## Synthesis and reactivity of a ferrocene-derived PCP-pincer ligand

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The 1,3-bis(diphosphinomethyl)ferrocene 3 readily reacts with  $[(C_2H_{4})_2RhCl]_2$  to form an equilibrating pair of diastereomers 8a and 8b by C–H insertion into the ferrocene.

Pincer-type ligands have played a significant role in homogeneous catalysis.<sup>1</sup> With the recent development of effective catalysts for dehydrogenation that involve Rh, or more especially Ir, complexes related to compound **1**, interest in their application has been heightened.<sup>2</sup> Among the factors that contribute to their reactivity in the critical C–H activation step will be the chelate bite angle and the electronic character of both the  $\sigma$ - and n-bonding entities. Synthesis and complexation chemistry of a higher homologue **2** based on cycloheptatriene has already been reported.<sup>3</sup> For these reasons we are interested in the synthesis of the ferrocene analogue **3**, and report its successful completion together with some preliminary reactions that demonstrate the access to pincer-type ligation.



General methods for the preparation of 1,3-disubstituted ferrocenes are severely lacking.<sup>4</sup> In order to synthesise a suitable precursor, we prepared compound 4 according to published procedures, involving the photochemical displacement of *p*-xylene from  $CpFe[p-xylene]^+BF_4^-$  in the presence of 3-methoxycarbonyl-6-dimethylaminofulvene.5,6 This was cleanly reduced by LiAlH4 in refluxing THF to the primary diol 5a in 89% yield.<sup>7</sup> Attempts to form the diacetate 5b by conventional means (Ac<sub>2</sub>O, py, DMAP) were unsuccessful, but it was demonstrated that direct  $S_N1$  reaction in AcOH at 80 °C proceeded cleanly, according to precedent.8 Rather than isolate the diacetate according to conventional phosphination procedures,9 diol 5a was converted directly into the desired diphosphine 3 by heating with HPBut<sub>2</sub> (2 equiv.) in AcOH (80 °C, 1 h), isolating the product<sup>†</sup> in 62% yield by precipitation from MeOH.

The new ligand was surveyed for its reactivity towards simple unsaturated metal complexes under mild conditions. A model reaction was first established, using ligand **1b**. By NMR (C<sub>6</sub>D<sub>6</sub>, 65 °C, 30 min), displacement of the pincer diamine ligand<sup>10</sup> in **6** by the diphosphine was observed. The formation of complex **7** was inferred,‡ although it was not isolated. Repeating this reaction with ligand **3** gave no evidence of complexation, however. Reaction of the ferrocene-derived ligand **3** with  $(PPh_3)_3RuCl_2$  or  $(MeCN)_2PdCl_2$  likewise failed to give characterisable products.



When ligand **3** was reacted with  $[(C_2H_4)_2RhCl]_2$  in  $CD_2Cl_2$  at ambient temperature a clean transformation occurred within minutes, the product being characterised by NMR and MS.<sup>†</sup> Initially, two sets of resonances were observed in the <sup>1</sup>H NMR spectrum, in particular two distinct Rh–H signals at *ca.* –27 ppm. Equilibration occurred over 4 h so that the lower field signal, initially the minor component, became strongly predominant. The equilibration could also be followed by <sup>31</sup>P NMR, it being observed that the signal at 87.5 ppm, decreases monotonically over 3 h as the signal at 93.5 ppm increases. The structures of several (PCP)RhHCl complexes or Ir analogues have been determined, and exhibit common geometric features.<sup>11</sup> The ligand and M–Cl form a distorted square plane with the M–H in orthogonal relationship (Fig. 1).

This structural arrangement gives rise to a single magnetic environment for the hydride as defined by the high-field signal in the <sup>1</sup>H NMR spectrum, with one reported exception.§ In the present case there are two distinct structural isomers possible for the insertion products **8a** and **8b** in which the ( $C_5H_5$ )Fe



moiety is respectively *cis*- (8a) or *trans*- (8b) to the hydride. The hydride resonance of the more stable isomer formed on standing the solution in  $CD_2Cl_2$  for several hours shows a lack of NOE communication with the  $C_5H_5$  ring, but an intense NOE to one



Fig. 1 Geometry of MHCl pincer complexes.

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pair of *tert*-butyl groups. On this basis the assignment of *cis*- or *trans*- to the preferred initial and final products of insertion is as defined in the spectroscopic data.<sup>†</sup> The results are consistent with initial intramolecular C–H insertion from the less hindered *exo*-face of the ferrocene to give complex **8a**. Equilibration of complexes **8a** and **8b** may occur through reversal of that step, or phosphine dissociation and recombination.

The reactivity of ligand 3 towards C–H insertion in a simple Rh complex demonstrates that the relevant bis-chelate can be readily formed, despite the additional strain energy engendered by the smaller template ring of ferrocene *vs.* arene. The catalytic chemistry of 3 and its relatives will be explored in future work.

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## Notes and references

† For **3**: Mp 157–158 °C (decomp.).  $\delta_{\rm H}$  (CDCl<sub>3</sub>): 4.35 (brs, 1H), 4.15 (d,  $J_{\rm HP}$  1 Hz, 2H), 4.06 (s, 5H), 2.59 (d,  $J_{\rm HP}$  2 Hz, 4H, CH<sub>2</sub>), 1.11 (2 × d,  $J_{\rm HP}$  11 Hz, 18H, 18H, diastereotopic Bu<sup>1</sup>).  $\delta_{\rm C}$  (CDCl<sub>3</sub>): 87.6 (d,  $J_{\rm CP}$  20 Hz, 2C), 70.3 (s, 1 CH), 70.1 (s, 5 CH), 68.9 (d,  $J_{\rm CP}$  6 Hz, 2 CH), 31.8 (2d,  $J_{\rm CP}$  22 Hz,  $J_{\rm CP}$  7 Hz, 4 C), 30.1 (2d,  $J_{\rm CP}$  13 Hz,  $J_{\rm CP}$  3 Hz, 12 Me), 22.7 (d,  $J_{\rm CP}$  23 Hz, 2 CH<sub>2</sub>).  $\delta_{\rm P}$  (CDCl<sub>3</sub>): 34.1 (s). MS AP+ 502.2601 (Calc. for C<sub>28</sub>H<sub>48</sub>P<sub>2</sub>Fe 502.2581. For **8a**:  $\delta_{\rm H}$  (CD<sub>2</sub>Cl<sub>2</sub>): –27.55 (dt, 1H Rh–H,  $J_{\rm RhH}$  47 Hz,  $^{2}J_{\rm PH}$  12.5 Hz), 1.23 (9H, virtual q, *J* 6.5 Hz), 1.52 (9H, virtual q, *J* = 6.5 Hz), 2.67 (1H, brq), 3.00 (1H, brq), 3.96 (s, 5H), 4.34 (s, 2H).  $\delta_{\rm P}$  (CD<sub>2</sub>Cl<sub>2</sub>): 88.45 (d,  $J_{\rm RhP}$  114Hz). For **8b**:  $\delta_{\rm H}$  (CD<sub>2</sub>Cl<sub>2</sub>): –27.45 (dt, 1H Rh–H,  $J_{\rm RhH}$  47 Hz,  $^{2}J_{\rm PH}$  13 Hz), 1.20 (9H, virtual q, *J* 6.5 Hz), 1.57 (9H, virtual q, *J* 6.5 Hz), 2.54 (1H, brq), 2.90 (1H, brq), 3.91 (5H, s), 4.42 (2H, s).  $\delta_{\rm P}$  (CD<sub>2</sub>Cl<sub>2</sub>): 93.5 (d,  $J_{\rm RhP}$  115 Hz). ES MS (**8a** + **8b**): 605.1634 (M<sup>+</sup> - Cl) (Calc. for C<sub>28</sub>H<sub>47</sub>FeP<sub>2</sub>Rh 605.1636); 604.1575 (M<sup>+</sup> - HCl) (Calc. for C<sub>28</sub>H<sub>47</sub>FeP<sub>2</sub>Rh 604.1557).

 $\ddagger$  For 7:  $\delta_{P}$  (C<sub>6</sub>D<sub>6</sub>): 46, 50 (ABq, J 5 Hz), 61 (s).

§ In ref. 11*b* the HRhCl pincer complex derived from ligand **1a** has been reported as giving rise to two distinct high-field <sup>1</sup>H NMR signals considered to be conformational isomers. This observation was confirmed by us when the synthesis was repeated starting with the ligand-(HBr)<sub>2</sub> salt. We suspected that a mixture of HRhCl and HRhBr complexes was responsible,

however. Further work has confirmed that the pure HRhCl complex, unambiguously synthesised, gives only a single high-field resonance  $\delta_{\rm H}$  –27.8 ppm,  $J_{\rm RhH}$  52Hz,  $^2J_{\rm PH}$  12 Hz.

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