vs $P_{\text{saline-air}}$ for all compounds presented in Table 1 (n = 137). The numbered points refer to specific compounds in Table 1. The inset shows the scatter of data points in a log-log presentation.

could not be obtained and a linear regression was performed on $P_{\text{liver,air}}$ and P_{oiltair} values to obtain estimates of α_o and c for human liver ($R^2 = 0.88$). With the exception of fat tissue, the estimated values of the intercept c were all smaller than 1. The intercepts showed large coefficients of variation and a large cross-correlation with the other regression parameters (≤ 0.22), illustrative of the minor contribution of this parameter to tissue partitioning.

The equations fitted to tissue partitioning of VOCs have been internally validated from plots of the logarithm of the ratio of predicted and observed values. Figure 3 is a double logarithmic plot against $P_{\text{oil:air}}$ and against $P_{\text{saline:air}}$. For all tissues, the data points are scattered around zero and their

distributions do not deviate from normal (p > 0.10). The means of the logarithmic ratios do not differ from 0 statistically, with the exception of blood and fat (p = 0.03 and 0.02, respectively). These deviations are caused by a few outliers, e.g., ethylene for fat (Fig. 3, compound 8). In order to assess the reliability of the predicted values, the 2.5 and 97.5 percentiles of the ratios of predicted and reported partition coefficients were calculated for each tissue. The resulting 95% confidence range is indicated in each panel of Fig. 3. Tissue concentrations were calculated from the quotient of predicted and from the quotient of experimental $P_{\text{tissue:air}}$ and $P_{\text{blood:air}}$. The tissue concentrations predicted by the model are within a factor of 4.0 from the tissue concentrations calculated from experimental data for 95% of the compounds for human brain, muscle, kidney, and fat tissue. The results indicate that the partitioning of VOCs in human tissues can be calculated on the basis of saline and olive oil partitioning according to Eq. (1) with a good predictive power.

Equation (1) was also applied to data on the partitioning of four terpenes (Falk *et al.*, 1990) and four gases (Steward *et al.*, 1973) in human blood. Both chemical classes are not included in the data of Table 1. The ratios of predicted versus experimental $P_{\text{blood:air}}$ were 1.4 for α -pinene, 1.4 for β -pinene, 1.1 for 3-carene, and 1.0 for limonene. For the gases, the ratios of predicted versus experimental $P_{\text{blood:air}}$ were 1.3 for Kr, 0.9 for Xe, 2.9 for nitrogen, and 1.0 for nitrous oxide. These results provide support for a more general applicability of Eq. (1) to blood partitioning of volatile organic compounds. Partition coefficients of the gases in other human tissues are very small (Steward *et al.*, 1973) and are in the same order of magnitude as the fitted intercept *c* (Table 2). This results in overestimation of the human tissue partition coefficients for gases.



FIG. 2. Three-dimensional representation of the bilinear regression between P_{oiltair} , $P_{\text{salineair}}$, and reported values of human P_{bolocair} , $P_{\text{brain-air}}$, and P_{fatair} according to Eq. (1). The grids represent the fitted planes of predicted partition coefficients and the dots represent experimental values. Identically spaced grids are given in the horizontal and vertical planes. Note that the orientation of the plane fitted for brain tissue partitioning is intermediate between that fitted for blood and fat tissue partitioning. Vertical lines are drawn from the horizontal plane to indicate the position of the data points.

	Results of BL	linear Regression	is for the Partition	and of vocs in	ito Human Tissue	es Filled Accor	aing to Eq. (1)	
Tissue	n	$lpha_{ m o}$	CV (%)	$lpha_{ m s}$	CV (%)	С	CV (%)	R^2
Blood	109	0.0072	18	0.898	2	0.03	2094	0.99
Fat	41	0.447	6	0.075	33	6.59	88	0.92
Brain	35	0.020	16	0.380	3	0.94	69	0.98
Liver	28	0.028	8	nd	nd	0.79	47	0.88
Muscle	35	0.014	20	0.384	3	0.94	64	0.99
Kidney	34	0.011	22	0.400	3	0.69	77	0.98

 TABLE 2

 Results of Bilinear Regressions for the Partitioning of VOCs into Human Tissues Fitted According to Eq. (1)

Note. The number of compounds used in the regression (*n*), the regression coefficients α_o and α_s , and the constant *c* together with their coefficients of variation (CV) are tabulated. The correlation coefficients of the regressions (R^2) are also indicated. nd, not determined.



FIG. 3. Double-log representation of the ratio of predicted and experimentally determined $P_{\text{tissue-air}}$ against P_{oilhair} and against $P_{\text{saline-air}}$ for human blood, fat, brain, liver, muscle, and kidney as indicated. Results are presented in two dimensions for a clear insight into the distribution of all points relative to the fitted regression planes. The 2.5 and 97.5 percentiles are drawn (dashed lines) and the values of the percentiles are indicated at the right for each tissue. Outliers are identified by their compound numbers.

Rat Tissue Partition Coefficients

For rat blood, fat, liver, and muscle, large numbers (n =76-92) of partition coefficients of VOCs are available from the literature (Table 1). However, data for rat brain (n = 19)and kidney (n = 16) appear to be less abundant. The regression planes obtained by fitting Eq. (1) to the data for P_{fatair} , $P_{\text{brain:air}}$, and $P_{\text{blood:air}}$ are plotted in three-dimensional graphs in Fig. 4. Qualitatively, the results are similar to those obtained for human tissue partition coefficients (see Fig. 2). The slopes of the planes describing fat tissue and blood partitioning are determined mainly by α_{o} and α_{s} , respectively. For brain and other tissues intermediate slopes with α_s as the main regression coefficient were obtained. The results of the regressions are summarized in Table 3. The quality of the regressions for rat tissues was not as good as that for human tissues, as indicated by slightly lower correlation coefficients ($R^2 = 0.82 - 0.93$). The cross-correlation between the fitted parameters α_0 and α_s was <0.03. The estimated values of the intercept were generally small, with the exception of the value for rat fat, which was estimated to be 9.4 and the coefficients of cross-correlation between the values estimated for the intercept and for the two other regression parameters ranged from 0.06 to 0.11. For rat kidney and brain, the intercept could not be determined reliably, because of the lack of data on VOCs with small $P_{\text{oil:air}}$ and $P_{\text{saline:air}}$ values (see Table 1 and Fig. 5). Despite the qualitative resemblance of the regressions of rat and human data, Table 3 shows a marked quantitative difference in the estimated values of α_s . Estimates of α_s for rat brain, liver, muscle, and kidney are all in a narrow range and are approximately twofold the corresponding values of α_s obtained for human tissues (see Table 2). Double logarithmic plots of the ratio of predicted and observed values show that the data points are scattered around zero for all tissues (Fig. 5), and their distributions do not deviate from normal (p > 0.10), except for rat fat (p =(0.03). The means of the logarithmic ratios do not differ from 0 statistically with the exception of blood (p < 0.0001), fat (p = 0.01), and muscle (p = 0.02). These deviations are caused by outliers, e.g., n-ethane, n-butane, and n-decane for blood (compounds 1, 2, and 7, respectively). Outliers for fat and muscle tissues are 1-nitropropane, 2-nitropropane (com-



FIG. 4. Three-dimensional representation of the bilinear regression between $P_{\text{oit-air}}$, $P_{\text{saline-air}}$, and reported values of rat $P_{\text{bloot-air}}$, $P_{\text{brain-air}}$, and P_{fatair} according to Eq. (1). The grids represent the fitted planes of predicted partition coefficients and the dots represent experimental values. Identically spaced grids are given in the horizontal and vertical planes. The orientation of the plane fitted for brain tissue partitioning is intermediate between that fitted for blood and fat tissue partitioning. Note that the regression of rat $P_{\text{brain-air}}$ depends more steeply on $P_{\text{saline-air}}$ than the regression of human $P_{\text{brain-air}}$ (see Fig. 2). Vertical lines are drawn from the horizontal plane to indicate the position of the data points.

pounds 69 and 70), and several esters (compounds 127–130). The 2.5 and 97.5 percentiles of the ratios of predicted and reported partition coefficients were calculated. The resulting 95% confidence range is indicated in each panel of Fig. 5. Predicted concentrations in liver, muscle, kidney, and fat tissue were within a factor of 5.0 from concentrations calculated from experimental values for 95% of the compounds.

DISCUSSION

The present results show that it is possible to predict tissue partition coefficients of VOCs from a simple linear combination of olive oil and saline partitioning, which is very similar to the method used before to predict partitioning of anesthetics in human tissues (Droz, 1978). Regression coefficients are estimated from data on a large set of VOCs, selected only by availability in the literature and not by the chemical nature of the compounds. The good quality of the regressions and the reliability of the predictions made by the model, particularly for human tissues, show that this approach is applicable to volatile compounds in general. In addition, the evaluation of VOC partitioning in six different human tissues and in the homologous rat tissues allows for making comparisons between tissues and between the two species.

Other studies describing empirical relations between tissue, oil, and saline partition coefficients, e.g., for chlorinated alkanes in human blood (Sato and Nakajima, 1979b) and for VOCs in several rat tissues (Gargas *et al.*, 1989), have performed regressions on logarithmically transformed data using equations for tissue partitioning similar to Eq. (1). However, a major problem with the use of logarithmic equations of the type log $P_{\text{tissue:air}} = a \log P_{\text{oil:air}} + b \log P_{\text{saline:air}}$ is that the tissue partition coefficient is implicitly assumed to be proportional to the product of water and oil partition coefficients (i.e., $P_{\text{tissue:air}}$ $\propto P_{\text{oil:air}} * P_{\text{saline:air}}$, or $P_{\text{tissue:saline}} \propto P_{\text{oil:air}}$). The meaning of such

 TABLE 3

 Results of Bilinear Regressions for the Partitioning of VOCs into Rat Tissues Fitted According to Eq. (1)

Tissue	п	$lpha_{ m o}$	CV (%)	$lpha_{ m s}$	CV (%)	с	CV (%)	R^2
Blood	92	0.0054	19	0.931	4	1.16	87	0.93
Fat	76	0.594	4	0.085	46	9.40	116	0.86
Brain	19	0.054	27	0.832	6	nd	nd	0.90
Liver	77	0.026	11	0.878	5	2.36	96	0.92
Muscle	76	0.010	17	0.772	5	0.29	532	0.82
Kidney	16	0.097	21	0.826	6	nd	nd	0.91

Note. The number of compounds used in the regression (*n*), the regression coefficients α_0 and α_s , and the constant *c* together with their coefficients of variation (CV) are tabulated. The correlation coefficients of the regressions (R^2) are also indicated. nd, not determined.



FIG. 5. Double-log representation of the ratio of predicted and experimentally determined $P_{\text{tissue-air}}$ against $P_{\text{oil-air}}$ and against $P_{\text{saline-air}}$ for rat blood, fat, brain, liver, muscle, and kidney as indicated. Results are presented in two dimensions for a clear insight into the distribution of all points relative to the fitted regression planes. The 2.5 and 97.5 percentiles are drawn (dashed lines) and the values of the percentiles are indicated at the right for each tissue. Outliers are identified by their compound numbers. *n*-Ethane (compound 1), which lies beyond the borders of the graph, has been indicated by arrows together with the value of the ratio (11.5).

a proportional relation between tissue:saline and oil:air partition coefficients is not clear. Fitting power functions to the nontransformed data of Table 1 by nonlinear regression, as an alternative for logarithmic transformation, did not result in a significant improvement of the fits, generally yielded exponent values close to 1, and caused a marked increase in crosscorrelation between the fitted parameters, indicating that the addition of the exponents in the equation caused redundancy in the parameters.

Inclusion of the constant term c in Eq. (1) did not improve the linear regressions and did not cause significant changes in α_o and α_s . With the exception of fat tissue, the magnitude of the fitted constant was consistently small and its CV value was consistently large (up to over 1000%). The constant c in Eq. (1) appeared to be redundant to a certain extent, because it caused one order of magnitude increase in the cross-correlation coefficients of the fitted parameters and failed to improve the correlation coefficients of the fits. However, neglecting the constant resulted in the underestimation of the mean tissue partition coefficients by up to 20%. Although the physical meaning of the constant is unclear, it may compensate for small systematic deviations in tissue partitioning or for small systematic errors in the values of reported partition coefficients. The relatively large values estimated for the intercepts for human and rat fat (Tables 2 and 3) cannot be explained at present. It should be noted, however, that the fat partition coefficients of compounds included in the regressions are generally large.

For the majority of compounds (>80%) $P_{\text{saline:air}}$ values were collected from the literature. For the remaining compounds either $P_{\text{water:air}}$ values were reported (~10%) or it is unclear whether the published values represent $P_{\text{water:air}}$ or $P_{\text{saline:air}}$ (~10%). For volatile anesthetics, it has been shown that $P_{\text{saline:air}} = 0.87-0.97 P_{\text{water:air}}$ (Steward *et al.*, 1973; Renzi and Waud, 1977; Halliday *et al.*, 1977; Lerman *et al.*, 1983). The magnitude of the difference is similar to that of the experimental error in the determination of the partition coefficient (e.g., see Sato and Nakajima, 1979a,b). The small difference between $P_{\text{water:air}}$ and $P_{\text{saline:air}}$ and the minority of compounds to which this applies indicates that the error introduced by ignoring the difference will not affect the results significantly.

In an alternative approach, VOC tissue partitioning is supposed to involve partitioning into specific tissue components, i.e., water, phospholipids, and neutral lipids (Poulin and Krishnan, 1995a,b, 1996a,b). The solubility of compounds in the various tissue fractions, estimated from partition coefficients in *n*-octanol or vegetable oil and in water, is used to calculate tissue partitioning. Reversible interactions with proteins or hemoglobin included in a study on blood partitioning are supposed to be due to the presence of hydrophobic binding pockets and the contribution of covalent interactions to VOC partitioning is considered negligible (Poulin and Krishnan, 1996b). The more detailed approach may contribute to unraveling processes and mechanisms involved in tissue partitioning. However, the more complex equations used (Poulin and Krishnan, 1995a) can be reduced to the simple form of Eq. (1). This shows that, for the prediction of VOC tissue partition coefficients in practice, prior knowledge of tissue composition is not required. Advantages of the simple approach are that it considers partitioning in tissues as a whole and that very few assumptions are required for the determination of regression coefficients from experimental data. A disadvantage is that the dimensionless regression coefficients α_0 and α_s , despite their relation to the relative proportions of hydrophilic and hydrophobic tissue constituents, are not associated with a specific process or mechanism.

The results (Tables 2 and 3) demonstrate that the regression of VOC blood partition coefficients is mainly determined by α_s , and the regression of VOC fat tissue partition coefficients is mainly determined by α_0 . The marked differences in the regression coefficients for blood and fat are coherent with differences in tissue water and lipid contents. The regression of VOC partitioning in other tissues is intermediate but always with a major component determined by α_s . Differences in estimated regression coefficients for brain, liver, muscle, and kidney are less prominent than those for blood and fat. The values obtained for α_s are remarkably constant within species and are an order of magnitude larger than values of α_0 , which are more variable but also less accurate as indicated by their larger CV values. It is concluded that, within species, the partitioning of the less lipophilic VOCs into brain, liver, muscle, and kidney is little tissue-dependent. Partitioning of lipophilic VOCs in these tissues will be moderately tissuedependent within species, as indicated by 3- to 10-fold differences in estimated values of α_0 for human and rat tissues, respectively.

A comparison between the two species shows a large, consistent difference in α_s values, which range between 0.380 and 0.400 for human brain, liver, muscle, and kidney and between 0.772 and 0.878 for the corresponding rat tissues. Literature sources did not always reveal whether fresh or frozen tissues were used for the determination of partition coefficients. Since fresh and frozen tissues were used for both species, and since systematic differences in α_o values are not observed between species, it seems highly unlikely that the difference in α_s is caused by differences in the processing of rat and human tissues. This indicates that the factor 2 larger α_s values for rat appear to reflect a genuine difference in rat and human tissue partitioning. Consequently, caution should be exercised in exchanging tissue partition coefficients between PBPK models for different species.

It has been reported before that partition coefficients for rat blood are higher than those for human blood (Gargas et al., 1989; Lam et al., 1990; Kaneko et al., 1994). When the bilinear model (Eq. (1)) is applied to the data of Gargas et al. (1989), who measured VOC partitioning in rat and human blood for 35 compounds in parallel, the species difference between the experimental values is reproduced. For 29 additional compounds the differences in rat and human blood partition coefficients, collected from various literature sources, are less pronounced and amount on average to a factor of 1.3 compared to 1.7 for the data of Gargas et al. (1989). A notable difference is that the compounds investigated by Gargas et al. (1989) are less hydrophilic than the additional 29 compounds. Apart from the data mentioned already in this section, we used data on 45 additional compounds for estimating the regression coefficients for human blood partitioning and on 28 additional, nonoverlapping compounds for estimating the regression coefficients for rat blood partitioning. In the final results of our bilinear regressions the species difference reported for the more restricted and more homogeneous set of data of Gargas et al. (1989) is no longer apparent. The variation in the very large

dataset used for the regressions in the present study may obscure more subtle species differences in tissue partition coefficients. However, this does not detract from the point that the present approach shows that it is possible to obtain a fairly accurate prediction of tissue concentrations of VOCs without prior knowledge of particular properties of the chemical class and without prior knowledge on tissue composition.

The reliability of the predictions, as assessed from the ratio of predicted and experimental values, indicates that it is possible to predict the concentration of VOCs in human tissues with an accuracy of a factor of 4.0 and in rat tissues with an accuracy of a factor of 5.0. Thus, the concentrations of 95% of the VOCs considered in this study are predicted with an accuracy that appears to be sufficiently high to be used in human risk assessment. Reliable prediction of tissue partition coefficients will enable systematic PBPK modeling of exposurerelated brain concentrations of VOCs, which is essential to obtain insight into the relation between brain concentrations and adverse neurotoxic effects.

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REFERENCES

- Andersen, M. E. (1991). Physiological modeling of organic compounds. Ann. Occup. Hyg. 35, 309–321.
- Chortkoff, B. S., Laster, M. J., Koblin, D. D., Taheri, S., Eger, E. I., II, and Halsey, M. J. (1994). Pharmacokinetics do not explain the absence of an anesthetic effect of perfluoropropane or perfluoropentane. *Anesth. Analg.* 79, 234–237.
- Coburn, C. M., and Eger, E. I., II (1986). The partial pressure of isoflurane or halothane does not affect their solubility in rabbit blood or brain or human brain: Inhaled anesthetics obey Henry's law. *Anesth. Analg.* 65, 960–962.
- Droz, P. O. (1978). Contribution à la recherche d'indices biologiques d'exposition aux solvants. Détermination de leurs coefficients de partage et étude de leur comportement dans l'organisme à l'aide de modéles de simulation. Thesis. Neuchâtel University, Switzerland.
- Eger, R. R., and Eger, E. I., II (1985). Effect of temperature and age on solubility of enflurane, halothane, isoflurane, and methoxyflurane in human blood. *Anesth. Analg.* **64**, 640–642.
- Eger, E. I., II (1987). Partition coefficients of I-653 in human blood, saline, and olive oil. Anesth. Analg. 66, 971–973.
- Eger, E. I., II, Liu, J., Koblin, D. D., Laster, M. J., Taheri, S., Halsey, M. J., Ionescu, P., Chortkoff, B. S., and Hudlicky, T. (1994). Molecular properties of the "ideal" inhaled anesthetic: Studies of fluorinated methanes, ethanes, propanes, and butanes. *Anesth. Analg.* **79**, 245–251.
- Falk, A., Gullstrand, E., Lof, A., and Wigaeus-Hjelm, E. (1990). Liquid/air partition coefficients of four terpenes. Br. J. Ind. Med. 47, 62–64.
- Fang, Z., Ionescu, P., Chortkoff, B. S., Kandel, L., Sonner, J., Laster, M. J., and Eger, E. I., II (1997a). Anesthetic potencies of *n*-alkanols: Results of additivity and solubility studies suggests a mechanism of action similar to that for conventional inhaled anesthetics. *Anesth. Analg.* 84, 1042–1048.
- Fang, Z., Laster, M. J., Gong, D., Ionescu, P., Koblin, D. D., Sonner, J., Eger,

E. I., II, and Halsey, M. J. (1997b). Convulsant activity of non-anesthetic gas combinations. *Anesth. Analg.* **84**, 634–640.

- Fang, Z., Sonner, J., Laster, M. J., Ionescu, P., Kandel, L., Koblin, D. D., Eger, E. I., II, and Halsey, M. J. (1996). Anesthetic and convulsant properties of aromatic compounds and cycloalkanes: Implications for mechanisms of narcosis. *Anesth. Analg.* 83, 1097–1104.
- Fassoulaki, A., and Eger, E. I., II (1986). Starvation increases the solubility of volatile anaesthetics in rat liver. *Br. J. Anaesth.* 58, 327–329.
- Fiserova-Bergerova, V., and Diaz, M. L. (1986). Determination and prediction of tissue-gas partition coefficients. *Int. Arch. Occup. Environ. Health* 58, 75–87.
- Fiserova-Bergerova, V., Tichy, M., and Di Carlo, F. J. (1984). Effects of biosolubility on pulmonary uptake and disposition of gases and vapors of lipophilic chemicals. *Drug Metab. Rev.* 15, 1033–1070.
- Gargas, M. L., Burgess, R. J., Voisard, D. E., Cason, G. H., and Andersen, M. E. (1989). Partition coefficients of low-molecular weight volatile chemicals in various liquids and tissues. *Toxicol. Appl. Pharmacol.* 98, 87–99.
- Gargas, M. L., Medinsky, M. A., and Andersen, M. E. (1995). Pharmacokinetic modeling approaches for describing the uptake, systemic distribution, and disposition of inhaled chemicals. *Crit. Rev. Toxicol.* 25, 237–254.
- Halliday, M. M., MacDonald, I., and MacGregor, M. H. G. (1977). Gas chromatographic determination of Ostwald solubility coefficients for cyclopropane, halothane and trichlorethene (trichloroethylene). *Br. J. Anaesth.* **49**, 413–417.
- Järnberg, J., and Johanson, G. (1995). Liquid/air partition coefficients of the trimethyl-benzenes. *Toxicol. Ind. Health* 11, 81–88.
- Johanson, G., and Dynésius, B. (1988). Liquid/air partition coefficients of six commonly used glycol ethers. Br. J. Ind. Med. 45, 561–564.
- Kaneko, T., Wang, P. Y., and Sato, A. (1994). Partition coefficients of some acetate esters and alcohols in water, blood, olive oil and tissues. *Occup. Environ. Med.* 51, 68–72.
- Krishnan, K., and Andersen, M. E. (1994). Physiologically based pharmacokinetic modeling in toxicology. In *Principles and Methods of Toxicology*. 3rd ed. (A. W. Hayes, Ed.), pp. 149–188. Raven Press, New York.
- Lam, C. W., Galen, T. J., Boyd, J. F., and Pierson, D. L. (1990). Mechanism of transport and distribution of organic solvents in blood. *Toxicol. Appl. Pharmacol.* **104**, 117–129.
- Lerman, J., Gregory, G. A., and Eger, E. I., II (1987). Effects of anaesthesia and surgery on the solubility of volatile anaesthetics in blood. *Can. J. Anaesth.* 34, 14–16.
- Lerman, J., Schmitt-Bantel, B. I., Gregory, G. A., Willis, M. M., and Eger, E. I., II (1986). Effect of age on the solubility of volatile anesthetics in human tissues. *Anesthesiology* 65, 307–311.
- Lerman, J., Willis, M. M., Gregory, G. A., and Eger, E. I., II (1983). Osmolarity determines the solubility of anesthetics in aqueous solutions at 37°C. *Anesthesiology* 59, 554–558.
- Liu, J., Laster, M. J., Taheri, S., Eger, E. I., II, Chortkoff, B. S., and Halsey, M. J. (1994). Effect of *n*-alkane kinetics in rats on potency estimation and the Meyer-Overton hypothesis. *Anesth. Analg.* **79**, 1049–1055.
- Mikkelsen, S. (1997). Epidemiological update on solvent neurotoxicity. *Environ. Res.* 73, 101–112.

- Nihlén, L. A., and Johanson, G. (1995). Liquid/air partition coefficients of methyl and ethyl *t*-butyl ethers, *t*-amyl methyl ether, and *t*-butyl alcohol. *J. Expo. Anal. Environ. Epidemiol.* 5, 573–582.
- Paterson, S., and Mackay, D. (1989). Correlation of tissue, blood and air partition coefficients of volatile organic chemicals. *Br. J. Ind. Med.* 46, 321–328.
- Perbellini, L., Brugnone, F., Caretta, D., and Maranelli, G. (1985). Partition coefficients of some industrial aliphatic hydrocarbons (C5–C7) in blood and human tissues. Br. J. Ind. Med. 42, 162–167.
- Pierce, C. H., Dills, R. L., Silvey, G. W., and Kalman, D. A. (1996). Partition coefficients between human blood or adipose tissue and air for aromatic solvents. *Scand. J. Work Environ. Health* **22**, 112–118.
- Poulin, P., and Krishnan, K. (1995a). A biologically-based algorithm for predicting human tissue:blood partition coefficients of organic chemicals. *Hum. Exp. Toxicol.* 14, 273–280.
- Poulin, P., and Krishnan, K. (1995b). An algorithm for predicting tissue:blood partition coefficients of organic chemicals from *n*-octanol:water partition coefficient data. J. Toxicol. Environ. Health 46, 101–113.
- Poulin, P., and Krishnan, K. (1996a). A tissue composition-based algorithm for predicting tissue:air partition coefficients or organic chemicals. *Toxicol. Appl. Pharmacol.* **136**, 126–130.
- Poulin, P., and Krishnan, K. (1996b). A mechanistic algorithm for predicting blood:air partition coefficients of organic chemicals with the consideration of reversible binding in hemoglobin. *Toxicol. Appl. Pharmacol.* **136**, 131– 137.
- Renzi, F., and Waud, B. E. (1977). Partition coefficients of volatile anesthetics in Krebs' solution. *Anesthesiology* 47, 62–63.
- Sato, A., and Nakajima, T. (1979a). Partition coefficients of some aromatic hydrocarbons and ketones in water, blood and oil. *Br. J. Ind. Med.* 36, 231–234.
- Sato, A., and Nakajima, T. (1979b). A structure-activity relationship of some chlorinated hydrocarbons. Arch. Environ. Health 34, 69–75.
- Steward, A., Allott, P. R., Cowles, A. L., and Mapelson, W. W. (1973). Solubility coefficients for inhaled anaesthetics for water, oil and biological media. *Br. J. Anaesth.* 45, 282–293.
- Strum, D. P., and Eger, E. I., II (1987). Partition coefficients for sevoflurane in human blood, saline, and olive oil. Anesth. Analg. 66, 654–656.
- Taheri, S., Laster, M. J., Liu, J., Eger, E. I., II, Halsey, M. J., and Koblin, D. D. (1993). Anesthesia by *n*-alkanes not consistent with the Meyer-Overton hypothesis: Determinations of the solubilities of alkanes in saline and various lipids. *Anesth. Analg.* 77, 7–11.
- Tardif, R., Charest-Tardif, G., Brodeur, J., and Krishnan, K. (1997). Physiologically based pharmacokinetic modeling of a ternary mixture of alkylbenzenes in rats and humans. *Toxicol. Appl. Pharmacol.* 144, 120–134.
- Yasuda, N., Targ, A. G., and Eger, E. I., II (1989). Solubility of I-563, sevoflurane, isoflurane, and halothane in human tissues. *Anesth. Analg.* 69, 370–373.
- White, R. F., and Proctor, S. P. (1997). Solvents and neurotoxicity. *Lancet* 349, 1239–1243.