

ENERGY TRANSFER PHENOMENA IN LANTHANIDE METABORATES (LnB_3O_6)

HAO ZHIRAN* and G. BLASSE

Physical Laboratory, State University, P.O. Box 80.000, 3508 TA Utrecht,
The Netherlands

It is shown that GdB_3O_6 is an excellent host lattice to obtain efficient photoluminescent materials. For this purpose we need a sensitizer (S) of the Gd^{3+} sublattice and an activator (A). Energy transport from S to A occurs via the Gd^{3+} sublattice. We studied $S = \text{Sb}^{3+}, \text{Bi}^{3+}, \text{Ce}^{3+}$ and $A = \text{Eu}^{3+}, \text{Tb}^{3+}$. Results confirm the transport process $S \rightarrow \text{Gd}(\rightarrow \text{Gd})_n \rightarrow A$.

1. INTRODUCTION

In recent years there is a growing interest in gadolinium compounds as host lattices for sensitizers and activators to obtain in this way new and efficient photoluminescent materials. In fact one of us proposed that a luminescent centre in a Gd^{3+} compound can transfer its excitation energy to the Gd^{3+} sublattice and that an activator can trap this energy when it migrates among the Gd^{3+} ions^{1,2}. De Hair *et al.* applied this principle to prepare efficient phosphors³⁻⁵. In this laboratory the work was continued with GdF_3 (ref.6) and $\text{GdMgB}_5\text{O}_{10}$ (ref.7).

Here we report results on GdB_3O_6 as a host lattice. We used $\text{Sb}^{3+}, \text{Bi}^{3+}$ and Ce^{3+} as possible sensitizers of the Gd^{3+} sublattice and Tb^{3+} and Eu^{3+} as activators. The crystal structure of the compounds LnB_3O_6 ($\text{Ln} = \text{La} - \text{Tb}$) has been described by Abdullaev *et al.*⁸. In this paper we report on double-doped borates; the single-doped were reported elsewhere⁹.

2. EXPERIMENTAL

The experimental techniques were described in ref.9.

3. RESULTS AND DISCUSSION

3.1. $\text{GdB}_3\text{O}_6-\text{Sb}^{3+}, \text{Tb}^{3+}$

Results at room temperature are simple, since the Sb^{3+} emission is quenched at 300 K. The excitation spectrum of the Tb^{3+} emission shows only Tb^{3+} and Gd^{3+} lines. At 4.2 K a weak band at 260 nm appears. This is the Sb^{3+} excitation band.

* On leave of absence from Changchun Institute of Physics, Chinese Academy of Sciences, Changchun, The Peoples Republic of China.

Excitation into this band irradiates also the Gd^{3+} lines around 250 nm ($^8S \rightarrow ^6D$). The resulting emission shows mainly Sb^{3+} emission with a small amount of Gd^{3+} and Tb^{3+} emission lines. The transfer from Sb^{3+} to Tb^{3+} will be inefficient, because of the low concentrations (both 1 at. %). Therefore, the transfer characteristics of $GdB_3O_6-Sb^{3+}, Tb^{3+}$ are as follows: (a) excitation into Sb^{3+} yields mainly Sb^{3+} emission; (b) excitation into Gd^{3+} yields mainly Tb^{3+} emission due to energy migration in the Gd^{3+} sublattice.

3.2. $GdB_3O_6-Bi^{3+}, Ln^{3+}$ ($Ln = Eu, Tb, Dy$)

Extremely efficient luminescent materials can be obtained from GdB_3O_6 by using Bi^{3+} as a sensitizer and several of the lanthanide ions as an activator. It has been shown that the Bi^{3+} ion transfers all of its excitation energy to the Gd^{3+} sublattice⁹. The lanthanide ions are able to trap the energy migrating in the Gd^{3+} sublattice. As an example we present in fig.1 the excitation spectrum of the Eu^{3+} emission of $GdB_3O_6-Bi^{3+}, Eu^{3+}$. Excitation into Bi^{3+} or Gd^{3+} results mainly in Eu^{3+} emission with only a few percent of Gd^{3+} emission. These results are obtained at 300 K as well as at 4.2 K. If we use LaB_3O_6 as the host lattice instead of GdB_3O_6 , excitation into the Bi^{3+} ion results mainly in Bi^{3+} emission. This shows that the Gd^{3+} ions play an intermediate role in the $Bi^{3+} \rightarrow Eu^{3+}$ energy transfer in GdB_3O_6 .

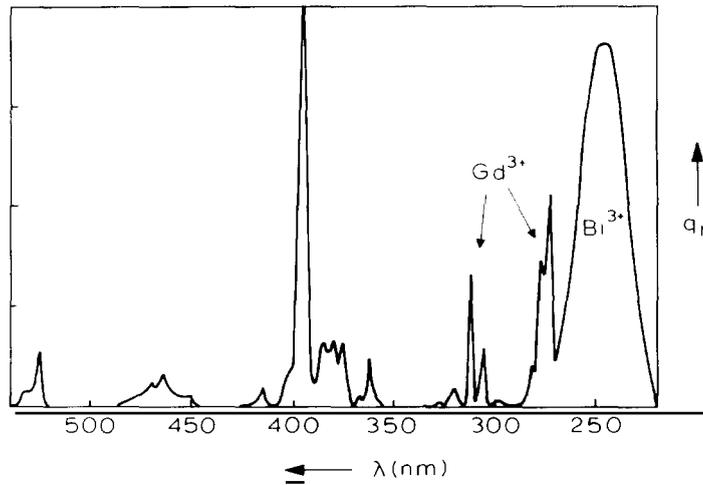


FIGURE 1

Excitation spectrum of the Eu^{3+} emission of GdB_3O_6-Bi, Eu at 300 K.

3.3. $\text{GdB}_3\text{O}_6\text{-Ce}^{3+}, \text{Ln}^{3+}$ (Ln = Eu, Tb)

There is no essential difference between the results for samples $\text{GdB}_3\text{O}_6\text{-Bi}^{3+}, \text{Ln}^{3+}$ and $\text{GdB}_3\text{O}_6\text{-Ce}^{3+}, \text{Ln}^{3+}$ with one exception of minor importance, viz. the Ce^{3+} ion does not transfer all of its excitation energy to the Gd^{3+} sublattice⁹. For the composition $\text{Gd}_{0.98}\text{Ce}_{0.01}\text{Tb}_{0.01}\text{B}_3\text{O}_6$, for example, the total light output under Ce^{3+} excitation consists for 15 % of Ce^{3+} emission and for 85 % of Gd^{3+} and Tb^{3+} emission.

Results for the mixed host lattice $(\text{La}, \text{Gd})\text{B}_3\text{O}_6$ confirm our results and will be presented elsewhere⁹.

REFERENCES

- 1) G. Blasse and A. Bril, *J. Luminescence* 3 (1970) 109.
- 2) G. Blasse, *J. Luminescence* 14 (1976) 231.
- 3) J.Th.W. de Hair, *J. Luminescence* 18/19 (1979) 797.
- 4) J.Th.W. de Hair and W.L. Konijnendijk, *J. Electrochem. Soc.* 127 (1980) 161.
- 5) J.Th.W. de Hair and J.T.C. van Kemenade, paper 54 at 3rd Int. Conf. Science and Technology of Light Sources, Toulouse, 1983.
- 6) G. Blasse, *Phys. Stat. Sol.(a)* 73 (1982) 205; 75 (1983) K 41.
- 7) M. Leskelä, M. Saakes and G. Blasse, *Mater. Res. Bull.* 19 (1984) 151.
- 8) G.K. Abdullaev, Kh.S. Mamedov and G.G. Dzhafarov. *Sov. Phys. Crystallogr.* 20 (1975) 161; 26 (1981) 473.
- 9) Hao Zhiran and G. Blasse, *Materials Chem. and Phys.*, to be published.